

## The structure of carboxypeptidase A†.

### VIII. Atomic interpretation at 0.2 nm resolution, a new study of the complex of glycyl-L-tyrosine with CPA, and mechanistic deductions

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#### I. INTRODUCTION

Bovine pancreatic carboxypeptidase A<sub>α</sub> (CPA),‡ the subject of these studies, is a zinc containing enzyme of molecular weight 34 600, which catalyses the hydrolysis of polypeptides and esters at the C-terminal peptide or ester bond. Experiments to date have shown that in order to be hydrolysed, the substrate must contain a C-terminal residue in the L configuration, with the carboxyl group free and α to the peptide or ester bond which is to be cleaved. In addition, the reaction is favoured if the C-terminal residue is aromatic.

Our crystallographic studies of CPA have yielded electron density maps of the native enzyme at 0.6 nm (Lipscomb, Coppola, Hartsuck, Ludwig, Muirhead, Searl & Steitz 1966), 0.28 nm (Ludwig, Hartsuck, Steitz, Muirhead, Coppola, Reeke & Lipscomb 1967) and 0.20 nm resolution (Reeke, Hartsuck, Ludwig, Quicho, Steitz & Lipscomb 1967; Lipscomb, Hartsuck, Reeke, Quicho, Bethge, Ludwig, Steitz, Muirhead & Coppola 1968). Concurrently, a study of the binding of a number of substrates and inhibitors at 0.6 nm resolution was under way (Steitz, Ludwig, Quicho & Lipscomb 1967). Subsequently, the most promising of these complexes, that of glycyl-L-tyrosine with CPA, was carried to atomic resolution (Reeke *et al.* 1967; Lipscomb *et al.* 1968). Even though chemical sequence is not available for several of the binding and catalytic groups of the enzyme, we have been able to deduce the identity of the binding residue Arg-145 and the catalytic residue Glu-270 (Reeke *et al.* 1967), to describe the mode of binding of Gly-Tyr, to extrapolate these conclusions to the binding of polypeptides, and to propose a mechanism, with certain ambiguities, for the action of the enzyme (Lipscomb *et al.* 1968).

We have now completed the detailed atomic interpretation of the 0.20 nm electron density map, supplying residue identifications where the primary sequence is not available, and have subjected the resulting atomic coordinates to a model building procedure (Diamond 1966) which forces them to conform to standard bond distances and angles. The improved coordinates have been entered in a structure factor calculation, which gave a standard crystallographic R factor of 0.44.

We have also analysed some characteristics of the various structural features in the molecule—helices, pleated sheet, and random coil—and we present the distribution of residues among them. The structure of the complex between Gly-Tyr and CPA has been further clarified at

† To M. L. Anson, 1901–1968.

‡ Abbreviations used in this paper are: CPA, Carboxypeptidase A in the α form (Sampath-Kumar, Clegg & Walsh 1964); MIR, multiple isomorphous replacement; SF, structure factor;  $F_o$ , observed structure factor;  $F_c$ , calculated structure factor; HPLA, hippuryl-L-β-phenyllactate; ES, enzyme-substrate complex; CBZ, carbobenzoxy.

0.20 nm resolution by use of the phases from the structure factor calculation and a new function for computing the difference electron density map. The conclusions which we have previously made concerning the mechanism of action of CPA are essentially unchanged but have become less ambiguous in certain aspects.

## 2. STRUCTURE OF CARBOXYPEPTIDASE A AT 0.20 nm RESOLUTION

### (a) Determination of structure

Carboxypeptidase A<sub>α</sub> crystallizes in the monoclinic space group P2<sub>1</sub>, with unit cell dimensions  $a = 5.141$  nm,  $b = 5.989$  nm,  $c = 4.719$  nm,  $\beta = 97.58^\circ$ . The preparation of heavy metal derivatives (Lipscomb *et al.* 1966) and the measurement of X-ray diffraction intensities (Lipscomb *et al.* 1968) have been described previously. In summary, complete data (20 600 reflexions) to 0.20 nm resolution were measured on the native enzyme, complete data to 0.28 nm resolution were measured on four heavy atom derivatives—Pb (2 sites), Hg (3 sites), Hg (1 site), and Pt (4 sites)—and those 6000 of the 14 000 reflexions between 0.30 and 0.20 nm which gave the largest intensities in the native data set were also measured for the Pb(2) and Hg(3) derivatives.

TABLE I. HEAVY ATOM BINDING TO CARBOXYPEPTIDASE†

atom	Z, e/mol	$B_{\ddagger}/\text{nm}^2$	heavy atom coordinates§			residue
			$x$	$y$	$z$	
Pb <sub>1</sub>	58	0.09	-0.094	0.500	-0.089	Glu-270
Pb <sub>2</sub>	53	0.26	-0.087	0.540	-0.147	citrate, not protein
Hg, g	50	0.16	-0.071	0.455	-0.115	His-69, Glu-72, Lys-196
Hg, s <sub>1</sub>	47	0.13	-0.071	0.452	-0.115	His-69, Glu-72, Lys-196
Hg, s <sub>2</sub>	46	0.31	-0.506	0.069	-0.257	His-29
Hg, s <sub>3</sub>	48	0.25	-0.475	0.109	-0.136	His-29, Lys-84
Pt <sub>1</sub>	74	0.71	0.341	0.430	0.034	Cys-161
Pt <sub>2</sub>	45	0.71	-0.438	0.305	-0.568	Met-103
Pt <sub>3</sub>	68	1.03	-0.292	0.082	0.141	N-terminus: Ala-1
Pt <sub>4</sub>	27	0.69	-0.484	0.485	-0.500	His-303
Ag <sub>1</sub>	—	—	0.238	0.498	-0.278	His-166, Ser-158
Ag <sub>2</sub>	—	—	-0.082	0.220	0.193	His-120
Ag <sub>3</sub>	—	—	-0.457	0.090	-0.143	His-29, (Lys-84)
Ag <sub>4</sub>	—	—	-0.483	0.477	-0.516	His-303
Co <sub>1</sub>	—	—	-0.500	0.500	-0.500	His-303
Co <sub>2</sub>	—	—	-0.500	0.070	-0.130	His-29, (Lys-84)
Zn	—	—	-0.087	0.443	-0.155	His-69, Glu-72, Lys-196

† All of these derivatives except Hg, g were prepared by dialysing CPA crystals against solutions of the indicated composition (mol l<sup>-1</sup>).

Pb: 0.003 PbCl<sub>2</sub>, 0.01 Na citrate, 0.2 LiCl, 0.02 tris pH 7.5.

Hg, s: data in this paper were taken on crystals soaked against 0.0008 HgCl<sub>2</sub>, 0.2 LiCl, 0.02 tris, pH 7.5, but the fourth site is not occupied except under prolonged soaking against 0.003 M *p*-acetoxymercurilaniline, 0.2 Na acetate and 0.02 tris pH 8. This fourth site is at  $x = -0.448$ ,  $y = 0.507$ ,  $z = 0.575$ , and is associated with His-303.

Pt: 0.003 K<sub>2</sub>PtCl<sub>4</sub>, 0.2 LiCl, 0.02 tris, pH 7.5.

Ag: 0.005 AgNO<sub>3</sub>, 0.2 Na acetate, pH 8.

Co: 0.01 CoCl<sub>2</sub>, 0.01 tris, pH 7.5.

Hg, g:  $5 \times 10^{-4}$  CPA was dialysed against 0.001 HgCl<sub>2</sub>, 1 LiCl, 0.02 tris pH 8 and crystallized by dialysis against 0.18 LiCl, 0.02 tris, pH 8.

‡ The effective isotropic temperature factor B for the protein is 0.16 nm<sup>2</sup>.

§ The transformation to the symmetry related position is  $x' = -x$ ,  $y' = \frac{1}{2} + y$ ,  $z' = -z$ .

|| Zn coordinates were found by interpolation of the electron density map.

Heavy atom positional, thermal, and occupancy parameters, obtained and refined previously (Lipscomb *et al.* 1968), are given in table 1, which also shows the binding site on the protein of each heavy atom. The 0.20 nm electron density map was computed from multiple isomorphous replacement (MIR) phases and contours were plotted on Mylar sheets. The sheets were supported vertically in a frame designed so that any desired portion of the map could be viewed by a seated observer. The detailed atomic interpretation was accomplished by viewing simultaneously the map and appropriate Kendrew models, which were on the same scale, and by placing coloured markers on the map at proposed atomic positions chosen to place the structure as much as possible in the density. Since the complete sequence was not known, at first only the backbone was constructed. Then an analysis (Lipscomb *et al.* 1968) was made of the known portion of sequence at the N-terminus of the molecule and atomic positions were derived for all side chains in this fragment. Simultaneously, tentative identifications of all residues not in known terminal sequences were made from the map. These tentative identifications enabled us to locate in the complete sequence all other chemically sequenced fragments known to us,† and to assign numbers to the residues in these fragments. The known fragments were then used to revise our initial identifications. A complete (but still tentative) sequence was then drawn up for use in refinements, structure factor calculations, and model building. This sequence, which is listed in table 2, contains 93 residues for which only X-ray identifications have been made. These identifications will be compared to the chemical results when they become available.

TABLE 2. SEQUENCE USED FOR THE STRUCTURE FACTOR CALCULATION‡

A R S T B T F B Y A	T Y H T L B Z I Y B	F M B L L V G Z H P
Z L L S K L Z I G R	T Y Z G R P I Y V L	K F S T G G S B R P
A L W I B L G I H S	R Z W I T Z A T G V	W F A K K F T Z B Y
G B Z P S F T A I L	B S M K L F L Z I V	T B P B G F A F T H
S Z Z R L W Z Z T R	S T G S G G Z C V G	V B A B R T W B A G
W G K K G A S S S P	C S Z T Y Y G K Y A	Z S Z T Z V K S I V
B F V K B H G B F K	A F S S L K G Y S Z	B S L Y P Y G Y T T
Z S L P B K T Z L B	Z V A K S A V A A L	K S L Y G T S Y K R
G S I I S S I Y Z A	S G G I S B H S Y B	Z G I K R S F V F Z
L R B V G S W G F B	H P A K Z I L P V S	H Z L W B G V B T I
M Z H T V B B		

A, Ala; B, Asx; C, Cys; F, Phe; G, Gly; H, His; I, Ile; K, Lys; L, Leu; M, Met; P, Pro; R, Arg; S, Ser; T, Thr; V, Val; W, Trp; Y, Tyr; Z, Glx.

‡ This table lists the X-ray identifications in the case of those residues for which the chemical and X-ray sequences have not yet been reconciled or for which no chemical identification is yet available. Except for those residues whose identification has been published by Professor Neurath's laboratory (see text) and for which the X-ray and chemical identifications are in agreement, this sequence should be treated as provisional. The present authors accept total responsibility for any errors in this sequence of CPA.

The mathematical model building program of R. Diamond (1966) was used to improve the atomic positions. This is a 'guided' model building program; that is, the atomic coordinates from the map are used to guide the positioning of the atoms of a model of idealized geometry. The criterion used is a least squares minimization of the distances between the guide atoms and the corresponding atoms of the model. The idealized coordinates for the peptide group and the

† These fragments are residues 1–22 (Sampath Kumar *et al.* 1964), 23–103 (Neurath *et al.* 1968), 138–144 and 152–165 (Sampath Kumar, Walsh, Bargetzi & Neurath 1963; Neurath 1964), 247–250 (Roholt & Pressman 1967) and 301–307 (Bargetzi *et al.* 1964). Other tentatively sequenced fragments (Neurath 1968 private communication) have been assigned provisionally by us as 104–125, 178–184, 198–201, 202–239, and 258–264. Only the X-ray evidence is available for the total number of residues, the identification of the 93 residues not in the above fragments, and the numbering beyond residue 103, including the numbering of the heptapeptide which is known chemically to be C-terminal.

various side chains were taken from X-ray structure determinations of appropriate model molecules. Initially, only rotations about single bonds in the model were allowed to vary, and all guide atoms differing in position by more than 0.05 nm from the corresponding refined model atoms were examined for errors and readjusted if necessary. During this examination we observed that some regions of the molecule must be distorted from the idealized geometry of the model we were attempting to fit. Inspection of the density contours suggested that in many cases these distortions involved the backbone, principally the backbone angle  $\tau$  ( $C_{\text{carbonyl}}-C_{\alpha}-N$ ). Accordingly, after all guide atoms were in satisfactory positions the model building program was repeated, this time allowing the bond angles around the alpha carbons to vary.† All distortions were accommodated as variations in these angles because it was not convenient to vary the shape of the peptide group itself. After this procedure, unrealistic bond angles (differing by more than  $\pm 10^\circ$  from tetrahedral) were found at 8% of the alpha carbons in the molecule. Where these occurred at large residues, the program probably placed too much weight on the guide positions of the side chains, and bond angle distortions were probably also present at the  $\beta$  carbons. The average value of  $\tau$  for the entire molecule is  $112 \pm 5^\circ$  rather than the expected  $109.2^\circ$ . This value probably reflects the fact that all peptide distortions were taken up in this one parameter, but might also indicate that packing of side chains in such a large globular molecule may require that, on the average, the peptides be spread apart slightly at the  $\alpha$  carbons. The r.m.s. deviation of guide atom positions from final coordinates is 0.023 nm. The final atomic coordinates resulting from the Diamond program are given in table 3.

A structure factor calculation was performed with the atomic positions given by the Diamond program. All atoms were given the Wilson plot isotropic thermal parameter of  $0.12 \text{ nm}^2$  except the atoms of residues 133–137, a loop on the outside of the molecule, which were assigned thermal parameters of  $0.23 \text{ nm}^2$ . A total of 128 atoms, or 5.3% of the molecule, was omitted either because these atoms had uncertain positions or because they belonged to important side chains, whose identities we wish to confirm by calculating a new electron density map from the structure factor phases and observed intensities. Solvent structure was also omitted from the structure factor calculation. Structure factors were calculated for 16 642 reflexions, all those measured to 0.20 nm resolution except those which our data correlation program had indicated were measured very poorly. The results of this structure factor calculation constitute table 4.‡ The overall standard crystallographic  $R$  factor is 0.44. The  $R$  factor and the difference between the MIR and calculated (SF) phases were examined as functions of structure factor and resolution (table 5). The value of  $R$  is 0.36 for the range 0.2 to 0.5 nm resolution, and surprisingly does not increase at the higher resolutions. The reflexions with large interplanar spacings give poor agreement because they are most affected by the solvent structure, which has not been

† In the notation of Diamond (1966), five probes were used with the following parameters:

probe	residues	delay 2	min. ratio ( $C_1$ )	min. value ( $C_2$ )
1	0	30	$1.00 \times 10^{-4}$	1.0
2	1	30	$1.00 \times 10^{-4}$	1.0
3	3	30	$1.00 \times 10^{-4}$	1.0
4	3	10	$0.20 \times 10^{-1}$	1.0
5	6	10	$0.60 \times 10^{-3}$	1.0

These values were adopted after some experimentation and may be helpful to others using this program.

‡ Table 4 is deposited in the Archives of the Royal Society where it may be consulted, and also at the National Lending Library (see note on p. 214) where copies may be ordered (reference SUP 10002).

TABLE 3. THE COORDINATES OF ATOMS OF CARBOXYPEPTIDASE A USED IN THE STRUCTURE FACTOR CALCULATION

1	ALA	CB	-0.293	0.060	0.109	15	CB	-0.076	0.160	0.027	29	CA	-0.442	0.044	-0.241		
1		CA	-0.322	0.063	0.100	15	CA	-0.105	0.161	0.033	29	CO	-0.424	0.034	-0.262		
1		CO	-0.331	0.042	0.082	15	CO	-0.110	0.138	0.027	29	O	-0.412	0.047	-0.277		
1		O	-0.320	0.037	0.061	15	O	-0.132	0.134	0.005	30	N	-0.424	0.012	-0.263		
2		N	-0.351	0.032	0.091	16	N	-0.112	0.123	0.049	30	PRO	CG	-0.436	-0.026	-0.257	
2	ARG	NH1	-0.349	-0.052	0.185	16	ASP	OD1	-0.075	0.067	0.080	30	CD	-0.440	-0.002	-0.247	
2		NH2	-0.371	-0.084	0.169	16	OD2	-0.071	0.101	0.087	30	CB	-0.413	-0.024	-0.273		
2		CZ	-0.366	-0.063	0.165	16	CG	-0.085	0.085	0.082	30	CA	-0.409	-0.001	-0.281		
2		NE	-0.378	-0.052	0.142	16	CB	-0.114	0.090	0.077	30	CO	-0.414	0.003	-0.314		
2		CD	-0.377	-0.028	0.138	16	CA	-0.122	0.100	0.048	30	O	-0.400	-0.006	-0.330		
2		CG	-0.365	-0.022	0.112	16	CO	-0.152	0.104	0.043	31	N	-0.435	0.015	-0.322		
2		CB	-0.379	-0.001	0.097	16	O	-0.166	0.096	0.022	31	GLU	OE1	-0.483	0.004	-0.309	
2		CA	-0.362	0.011	0.077	17	N	-0.161	0.115	0.064	31	OE2	-0.495	-0.031	-0.310		
2		CO	-0.380	0.017	0.049	17	GLU	OE1	-0.197	0.138	31	CD	-0.487	-0.014	-0.319		
2		O	-0.391	0.002	0.034	17	OE2	-0.198	0.134	0.157	31	CG	-0.477	-0.014	-0.350		
3		N	-0.381	0.038	0.044	17	CD	-0.193	0.147	0.140	31	CB	-0.472	0.010	-0.361		
3	SER	OG	-0.443	0.044	0.001	17	CG	-0.177	0.140	0.114	31	CA	-0.444	0.020	-0.352		
3		CB	-0.424	0.042	0.025	17	CB	-0.194	0.140	0.085	31	CO	-0.445	0.045	-0.360		
3		CA	-0.397	0.047	0.018	17	CA	-0.188	0.120	0.064	31	O	-0.459	0.052	-0.382		
3		CO	-0.396	0.072	0.010	17	CO	-0.204	0.122	0.035	32	N	-0.429	0.058	-0.342		
3		O	-0.387	0.086	0.028	17	O	-0.223	0.110	0.026	32	LEU	CD1	-0.488	0.101	-0.352	
4		N	-0.406	0.077	-0.016	18	N	-0.196	0.139	0.019	32	CD2	-0.449	0.126	-0.349		
4	THR	OG1	-0.433	0.089	-0.071	18	ILE	CD1	-0.176	0.190	-0.054	32	CG	-0.463	0.109	-0.333	
4		CG2	-0.387	0.095	-0.073	18	CG1	-0.191	0.169	-0.048	32	CB	-0.445	0.090	-0.324		
4		CB	-0.411	0.103	-0.061	18	CB	-0.198	0.167	-0.018	32	CA	-0.427	0.082	-0.346		
4		CA	-0.408	0.099	-0.028	18	CG2	-0.175	0.170	0.007	32	CO	-0.400	0.088	-0.332		
4		CO	-0.428	0.114	-0.017	18	CA	-0.208	0.144	-0.010	32	O	-0.387	0.104	-0.340		
4		O	-0.425	0.135	-0.014	18	CO	-0.202	0.129	-0.035	33	N	-0.390	0.075	-0.310		
5		N	-0.450	0.104	-0.013	18	O	-0.220	0.125	-0.056	33	LEU	CD1	-0.390	0.116	-0.264	
5	ASN	NOD1	-0.514	0.089	-0.059	19	N	-0.178	0.121	-0.033	33	CD2	-0.342	0.116	-0.256		
5		NOD2	-0.515	0.125	-0.049	19	TYR	OH	-0.077	0.158	-0.107	33	CG	-0.366	0.104	-0.250	
5		CG	-0.510	0.106	-0.040	19	CD2	-0.097	0.110	-0.067	33	CB	-0.367	0.080	-0.262		
5		CB	-0.497	0.101	-0.011	19	CE2	-0.081	0.124	-0.082	33	CA	-0.364	0.077	-0.294		
5		CA	-0.472	0.115	-0.002	19	CZ	-0.092	0.145	-0.093	33	CO	-0.342	0.061	-0.298		
5		CO	-0.464	0.117	0.030	19	CE1	-0.116	0.151	-0.089	33	O	-0.345	0.040	-0.295		
5		O	-0.476	0.130	0.045	19	CD1	-0.133	0.136	-0.074	34	N	-0.320	0.070	-0.304		
6		N	-0.444	0.104	0.040	19	CB	-0.122	0.116	-0.063	34	SER	OG	-0.294	0.083	-0.347	
6	THR	OG1	-0.453	0.071	0.088	19	CG	-0.139	0.101	-0.048	34	CB	-0.294	0.061	-0.339		
6		CG2	-0.406	0.074	0.100	19	CA	-0.169	0.107	-0.054	34	CA	-0.297	0.058	-0.308		
6		CB	-0.429	0.078	0.078	19	CO	-0.190	0.089	-0.052	34	CO	-0.274	0.070	-0.291		
6		CA	-0.433	0.103	0.070	19	O	-0.203	0.081	-0.074	34	O	-0.273	0.091	-0.290		
6		CO	-0.408	0.117	0.076	20	N	-0.193	0.084	-0.025	35	N	-0.255	0.056	-0.278		
6		O	-0.394	0.115	0.100	20	ASP	OD1	-0.191	0.031	0.015	35	LYS	NZ	-0.176	0.028	-0.152
7		N	-0.464	0.130	0.055	20	OD2	-0.187	0.051	0.052	35	CE	-0.178	0.050	-0.167		
7	PHE	CD2	-0.323	0.136	0.034	20	CG	-0.196	0.048	0.028	35	CD	-0.186	0.047	-0.199		
7		CE2	-0.296	0.142	0.038	20	CB	-0.212	0.067	0.015	35	CG	-0.204	0.065	-0.219		
7		CZ	-0.288	0.164	0.038	20	CA	-0.212	0.068	-0.018	35	CB	-0.228	0.056	-0.230		
7		CE1	-0.307	0.180	0.036	20	CO	-0.241	0.072	-0.032	35	CA	-0.231	0.064	-0.261		
7		CD1	-0.334	0.174	0.033	20	O	-0.255	0.057	-0.045	35	CO	-0.209	0.057	-0.277		
7		CG	-0.342	0.152	0.031	21	N	-0.248	0.093	-0.031	35	O	-0.205	0.037	-0.282		
7		CB	-0.370	0.147	0.028	21	PHE	CD2	-0.301	0.150	-0.078	36	N	-0.194	0.074	-0.284	
7		CA	-0.381	0.145	0.056	21	CE2	-0.326	0.160	-0.091	36	LEU	CD1	-0.154	0.047	-0.356	
7		CO	-0.408	0.168	0.062	21	CZ	-0.349	0.153	-0.084	36	CD2	-0.195	0.063	-0.380		
7		N	-0.468	0.178	0.062	21	CE1	-0.349	0.139	-0.062	36	CG	-0.180	0.056	-0.352		
8		ASN	NOD1	-0.366	0.175	0.087	21	CD1	-0.325	0.130	-0.048	36	CB	-0.178	0.076	-0.331	
8		NOD2	-0.373	0.234	0.131	21	CG	-0.302	0.135	-0.056	36	CA	-0.171	0.071	-0.299		
8		CG	-0.349	0.223	0.172	21	CB	-0.277	0.126	-0.040	36	CO	-0.146	0.082	-0.286		
8		CB	-0.359	0.218	0.147	21	CA	-0.274	0.101	-0.043	36	O	-0.147	0.100	-0.273		
8		CA	-0.355	0.196	0.133	21	CO	-0.280	0.093	-0.074	37	N	-0.124	0.071	-0.289		
8		CO	-0.367	0.197	0.101	21	O	-0.297	0.079	-0.082	37	GLN	NDE1	-0.035	0.041	-0.207	
8		O	-0.352	0.213	0.084	22	N	-0.264	0.103	-0.091	37	NDE2	-0.065	0.022	-0.237		
8		N	-0.327	0.214	0.087	22	MET	CE	-0.191	0.155	-0.132	37	CD	-0.054	0.040	-0.228	
9		TYR	OH	-0.405	0.163	-0.033	22	SD	-0.196	0.126	-0.138	37	CG	-0.065	0.061	-0.242	
9		CD2	-0.403	0.211	0.015	22	CG	-0.229	0.124	-0.131	37	CB	-0.081	0.059	-0.269		
9		CE2	-0.411	0.190	0.003	22	CB	-0.238	0.100	-0.130	37	CA	-0.098	0.079	-0.278		
9		CZ	-0.398	0.183	-0.020	22	CA	-0.266	0.098	-0.122	37	CO	-0.087	0.092	-0.301		
9		CE1	-0.378	0.194	-0.029	22	CO	-0.274	0.074	-0.126	37	O	-0.082	0.083	-0.323		
9		CD1	-0.370	0.216	-0.015	23	N	-0.292	0.068	-0.145	38	N	-0.083	0.114	-0.295		
9		CG	-0.383	0.223	0.007	23	ASP	OD1	-0.260	0.059	-0.109	38	ILE	CD1	-0.079	0.186	-0.275
9		CB	-0.375	0.246	0.021	23	OD2	-0.227	0.015	-0.096	38	CG1	-0.081	0.161	-0.280		
9		CA	-0.357	0.244	0.049	23	CG	-0.224	0.012	-0.107	38	CB	-0.081	0.154	-0.311		
9		CO	-0.355	0.265	0.066	23	CB	-0.244	0.022	-0.090	38	CG2	-0.107	0.154	-0.331		
9		O	-0.345	0.282	0.057	23	CA	-0.265	0.035	-0.109	38	CA	-0.072	0.129	-0.314		
10		N	-0.363	0.264	0.091	23	CO	-0.292	0.028	-0.106	38	CO	-0.042	0.130	-0.308		
10	ALA	CB	-0.386	0.281	0.128	23	CB	-0.302	0.010	-0.119	38	O	-0.028	0.138	-0.325		
10		CA	-0.363	0.283	0.111	24	N	-0.305	0.041	-0.090	39	N	-0.033	0.120	-0.283		
10		CO	-0.338	0.281	0.133	24	LEU	CD1	-0.333	0.020	-0.019	39	GLY	CA	-0.004	0.119	-0.273
10		O	-0.340	0.285	0.159	24	CD2	-0.320	0.059	-0.007	39	CO	0.003	0.112	-0.241		
11		N	-0.317	0.276	0.123	24	CG	-0.321	0.041	-0.029	39	O	-0.014	0.104	-0.228		
11	THR	OG1	-0.269	0.261	0.183	24	CB	-0.335	0.049	-0.057	40	N	0.027	0.116	-0.231		
11		CG2	-0.289	0.233	0.150	24	CA	-0.332	0.036	-0.084	40	ARG	NH1	0.108	0.068	-0.174	
11		CB	-0.291	0.256	0.163	24	CO	-0.351	0.047	-0.108	40	NH2	0.093	0.045	-0.141		
11		CA	-0.291	0.274	0.140	24	O	-0.368	0.036	-0.122	40	CZ	0.092	0.053	-0.167		
11		CO	-0.269	0.270	0.122	25	N	-0.346	0.068	-0.113	40	NE	0.073	0.044	-0.187		
11		O	-0.274	0.263	0.096	25	LEU	CD1	-0.358	0.140	-0.171	40	CD	0.067	0.053	-0.216	
12		N	-0.245	0.274	0.135	25	CD2	-0.397	0.116	-0.170	40	CG	0.046	0.071	-0.217		
12	TYR	OH	-0.233	0.377	0.104	25	CB	-0.367	0.116	-0.169	40	CB	0.056	0.091	-0.198		
12		CD2	-0.221	0.324	0.143	25	CG	-0.355	0.106	-0.141	40	CA	0.038	0.11			

TABLE 3 (cont.)

44	CO	0.061	0.088	-0.104	59	O	-0.318	0.192	-0.546	73	O	-0.196	0.326	0.011
44	O	0.049	0.070	-0.107	60	N	-0.350	0.211	-0.574	74	N	-0.196	0.363	0.017
45	N	0.050	0.107	-0.098	60	PRO CG	-0.382	0.235	-0.599	74	ILE CD1	-0.274	0.417	-0.004
45	ARG NH1	0.065	0.170	0.012	60	CD	-0.374	0.211	-0.595	74	CG1	-0.264	0.393	-0.001
45	NH2	0.066	0.145	0.050	60	CB	-0.359	0.248	-0.583	74	CB	-0.233	0.391	0.006
45	CZ	0.057	0.150	0.023	60	CA	-0.342	0.233	-0.564	74	CG2	-0.217	0.403	-0.015
45	NE	0.041	0.136	0.007	60	CO	-0.343	0.235	-0.532	74	CA	-0.223	0.367	0.005
45	CD	0.027	0.142	-0.021	60	C	-0.363	0.231	-0.521	74	O	-0.228	0.359	-0.026
45	CG	0.024	0.122	-0.041	61	N	-0.320	0.242	-0.517	74	O	-0.251	0.355	-0.038
45	CB	0.019	0.129	-0.072	61	ALA CB	-0.292	0.232	-0.473	75	N	-0.207	0.356	-0.038
45	CA	0.023	0.110	-0.093	61	CA	-0.317	0.245	-0.486	75	THR CG1	-0.162	0.344	-0.071
45	CO	0.006	0.113	-0.123	61	CO	-0.309	0.267	-0.472	75	CG2	-0.180	0.380	-0.083
45	O	0.009	0.130	-0.138	61	O	-0.298	0.282	-0.484	75	CB	-0.185	0.355	-0.084
46	N	-0.011	0.097	-0.130	62	N	-0.314	0.269	-0.445	75	CA	-0.207	0.348	-0.067
46	PRO CD	-0.041	0.067	-0.125	62	LEU CD1	-0.360	0.324	-0.447	75	CO	-0.214	0.323	-0.073
46	CC	-0.017	0.079	-0.110	62	CD2	-0.381	0.297	-0.417	75	O	-0.234	0.318	-0.089
46	CB	-0.044	0.076	-0.155	62	CG	-0.358	0.301	-0.434	76	N	-0.197	0.310	-0.058
46	CA	-0.029	0.097	-0.156	62	CB	-0.332	0.298	-0.414	76	GLU CE1	-0.147	0.308	-0.024
46	CO	-0.046	0.118	-0.162	62	CA	-0.309	0.289	-0.427	76	CE2	-0.109	0.296	-0.033
46	O	-0.058	0.126	-0.142	62	CO	-0.287	0.279	-0.404	76	CD	-0.132	0.296	-0.034
47	N	-0.046	0.127	-0.187	62	O	-0.290	0.261	-0.392	76	CG	-0.149	0.277	-0.052
47	ILE CD1	-0.006	0.149	-0.186	63	N	-0.266	0.292	-0.399	76	CB	-0.176	0.273	-0.042
47	CG1	-0.019	0.164	-0.210	63	TRP CD1	-0.180	0.267	-0.355	76	CA	-0.200	0.285	-0.060
47	CB	-0.048	0.160	-0.220	63	CE1	-0.185	0.245	-0.361	76	CO	-0.226	0.278	-0.049
47	CG2	-0.066	0.180	-0.224	63	CZ1	-0.167	0.230	-0.346	76	O	-0.238	0.261	-0.058
47	CA	-0.061	0.147	-0.197	63	CH	-0.147	0.236	-0.325	77	N	-0.233	0.292	-0.029
47	CO	-0.086	0.134	-0.209	63	CZ2	-0.141	0.257	-0.318	77	ALA CB	-0.259	0.309	0.004
47	O	-0.085	0.119	-0.227	63	CE2	-0.158	0.273	-0.333	77	CA	-0.257	0.289	-0.016
48	N	-0.108	0.141	-0.200	63	NE	-0.157	0.294	-0.331	77	CO	-0.281	0.292	-0.038
48	TYR CH	-0.111	0.035	-0.152	63	CD2	-0.179	0.303	-0.350	77	O	-0.298	0.276	-0.043
48	CD2	-0.129	0.094	-0.147	63	CG	-0.192	0.286	-0.365	78	N	-0.284	0.311	-0.051
48	CE2	-0.119	0.072	-0.139	63	CB	-0.217	0.288	-0.389	78	THR OG1	-0.299	0.355	-0.059
48	CZ	-0.120	0.056	-0.160	63	CA	-0.245	0.286	-0.378	78	CG2	-0.314	0.348	-0.109
48	CE1	-0.126	0.061	-0.188	63	CO	-0.245	0.301	-0.352	78	CB	-0.297	0.340	-0.083
48	CD1	-0.136	0.083	-0.196	63	C	-0.253	0.321	-0.355	78	CA	-0.305	0.317	-0.073
48	CG	-0.136	0.099	-0.175	64	N	-0.237	0.292	-0.327	78	CO	-0.304	0.300	-0.096
48	CB	-0.146	0.123	-0.184	64	ILE CD1	-0.314	0.305	-0.298	78	O	-0.325	0.291	-0.109
48	CO	-0.134	0.132	-0.210	64	CG1	-0.287	0.309	-0.309	79	N	-0.280	0.295	-0.102
48	O	-0.151	0.149	-0.227	64	CB	-0.263	0.302	-0.287	79	GLY CA	-0.275	0.279	-0.123
49	N	-0.150	0.169	-0.221	64	CG2	-0.257	0.315	-0.260	79	CO	-0.286	0.257	-0.116
49	VAL CG1	-0.167	0.140	-0.248	64	CA	-0.237	0.303	-0.300	79	O	-0.295	0.243	-0.135
49	CG2	-0.149	0.171	-0.291	64	CO	-0.214	0.293	-0.279	80	N	-0.285	0.253	-0.088
49	CB	-0.195	0.169	-0.319	64	O	-0.211	0.273	-0.276	80	VAL CG1	-0.262	0.216	-0.038
49	CA	-0.175	0.159	-0.295	65	N	-0.198	0.309	-0.266	80	CG2	-0.308	0.224	-0.026
49	CO	-0.186	0.154	-0.268	65	ASP OD1	-0.144	0.345	-0.231	80	CB	-0.286	0.230	-0.044
49	O	-0.211	0.140	-0.272	65	OD2	-0.143	0.346	-0.275	80	CA	-0.295	0.233	-0.076
50	N	-0.233	0.152	-0.269	65	CG	-0.146	0.335	-0.254	80	CO	-0.325	0.233	-0.081
50	LEU CD1	-0.264	0.115	-0.217	65	CB	-0.151	0.311	-0.258	80	O	-0.338	0.215	-0.087
50	CD2	-0.282	0.149	-0.197	65	CO	-0.178	0.318	-0.218	81	N	-0.336	0.253	-0.080
50	CG	-0.264	0.141	-0.218	65	O	-0.191	0.335	-0.221	81	TRP CD1	-0.419	0.291	-0.090
50	CB	-0.274	0.149	-0.248	66	N	-0.166	0.309	-0.194	81	CE1	-0.415	0.301	-0.116
50	CA	-0.259	0.142	-0.272	66	LEU CD1	-0.241	0.324	-0.159	81	CZ1	-0.437	0.310	-0.132
50	CO	-0.273	0.149	-0.302	66	CD2	-0.222	0.300	-0.193	81	CH	-0.462	0.311	-0.124
50	O	-0.275	0.169	-0.308	66	CG	-0.217	0.321	-0.175	81	CZ2	-0.467	0.301	-0.099
51	N	-0.282	0.132	-0.318	66	CB	-0.192	0.318	-0.154	81	CE2	-0.445	0.292	-0.082
51	LYS NZ	-0.299	0.112	-0.473	66	CA	-0.166	0.320	-0.166	81	NE	-0.445	0.282	-0.058
51	CE	-0.298	0.129	-0.451	66	CO	-0.145	0.308	-0.147	81	CD2	-0.418	0.276	-0.048
51	CD	-0.300	0.118	-0.422	66	O	-0.142	0.287	-0.149	81	CG	-0.403	0.281	-0.068
51	CG	-0.285	0.132	-0.398	67	N	-0.131	0.320	-0.128	81	CB	-0.372	0.277	-0.068
51	CB	-0.286	0.120	-0.369	67	GLY CA	-0.110	0.311	-0.107	81	CA	-0.364	0.257	-0.084
51	CA	-0.296	0.136	-0.347	67	CO	-0.083	0.314	-0.116	81	CO	-0.374	0.259	-0.116
51	CO	-0.326	0.132	-0.346	67	O	-0.064	0.301	-0.108	81	N	-0.398	0.258	-0.125
51	O	-0.335	0.113	-0.342	68	N	-0.081	0.331	-0.134	82	PHE N	-0.355	0.261	-0.133
52	N	-0.340	0.151	-0.350	68	ILE CD1	-0.052	0.337	-0.223	82	CD2	-0.329	0.312	-0.198
52	PHE CD2	-0.387	0.161	-0.286	68	CG1	-0.040	0.346	-0.194	82	CE2	-0.334	0.335	-0.201
52	CE2	-0.379	0.162	-0.256	68	CB	-0.061	0.351	-0.173	82	CZ	-0.353	0.345	-0.189
52	CZ	-0.357	0.170	-0.244	68	CG2	-0.061	0.374	-0.161	82	CE1	-0.367	0.333	-0.172
52	CE1	-0.340	0.180	-0.260	68	CA	-0.057	0.337	-0.146	82	CD1	-0.362	0.310	-0.168
52	CD1	-0.347	0.180	-0.290	68	CO	-0.040	0.348	-0.121	82	CG	-0.343	0.299	-0.181
52	CG	-0.370	0.170	-0.303	68	O	-0.016	0.346	-0.116	82	CB	-0.338	0.275	-0.175
52	CB	-0.376	0.171	-0.335	69	N	-0.054	0.361	-0.105	82	CA	-0.360	0.263	-0.164
52	CA	-0.368	0.151	-0.351	69	HIS CD2	-0.014	0.406	-0.115	82	CO	-0.368	0.240	-0.178
52	CO	-0.378	0.152	-0.383	69	NE2	-0.010	0.423	-0.133	82	O	-0.388	0.238	-0.195
52	O	-0.374	0.169	-0.397	69	CE1	-0.029	0.438	-0.133	83	ALA CB	-0.351	0.224	-0.169
53	N	-0.391	0.134	-0.393	69	ND1	-0.046	0.431	-0.116	83	CA	-0.340	0.185	-0.162
53	SER OG	-0.387	0.111	-0.462	69	CG	-0.036	0.410	-0.104	83	CO	-0.361	0.201	-0.179
53	CB	-0.381	0.128	-0.441	69	CB	-0.051	0.398	-0.084	83	CA	-0.387	0.192	-0.173
53	CA	-0.403	0.131	-0.423	69	CA	-0.043	0.373	-0.080	83	O	-0.404	0.186	-0.194
53	CO	-0.425	0.114	-0.427	69	CO	-0.052	0.362	-0.053	84	N	-0.390	0.191	-0.146
53	O	-0.425	0.097	-0.412	69	O	-0.071	0.370	-0.042	84	LYS NZ	-0.461	0.116	-0.097
54	N	-0.444	0.119	-0.448	70	N	-0.039	0.344	-0.044	84	CE	-0.470	0.139	-0.102
54	THR OG1	-0.478	0.139	-0.480	70	SER OG	0.003	0.328	-0.014	84	CG	-0.449	0.153	-0.114
54	CG2	-0.499	0.132	-0.439	70	CB	-0.020	0.320	-0.004	84	CD	-0.443	0.174	-0.097
54	CB	-0.489	0.122	-0.465	70	CA	-0.044	0.331	-0.020	84	CB	-0.419	0.186	-0.105
54	CA	-0.467	0.105	-0.457	70	CO	-0.058	0.344	0.002	84	CA	-0.415	0.183	-0.136
54	CO	-0.462	0.089	-0.480	70	O	-0.080	0.338	0.008	84	CO	-0.438	0.194	-0.154
54	O	-0.474	0.071	-0.483	71	N	-0.044	0.361	0.015	84	O	-0.457	0.183	-0.165
55	N	-0.445	0.095	-0.497	71	ARG NH1	-0.001	0.471	0.009	85	N	-0.436	0.216	-0.156
55	GLY CA	-0.438	0.082	-0.521	71	NH2	0.023	0.467	0.053	85	LYS NZ	-0.539	0.265	-0.154
55	CO	-0.423	0.096	-0.540	71	CZ	0.009	0.458	0.031	85	CE	-0.517	0.276	-0.165
55	O	-0.409	0.112	-0.530	71	NE	0.004	0.436	0.031	85	CO	-0.496	0.260	-0.171

TABLE 3 (cont.)

88	CA	-0.532	0.180	-0.203	104	CD	-0.352	0.143	-0.481	119	CG2	-0.007	0.315	0.066		
88	CG	-0.551	0.194	-0.222	104	GG	-0.357	0.168	-0.488	119	CB	-0.018	0.304	0.091		
88	N	-0.575	0.192	-0.224	104	CB	-0.347	0.183	-0.463	119	CA	0.004	0.292	0.112		
89	O	-0.540	0.210	-0.236	104	CA	-0.366	0.200	-0.456	119	CO	-0.009	0.283	0.136		
89	ASP	GD1	-0.564	0.256	-0.213	104	CO	-0.352	0.219	-0.436	119	O	-0.022	0.295	0.151	
89	OD2	-0.548	0.282	-0.233	104	O	-0.348	0.237	-0.446	120	N	-0.006	0.261	0.140		
89	CG	-0.552	0.263	-0.232	105	N	-0.346	0.213	-0.410	120	HIS	CD2	-0.067	0.246	0.108	
89	CB	-0.541	0.249	-0.253	105	LEU	CD1	-0.343	0.257	-0.326	120	NE2	-0.094	0.248	0.105	
89	CA	-0.554	0.226	-0.256	105	CD2	-0.387	0.258	-0.356	120	CE1	-0.103	0.239	0.127		
89	CO	-0.562	0.219	-0.288	105	CG	-0.358	0.254	-0.356	120	ND1	-0.083	0.232	0.145		
89	O	-0.572	0.232	-0.306	105	CB	-0.353	0.230	-0.367	120	CG	-0.060	0.236	0.134		
90	N	-0.555	0.198	-0.293	105	CA	-0.333	0.227	-0.388	120	CB	-0.034	0.229	0.149		
90	TYR	OH	-0.569	0.132	-0.438	105	CO	-0.310	0.214	-0.371	120	CA	-0.017	0.248	0.162	
90	CD2	-0.549	0.139	-0.361	105	O	-0.314	0.195	-0.362	120	CO	0.003	0.240	0.187		
90	CE2	-0.556	0.128	-0.388	106	N	-0.288	0.225	-0.367	120	O	-0.002	0.237	0.211		
90	CZ	-0.561	0.142	-0.412	106	PHE	CD2	-0.247	0.174	-0.392	121	N	0.027	0.236	0.179	
90	CE1	-0.558	0.164	-0.411	106	CE2	-0.253	0.160	-0.416	121	SER	OG	0.069	0.195	0.181	
90	CD1	-0.551	0.175	-0.384	106	CZ	-0.262	0.169	-0.442	121	CB	0.069	0.219	0.181		
90	CG	-0.546	0.162	-0.359	106	CE1	-0.265	0.191	-0.445	121	CA	0.049	0.228	0.199		
90	CB	-0.539	0.173	-0.330	106	CD1	-0.258	0.205	-0.422	121	CO	0.064	0.247	0.216		
90	CA	-0.561	0.188	-0.322	106	GG	-0.250	0.197	-0.395	121	O	0.061	0.250	0.242		
90	CO	-0.586	0.176	-0.318	106	CB	-0.243	0.213	-0.371	122	N	0.078	0.260	0.202		
90	N	-0.587	0.162	-0.299	106	CA	-0.264	0.216	-0.351	122	GLU	OE1	0.143	0.226	0.238	
91	GLY	CA	-0.606	0.181	-0.338	106	CO	-0.257	0.230	-0.325	122	CE2	0.120	0.230	0.195	
91	CG	-0.632	0.171	-0.339	106	C	-0.250	0.250	-0.327	122	CD	0.133	0.236	0.217		
91	O	-0.653	0.185	-0.328	106	N	-0.259	0.220	-0.300	122	CG	0.140	0.262	0.221		
91	N	-0.677	0.184	-0.340	107	LEU	CD1	-0.291	0.220	-0.209	122	CB	0.121	0.278	0.205	
92	ASP	OD1	-0.678	0.198	-0.306	107	CD2	-0.277	0.258	-0.216	122	CA	0.093	0.280	0.214	
92	OD2	-0.692	0.213	-0.238	107	CG	-0.273	0.234	-0.224	122	CO	0.085	0.304	0.208		
92	CG	-0.677	0.201	-0.247	107	CB	-0.278	0.231	-0.256	122	O	0.088	0.318	0.228		
92	CB	-0.655	0.212	-0.260	107	CA	-0.254	0.231	-0.272	123	N	0.076	0.308	0.181		
92	CA	-0.662	0.214	-0.293	107	CO	-0.229	0.221	-0.255	123	GLU	OE1	0.072	0.389	0.155	
92	CO	-0.659	0.238	-0.303	107	N	-0.228	0.201	-0.249	123	OE2	0.109	0.394	0.138		
92	C	-0.678	0.251	-0.306	108	GLN	NDE1	-0.111	0.251	-0.249	123	CD	0.095	0.385	0.151	
93	N	-0.635	0.243	-0.309	108	NDE2	-0.093	0.220	-0.231	123	CG	0.103	0.362	0.168		
93	GLN	NDE1	-0.599	0.330	-0.330	108	CD	-0.112	0.231	-0.241	123	CB	0.089	0.341	0.155	
93	NDE2	-0.566	0.305	-0.321	108	CG	-0.140	0.220	-0.242	123	CA	0.068	0.330	0.171		
93	CD	-0.589	0.312	-0.318	108	CB	-0.162	0.236	-0.246	123	O	0.042	0.334	0.151		
93	CG	-0.605	0.298	-0.299	108	CA	-0.185	0.229	-0.232	123	N	0.041	0.332	0.125		
93	CB	-0.604	0.274	-0.302	108	CO	-0.185	0.239	-0.202	124	ARG	NH1	-0.077	0.375	0.115	
93	CA	-0.629	0.265	-0.319	108	O	-0.177	0.258	-0.196	124	NH2	-0.120	0.381	0.119		
93	CO	-0.627	0.262	-0.351	109	N	-0.193	0.225	-0.184	124	CZ	-0.097	0.372	0.129		
93	O	-0.614	0.247	-0.351	109	ILE	CD1	-0.255	0.193	-0.150	124	NE	-0.096	0.358	0.152	
94	PRO	CG	-0.640	0.278	-0.367	109	CG1	-0.241	0.215	-0.150	124	CD	-0.072	0.346	0.163	
94	N	-0.676	0.303	-0.382	109	CB	-0.211	0.214	-0.139	124	CG	-0.048	0.360	0.164		
94	CD	-0.659	0.293	-0.357	109	CG2	-0.203	0.220	-0.107	124	CB	-0.023	0.346	0.173		
94	CB	-0.665	0.293	-0.408	109	CA	-0.194	0.230	-0.154	124	CA	-0.004	0.344	0.150		
94	CA	-0.641	0.275	-0.398	109	CO	-0.170	0.240	-0.136	124	CO	-0.007	0.365	0.132		
94	CO	-0.617	0.289	-0.408	109	O	-0.170	0.258	-0.123	124	O	-0.027	0.369	0.114		
94	O	-0.600	0.278	-0.418	110	N	-0.148	0.228	-0.138	125	N	0.013	0.379	0.137		
95	N	-0.616	0.311	-0.405	110	VAL	CG1	-0.138	0.221	-0.078	125	LEU	CD1	-0.016	0.440	0.130
95	SER	OG	-0.567	0.358	-0.406	110	CG2	-0.090	0.231	-0.076	125	CD2	0.021	0.457	0.161	
95	CB	-0.587	0.346	-0.396	110	CB	-0.115	0.222	-0.095	125	CG	0.007	0.436	0.153		
95	CA	-0.594	0.325	-0.413	110	CA	-0.122	0.234	-0.123	125	CB	0.026	0.420	0.141		
95	CO	-0.572	0.308	-0.411	110	CO	-0.102	0.232	-0.144	125	CA	0.015	0.401	0.122		
95	O	-0.563	0.301	-0.434	110	O	-0.094	0.214	-0.151	125	CO	0.029	0.397	0.096		
96	N	-0.562	0.302	-0.385	111	N	-0.094	0.252	-0.152	125	O	0.037	0.414	0.083		
96	PHE	CD2	-0.486	0.278	-0.344	111	THR	OG1	-0.108	0.274	-0.199	126	N	0.033	0.378	0.090
96	CE2	-0.462	0.267	-0.337	111	CG2	-0.065	0.278	-0.214	126	TRP	CD1	0.104	0.346	0.050	
96	CZ	-0.461	0.246	-0.325	111	CB	-0.081	0.276	-0.189	126	CE1	0.119	0.358	0.072		
96	CE1	-0.484	0.236	-0.321	111	CA	-0.074	0.255	-0.171	126	CZ1	0.145	0.360	0.069		
96	CD1	-0.508	0.247	-0.329	111	CO	-0.046	0.255	-0.155	126	CH	0.157	0.383	0.046		
96	CG	-0.509	0.260	-0.340	111	O	-0.027	0.247	-0.166	126	CZ2	0.143	0.341	0.024		
96	CB	-0.535	0.279	-0.348	112	N	-0.044	0.264	-0.129	126	CE2	0.117	0.338	0.027		
96	CA	-0.541	0.285	-0.379	112	ASP	OD1	0.029	0.308	-0.081	126	NE	0.100	0.327	0.010	
96	CO	-0.543	0.264	-0.397	112	OD2	0.025	0.274	-0.078	126	CD2	0.075	0.329	0.021		
96	O	-0.525	0.258	-0.411	112	CG	0.016	0.291	-0.086	126	CG	0.078	0.339	0.046		
97	N	-0.565	0.253	-0.396	112	CB	-0.011	0.290	-0.102	126	CB	0.057	0.345	0.066		
97	THR	CG1	-0.617	0.232	-0.418	112	CA	-0.019	0.266	-0.110	126	CA	0.047	0.369	0.066	
97	CG2	-0.597	0.215	-0.375	112	CO	-0.022	0.250	-0.085	126	CO	0.031	0.374	0.036		
97	CB	-0.596	0.218	-0.406	112	O	-0.024	0.258	-0.061	126	O	0.007	0.373	0.033		
97	CA	-0.571	0.232	-0.411	113	N	-0.021	0.229	-0.092	127	N	0.046	0.378	0.016		
97	CO	-0.567	0.234	-0.443	113	PRO	CG	-0.026	0.196	-0.120	127	GLN	NDE1	0.064	0.450	-0.031
97	O	-0.556	0.219	-0.455	113	CD	-0.017	0.220	-0.120	127	NDE2	0.027	0.454	-0.062		
98	N	-0.576	0.253	-0.454	113	CB	-0.024	0.190	-0.089	127	CD	0.043	0.442	-0.045		
98	ALA	CB	-0.582	0.284	-0.486	113	CO	-0.023	0.211	-0.071	127	CG	0.038	0.417	-0.044	
98	CA	-0.574	0.259	-0.484	113	CO	-0.022	0.213	-0.046	127	CB	0.040	0.408	-0.015		
98	CO	-0.545	0.261	-0.487	113	O	-0.007	0.220	-0.022	127	CA	0.035	0.383	-0.014		
98	O	-0.537	0.257	-0.510	114	N	0.022	0.209	-0.052	127	CO	0.046	0.371	-0.038		
99	N	-0.530	0.266	-0.463	114	ASP	OD1	0.102	0.194	-0.066	127	O	0.032	0.360	-0.057	
99	ILE	CC1	-0.474	0.323	-0.422	114	OD2	0.069	0.178	-0.055	128	N	0.072	0.372	-0.037	
99	CG1	-0.487	0.308	-0.446	114	CG	0.081	0.195	-0.057	128	GLU	OE1	0.146	0.423	-0.072	
99	CB	-0.490	0.283	-0.436	114	CB	0.069	0.216	-0.047	128	CE2	0.165	0.394	-0.051		
99	CG2	-0.465	0.271	-0.423	114	CA	0.046	0.210	-0.031	128	CD	0.147	0.405	-0.061		
99	CA	-0.501	0.268	-0.460	114	CO	0.045	0.228	-0.008	128	CG	0.118	0.396	-0.061		
99	CO	-0.486	0.246	-0.458	114	O	0.051	0.224	0.018	128	CB	0.115	0.371	-0.056		
99	O	-0.470	0.241	-0.475	115	N	0.037	0.248	-0.019	128	CA	0.087	0.362	-0.058		
100	N	-0.492	0.233	-0.437	115	GLY	CA	0.034	0.268	-0.001	128	CO	0.087	0.337	-0.054	
100	LEU	CC1	-0.444	0.206	-0.379	115	CO									

TABLE 3 (cont.)

132		G	0.276	0.270	-0.111	150		CG	0.102	0.504	-0.401	166		CB	0.166	0.449	-0.229
133		N	0.311	0.262	-0.078	150		G	0.110	0.518	-0.382	166		CA	0.163	0.430	-0.207
133	GLY	CA	0.331	0.263	-0.098	151		N	0.109	0.482	-0.400	166		CO	0.174	0.409	-0.218
133		CD	0.338	0.287	-0.104	151	TRP	CC1	0.162	0.428	-0.345	166		O	0.160	0.397	-0.236
133		G	0.321	0.302	-0.108	151		CE1	0.146	0.426	-0.325	167		N	0.198	0.404	-0.207
134		N	0.363	0.290	-0.105	151		CZ1	0.157	0.415	-0.299	167	GLY	CA	0.212	0.384	-0.214
134	SER	CG	0.345	0.335	-0.144	151		CH	0.183	0.407	-0.295	167		CO	0.231	0.387	-0.236
134		CB	0.352	0.329	-0.115	151		CZ2	0.199	0.408	-0.315	167		O	0.232	0.405	-0.249
134		CA	0.374	0.312	-0.111	151		CE2	0.188	0.419	-0.340	168		N	0.246	0.369	-0.238
134		CD	0.395	0.322	-0.089	151		NE	0.199	0.422	-0.362	168	LYS	NZ	0.265	0.277	-0.218
134		C	0.407	0.310	-0.069	151		CC2	0.181	0.434	-0.383	168		CE	0.254	0.290	-0.243
135		N	0.399	0.343	-0.092	151		CG	0.158	0.437	-0.373	168		CD	0.274	0.307	-0.250
135	GLY	CA	0.419	0.356	-0.073	151		CB	0.133	0.449	-0.387	168		CG	0.272	0.329	-0.235
135		CO	0.403	0.370	-0.053	151		CA	0.128	0.473	-0.378	168		CB	0.283	0.348	-0.251
135		G	0.400	0.390	-0.055	151		CO	0.155	0.483	-0.368	168		CA	0.265	0.368	-0.258
136		N	0.394	0.357	-0.033	151		O	0.169	0.490	-0.386	168		CO	0.256	0.367	-0.290
136	GLY	CA	0.379	0.367	-0.011	152		N	0.161	0.484	-0.340	168		O	0.271	0.374	-0.308
136		CO	0.367	0.348	0.004	152	GLY	CA	0.186	0.493	-0.326	169		N	0.232	0.360	-0.297
136		O	0.361	0.329	-0.008	152		CO	0.185	0.518	-0.330	169	TYR	OH	0.245	0.254	-0.311
137		N	0.363	0.352	0.031	152		O	0.206	0.530	-0.328	169		CD2	0.258	0.313	-0.321
137	GLU	OE1	0.402	0.369	0.077	153		N	0.161	0.526	-0.335	169		CE2	0.262	0.291	-0.312
137		OE2	0.433	0.343	0.084	153	LYS	NZ	0.114	0.566	-0.469	169		CZ	0.242	0.276	-0.320
137		CO	0.411	0.350	0.084	153		CE	0.135	0.563	-0.445	169		CE1	0.220	0.281	-0.337
137		CG	0.389	0.335	0.094	153		CO	0.124	0.560	-0.417	169		CD1	0.216	0.304	-0.346
137		CB	0.360	0.342	0.081	153		CG	0.145	0.562	-0.392	169		CG	0.236	0.320	-0.338
137		CA	0.352	0.337	0.049	153		CB	0.134	0.556	-0.364	169		CB	0.232	0.344	-0.348
137		O	0.322	0.339	0.043	153		CO	0.155	0.550	-0.339	169		CA	0.219	0.358	-0.327
137		G	0.307	0.323	0.046	153		CO	0.148	0.560	-0.311	169		O	0.189	0.357	-0.329
138		N	0.314	0.340	0.035	154		N	0.150	0.582	-0.292	169		O	0.178	0.353	-0.308
138	CYH	SG	0.296	0.411	0.039	154		NZ	0.207	0.685	-0.256	170	ALA	CB	0.142	0.372	-0.355
138		CB	0.285	0.388	0.013	154	LYS	NZ	0.207	0.685	-0.256	170		CO	0.149	0.361	-0.363
138		CA	0.286	0.366	0.023	154		CE	0.206	0.660	-0.260	170		CA	0.138	0.337	-0.366
138		CO	0.273	0.349	0.007	154		CD	0.178	0.653	-0.272	170		O	0.141	0.326	-0.388
138		O	0.283	0.341	-0.014	154		CG	0.177	0.628	-0.278	170		N	0.127	0.330	-0.344
139		N	0.249	0.343	0.013	154		CB	0.149	0.621	-0.289	171	GLU	OE1	0.072	0.321	-0.422
139	VAL	CG1	0.199	0.300	0.012	154		CA	0.144	0.596	-0.286	171		OE2	0.054	0.329	-0.383
139		CG2	0.238	0.311	0.048	154		CO	0.116	0.587	-0.284	171		CG	0.067	0.318	-0.398
139		CB	0.228	0.307	0.016	154		O	0.099	0.590	-0.305	171		CD	0.081	0.296	-0.387
139		CA	0.232	0.326	-0.004	155		N	0.113	0.577	-0.260	171		CB	0.110	0.298	-0.374
139		CO	0.205	0.335	-0.017	155	GLY	CA	0.088	0.568	-0.254	171		CA	0.115	0.308	-0.343
139		O	0.194	0.349	-0.005	155		CO	0.098	0.547	-0.237	171		CO	0.133	0.291	-0.326
140		N	0.197	0.325	-0.042	155		O	0.082	0.531	-0.233	171		O	0.128	0.271	-0.326
140	GLY	CA	0.171	0.330	-0.058	156	ALA	CB	0.123	0.548	-0.226	171		N	0.153	0.301	-0.310
140		CO	0.171	0.348	-0.082	156		CA	0.131	0.510	-0.231	172	SER	OG	0.182	0.323	-0.265
140		O	0.187	0.363	-0.080	156		CO	0.136	0.530	-0.208	172		CB	0.193	0.303	-0.275
141		N	0.153	0.345	-0.104	156		CO	0.166	0.530	-0.203	172		CA	0.173	0.288	-0.291
141	VAL	CG1	0.185	0.341	-0.153	156		O	0.179	0.538	-0.222	172		CO	0.157	0.275	-0.271
141		CG2	0.151	0.365	-0.183	157		N	0.176	0.521	-0.178	172		O	0.163	0.255	-0.263
141		CB	0.157	0.350	-0.156	157	SER	OG	0.237	0.515	-0.127	172		N	0.137	0.286	-0.263
141		CA	0.149	0.360	-0.129	157		CB	0.211	0.523	-0.137	173	GLU	OE1	0.145	0.292	-0.174
141		CO	0.121	0.367	-0.133	157		CA	0.204	0.519	-0.169	173		OE2	0.165	0.306	-0.208
141		O	0.103	0.354	-0.127	157		CO	0.220	0.499	-0.177	173		CD	0.146	0.302	-0.196
142		N	0.116	0.387	-0.143	157		O	0.208	0.481	-0.183	173		CG	0.118	0.310	-0.208
142	ASP	OD1	0.040	0.421	-0.155	158		N	0.245	0.502	-0.175	173		CB	0.103	0.295	-0.231
142		OD2	0.058	0.445	-0.127	158	SER	CG	0.274	0.512	-0.215	173		CA	0.119	0.277	-0.244
142		CG	0.060	0.430	-0.143	158		CB	0.284	0.493	-0.199	173		CO	0.102	0.258	-0.257
142		CB	0.088	0.423	-0.145	158		CA	0.263	0.484	-0.181	173		O	0.086	0.261	-0.279
142		CA	0.090	0.397	-0.148	158		CO	0.279	0.473	-0.155	173		N	0.107	0.239	-0.243
142		CO	0.076	0.387	-0.176	158		G	0.295	0.458	-0.158	174	THR	OG1	0.102	0.180	-0.246
142		O	0.078	0.396	-0.199	159		N	0.273	0.481	-0.130	174		CG2	0.092	0.199	-0.204
143		N	0.062	0.369	-0.172	159	SER	OG	0.297	0.511	-0.091	174		CB	0.105	0.201	-0.230
143	ALA	CB	0.026	0.344	-0.184	159		CB	0.289	0.490	-0.080	174		CA	0.092	0.218	-0.251
143		CA	0.047	0.357	-0.196	159		CA	0.285	0.473	-0.103	174		CO	0.063	0.221	-0.249
143		CO	0.031	0.372	-0.219	159		CO	0.268	0.454	-0.094	174		O	0.047	0.207	-0.260
143		O	0.030	0.367	-0.245	159		O	0.244	0.457	-0.090	174		N	0.057	0.238	-0.233
144		N	0.020	0.389	-0.208	160	PRO	CG	0.280	0.434	-0.091	175	GLU	OE1	0.077	0.245	-0.181
144	ASN	NOD1	-0.047	0.413	-0.256	160		N	0.310	0.405	-0.098	175		OE2	0.071	0.225	-0.144
144		NOD2	-0.051	0.443	-0.229	160		CD	0.307	0.430	-0.095	175		CD	0.064	0.237	-0.164
144		CG	-0.040	0.425	-0.231	160		CO	0.286	0.396	-0.086	175		CG	0.035	0.244	-0.172
144		CB	-0.018	0.416	-0.210	160		CA	0.267	0.414	-0.082	175		CB	0.029	0.257	-0.200
144		CA	0.004	0.405	-0.226	160		CO	0.256	0.417	-0.054	175		CA	0.030	0.243	-0.228
144		CO	0.020	0.423	-0.238	160		O	0.237	0.406	-0.047	175		CO	0.016	0.256	-0.253
144		O	0.011	0.441	-0.246	161		N	0.269	0.433	-0.037	175		N	-0.008	0.254	-0.260
145		N	0.045	0.416	-0.240	161	CYH	SG	0.311	0.430	0.022	175		O	0.031	0.270	-0.266
145	ARG	NH1	0.043	0.496	-0.167	161		CB	0.285	0.450	0.008	176	VAL	CG1	0.035	0.319	-0.263
145		NH2	0.058	0.531	-0.179	161		CA	0.262	0.439	-0.009	176		CG2	0.030	0.320	-0.317
145		CZ	0.059	0.509	-0.180	161		CO	0.238	0.455	-0.011	176		CB	0.036	0.306	-0.290
145		NE	0.078	0.499	-0.195	161		O	0.226	0.458	0.010	176		CA	0.021	0.284	-0.290
145		CD	0.082	0.475	-0.196	162		N	0.232	0.463	-0.037	176		CO	0.020	0.271	-0.318
145		CG	0.079	0.467	-0.227	162	SER	EG	0.201	0.513	-0.072	176		O	-0.000	0.271	-0.336
145		CB	0.082	0.442	-0.228	162		CB	0.212	0.491	-0.072	177	LYS	NZ	0.042	0.260	-0.320
145		CA	0.064	0.430	-0.252	162		CA	0.210	0.479	-0.044						



TABLE 3 (cont.)

182	PHE	CD2	-0.127	0.152	-0.410	196	O	-0.183	0.450	-0.111	211	CR	-0.047	0.622	-0.532		
182		CE2	-0.152	0.141	-0.414	197	N	-0.148	0.432	-0.087	211	CA	-0.061	0.641	-0.518		
182		CZ	-0.174	0.152	-0.416	197	GLY	CA	-0.137	0.452	-0.070	211	CO	-0.089	0.635	-0.515	
182		CE1	-0.173	0.175	-0.415	197	CD	-0.158	0.463	-0.055	211	C	-0.094	0.617	-0.502		
182		CD1	-0.148	0.186	-0.412	197	C	-0.177	0.453	-0.047	212	N	-0.107	0.649	-0.527		
182		CG	-0.126	0.174	-0.408	198	N	-0.155	0.485	-0.053	212	SER	OG	-0.151	0.684	-0.521	
182		CB	-0.100	0.187	-0.405	198	TYR	OH	-0.100	0.570	0.039	212	CB	-0.150	0.667	-0.541	
182		CA	-0.088	0.190	-0.433	198	CD2	-0.156	0.540	0.001	212	CA	-0.135	0.647	-0.527		
182		C	-0.106	0.204	-0.454	198	CF2	-0.140	0.558	0.012	212	CO	-0.144	0.625	-0.543		
182		CG	-0.121	0.194	-0.474	198	C	-0.116	0.553	0.028	212	O	-0.137	0.621	-0.567		
183		N	-0.104	0.225	-0.450	198	CE1	-0.108	0.531	0.032	213	N	-0.158	0.612	-0.529		
183	VAL	CG1	-0.134	0.266	-0.430	198	CD1	-0.125	0.513	0.021	213	LEU	CD1	-0.188	0.543	-0.532	
183		CG2	-0.128	0.284	-0.478	198	CG	-0.149	0.518	0.004	213	CD2	-0.211	0.556	-0.493		
183		CB	-0.119	0.265	-0.456	198	CB	-0.167	0.499	-0.007	213	CO	-0.188	0.560	-0.509		
183		CA	-0.119	0.242	-0.468	198	CA	-0.174	0.500	-0.040	213	CB	-0.190	0.584	-0.521		
183		CO	-0.114	0.240	-0.499	198	CC	-0.171	0.524	-0.050	213	CA	-0.169	0.590	-0.540		
183		C	-0.133	0.240	-0.519	198	O	-0.150	0.531	-0.057	213	CO	-0.184	0.596	-0.570		
184		N	-0.089	0.238	-0.503	199	N	-0.194	0.535	-0.052	213	C	-0.193	0.615	-0.575		
184	LYS	NZ	0.034	0.261	-0.549	199	SER	OG	-0.174	0.554	-0.104	214	N	-0.185	0.579	-0.588	
184		CE	0.006	0.260	-0.560	199	CG	-0.172	0.564	-0.076	214	PRO	CG	-0.188	0.543	-0.607	
184		CD	-0.010	0.253	-0.536	199	CA	-0.196	0.558	-0.062	214	CO	-0.175	0.557	-0.581		
184		CG	-0.037	0.244	-0.549	199	CO	-0.219	0.563	-0.084	214	CB	-0.193	0.559	-0.631		
184		CB	-0.051	0.234	-0.526	199	O	-0.222	0.582	-0.096	214	CA	-0.198	0.581	-0.618		
184		CA	-0.080	0.236	-0.531	200	N	-0.236	0.546	-0.089	214	CO	-0.228	0.584	-0.617		
184		CO	-0.091	0.215	-0.549	200	GLU	OE1	-0.306	0.558	-0.026	214	O	-0.241	0.599	-0.631	
184		O	-0.101	0.217	-0.574	200	OE2	-0.312	0.584	-0.059	214	N	-0.237	0.569	-0.600		
185		N	-0.088	0.196	-0.535	200	CD	-0.303	0.567	-0.048	215	ASP	OD1	-0.239	0.531	-0.613	
185	ASP	CD1	-0.047	0.163	-0.519	200	CG	-0.285	0.552	-0.064	215	OD2	-0.275	0.514	-0.625		
185		CD2	-0.062	0.139	-0.493	200	CB	-0.283	0.558	-0.095	215	CG	-0.263	0.530	-0.614		
185		CG	-0.066	0.153	-0.512	200	CA	-0.260	0.548	-0.109	215	CB	-0.280	0.547	-0.621		
185		CB	-0.094	0.157	-0.523	200	CO	-0.255	0.562	-0.135	215	CA	-0.245	0.569	-0.595		
185		CA	-0.086	0.175	-0.547	200	O	-0.259	0.582	-0.135	215	CO	-0.268	0.575	-0.565		
185		CO	-0.126	0.178	-0.558	201	N	-0.245	0.550	-0.155	215	O	-0.278	0.562	-0.549		
185		O	-0.134	0.176	-0.584	201	ASP	OD1	-0.217	0.604	-0.173	216	N	-0.259	0.596	-0.557	
186		N	-0.140	0.183	-0.538	201	OD2	-0.179	0.592	-0.176	216	LYS	NZ	-0.213	0.687	-0.475	
186	HIS	CD2	-0.226	0.213	-0.525	201	CG	-0.202	0.588	-0.175	216	CE	-0.237	0.675	-0.472		
186		NE2	-0.249	0.204	-0.521	201	CB	-0.209	0.564	-0.176	216	CO	-0.247	0.663	-0.499		
186		CE1	-0.246	0.183	-0.513	201	CA	-0.239	0.560	-0.182	216	CG	-0.244	0.637	-0.495		
186		ND1	-0.221	0.178	-0.510	201	CO	-0.243	0.544	-0.207	216	CB	-0.243	0.626	-0.523		
186		CG	-0.207	0.197	-0.518	201	O	-0.235	0.525	-0.205	216	CA	-0.260	0.605	-0.529		
186		CB	-0.178	0.197	-0.517	202	N	-0.257	0.553	-0.230	216	CO	-0.288	0.611	-0.525		
186		CA	-0.168	0.187	-0.544	202	SER	OG	-0.295	0.518	-0.286	216	O	-0.297	0.609	-0.501	
186		CO	-0.175	0.203	-0.569	202	CB	-0.291	0.536	-0.267	217	N	-0.302	0.619	-0.548		
186		O	-0.185	0.195	-0.593	202	CA	-0.262	0.540	-0.257	217	THR	OG1	-0.333	0.655	-0.583	
187		N	-0.170	0.224	-0.564	202	CO	-0.249	0.553	-0.279	217	CG2	-0.371	0.631	-0.585		
187	GLY	CA	-0.175	0.242	-0.585	202	O	-0.256	0.572	-0.287	217	CB	-0.342	0.633	-0.580		
187		CO	-0.204	0.246	-0.598	203	N	-0.229	0.542	-0.287	217	CA	-0.330	0.626	-0.550		
187		O	-0.210	0.251	-0.623	203	LEU	CD1	-0.183	0.557	-0.241	217	CO	-0.345	0.607	-0.537	
188		N	-0.221	0.243	-0.579	203	CD2	-0.200	0.589	-0.271	217	O	-0.356	0.611	-0.515		
188	LEU	CD1	-0.291	0.247	-0.631	203	CG	-0.180	0.570	-0.268	218	N	-0.345	0.588	-0.550		
188		CD2	-0.306	0.210	-0.620	203	CB	-0.185	0.555	-0.294	218	GLU	OE1	-0.365	0.524	-0.605	
188		CG	-0.282	0.224	-0.623	203	CA	-0.213	0.551	-0.308	218	OE2	-0.373	0.492	-0.585		
188		CB	-0.264	0.224	-0.595	203	CO	-0.211	0.536	-0.334	218	CD	-0.368	0.512	-0.585		
188		CA	-0.249	0.246	-0.586	203	O	-0.210	0.516	-0.332	218	CG	-0.364	0.526	-0.556		
188		CO	-0.259	0.260	-0.562	204	N	-0.212	0.547	-0.359	218	CB	-0.349	0.548	-0.558		
188		O	-0.282	0.257	-0.557	204	TYR	OH	-0.255	0.636	-0.412	218	CA	-0.359	0.568	-0.541	
189		N	-0.242	0.274	-0.550	204	CD2	-0.223	0.584	-0.429	218	CO	-0.352	0.560	-0.510		
189	PHE	CD2	-0.233	0.263	-0.467	204	CE2	-0.229	0.608	-0.429	218	O	-0.369	0.559	-0.494		
189		CE2	-0.226	0.244	-0.450	204	CZ	-0.249	0.614	-0.412	219	N	-0.327	0.556	-0.503		
189		CZ	-0.204	0.232	-0.454	204	CE1	-0.260	0.600	-0.396	219	LEU	CD1	-0.249	0.522	-0.485	
189		CE1	-0.188	0.239	-0.473	204	CD1	-0.254	0.576	-0.396	219	CD2	-0.294	0.506	-0.486		
189		CG1	-0.195	0.259	-0.489	204	CG	-0.234	0.569	-0.413	219	CG	-0.274	0.523	-0.471		
189		CG2	-0.217	0.270	-0.487	204	CB	-0.228	0.544	-0.413	219	CB	-0.286	0.546	-0.472		
189		CB	-0.223	0.291	-0.504	204	CA	-0.210	0.536	-0.386	219	CA	-0.316	0.548	-0.474		
189		CA	-0.247	0.289	-0.527	204	CO	-0.180	0.535	-0.383	219	CO	-0.323	0.565	-0.452		
189		CO	-0.254	0.312	-0.540	204	O	-0.167	0.535	-0.359	219	O	-0.331	0.558	-0.429		
189		O	-0.242	0.320	-0.559	205	N	-0.171	0.534	-0.408	220	N	-0.320	0.586	-0.459		
190		N	-0.274	0.322	-0.530	205	PRO	CG	-0.167	0.505	-0.441	220	ASN	NDD1	-0.341	0.663	-0.442
190	LYS	NZ	-0.305	0.295	-0.633	205	CD	-0.186	0.519	-0.427	220	NDD2	-0.306	0.648	-0.415		
190		CE	-0.313	0.293	-0.604	205	CB	-0.140	0.513	-0.429	220	CG	-0.323	0.646	-0.436		
190		CD	-0.325	0.315	-0.596	205	CA	-0.142	0.533	-0.410	220	CB	-0.324	0.627	-0.456		
190		CG	-0.321	0.320	-0.564	205	CO	-0.125	0.552	-0.418	220	CA	-0.327	0.605	-0.441		
190		CB	-0.310	0.343	-0.558	205	O	-0.135	0.671	-0.425	220	CO	-0.355	0.600	-0.435		
190		CA	-0.283	0.344	-0.539	206	N	-0.100	0.547	-0.416	220	O	-0.360	0.600	-0.410		
190		CO	-0.286	0.360	-0.515	206	TYR	OH	-0.074	0.517	-0.547	221	N	-0.371	0.596	-0.459	
190		O	-0.296	0.379	-0.519	206	CD2	-0.079	0.536	-0.474	221	GLU	OE1	-0.465	0.618	-0.464	
191		N	-0.278	0.352	-0.489	206	CE2	-0.075	0.521	-0.497	221	OE2	-0.466	0.582	-0.464		
191	ALA	CB	-0.304	0.360	-0.451	206	CZ	-0.077	0.531	-0.524	221	CD	-0.459	0.600	-0.471		
191		CA	-0.278	0.364	-0.462	206	CE1	-0.082	0.552	-0.529	221	CG	-0.440	0.602	-0.495		
191		CO	-0.257	0.354	-0.440	206	CD1	-0.086	0.567	-0.506	221	CB	-0.414	0.589	-0.489		
191		O	-0.257	0.334	-0.433	206	CG	-0.084	0.558	-0.478	221	CA	-0.399	0.590	-0.458		
192		N	-0.240	0.370	-0.429	206	CB	-0.088	0.574	-0.453	221	CO	-0.405	0.570	-0.441		
192	PHE	CD2	-0.162	0.336	-0.403	206	CA	-0.080	0.563	-0.424	221	O	-0.421	0.571	-0.423		
192		CE2	-0.140	0.327	-0.384	206	CO	-0.072	0.575	-0.396	222	N	-0.392	0.552	-0.447		
192		CZ	-0.126	0.340	-0.365	206	O	-0.078	0.595	-0.392	222	VAL	CG1	-0.394	0.502	-0.472	
192		CE1	-0.132	0.361	-0.361	207	N	-0.057	0.563	-0.376							

TABLE 3 (cont.)

227	CO	-0.435	0.556	-0.251	243	CO	-0.100	0.615	-0.297	260	N	-0.102	0.434	-0.522			
227	N	-0.437	0.555	-0.225	243	C	-0.101	0.626	-0.274	260	ASN	NDD1	-0.073	0.477	-0.555		
228	N	-0.454	0.562	-0.271	244	N	-0.087	0.620	-0.318	260	NDD2	-0.038	0.477	-0.520			
228	ALA	CB	-0.499	0.573	-0.293	244	CD1	-0.009	0.654	-0.350	260	CG	-0.055	0.466	-0.534		
228	CA	-0.481	0.567	-0.266	244	CG1	-0.036	0.661	-0.342	260	CB	-0.057	0.441	-0.533			
228	CG	-0.493	0.547	-0.252	244	CB	-0.057	0.643	-0.345	260	CA	-0.085	0.433	-0.545			
228	O	-0.502	0.549	-0.229	244	CG2	-0.081	0.647	-0.367	260	CO	-0.087	0.410	-0.557			
229	N	-0.492	0.528	-0.266	244	CA	-0.070	0.640	-0.317	260	O	-0.078	0.405	-0.580			
229	ALA	CB	-0.495	0.487	-0.274	244	CO	-0.087	0.661	-0.315	261	N	-0.099	0.395	-0.543		
229	CA	-0.502	0.507	-0.257	244	O	-0.077	0.680	-0.315	261	GLU	OE1	-0.053	0.324	-0.549		
229	CO	-0.493	0.504	-0.225	245	N	-0.112	0.657	-0.312	261	CE2	-0.048	0.359	-0.535			
229	O	-0.508	0.503	-0.207	245	SER	OG	-0.170	0.682	-0.343	261	CC	-0.059	0.341	-0.537		
230	N	-0.467	0.503	-0.219	245	CB	-0.157	0.667	-0.322	261	CG	-0.084	0.337	-0.522			
230	LEU	CD1	-0.397	0.512	-0.148	245	CA	-0.131	0.675	-0.309	261	CB	-0.106	0.354	-0.528		
230	CD2	-0.381	0.481	-0.176	245	CO	-0.135	0.683	-0.279	261	CA	-0.103	0.372	-0.551			
230	CG	-0.405	0.492	-0.167	245	O	-0.130	0.702	-0.270	261	C	-0.126	0.371	-0.575			
230	CB	-0.424	0.499	-0.193	246	N	-0.144	0.667	-0.263	261	O	-0.130	0.355	-0.592			
230	CA	-0.453	0.500	-0.190	246	SER	OG	-0.190	0.675	-0.211	262	N	-0.141	0.390	-0.577		
230	CO	-0.455	0.518	-0.166	246	CB	-0.175	0.660	-0.227	262	GLY	CA	-0.164	0.393	-0.598		
230	O	-0.446	0.514	-0.141	246	CA	-0.150	0.670	-0.234	262	CO	-0.189	0.390	-0.584			
231	N	-0.468	0.536	-0.176	246	CO	-0.125	0.664	-0.213	262	O	-0.209	0.382	-0.598			
231	LYS	NZ	-0.456	0.657	-0.155	246	O	-0.120	0.674	-0.190	263	N	-0.187	0.397	-0.558		
231	CE	-0.475	0.639	-0.148	247	N	-0.111	0.648	-0.223	263	ILE	CD1	-0.171	0.355	-0.539		
231	CD	-0.474	0.619	-0.167	247	ILE	CD1	-0.044	0.626	-0.167	263	CG1	-0.197	0.356	-0.524		
231	CG	-0.471	0.597	-0.150	247	CG1	-0.063	0.607	-0.178	263	CB	-0.203	0.380	-0.515			
231	CB	-0.468	0.578	-0.170	247	CB	-0.083	0.614	-0.204	263	CG2	-0.228	0.382	-0.501			
231	CA	-0.473	0.555	-0.157	247	CG2	-0.075	0.608	-0.233	263	CA	-0.209	0.396	-0.540			
231	CO	-0.501	0.555	-0.150	247	CA	-0.087	0.639	-0.207	263	CO	-0.216	0.420	-0.530			
231	C	-0.508	0.570	-0.134	247	CO	-0.062	0.650	-0.216	263	O	-0.201	0.430	-0.512			
232	N	-0.516	0.539	-0.162	247	C	-0.061	0.671	-0.219	264	N	-0.239	0.427	-0.542			
232	SER	CG	-0.552	0.505	-0.191	248	N	-0.044	0.636	-0.220	264	LYS	NZ	-0.328	0.477	-0.633	
232	CB	-0.559	0.527	-0.185	248	TYR	CH	0.092	0.696	-0.215	264	CE	-0.303	0.470	-0.616		
232	CA	-0.543	0.536	-0.158	248	CE2	0.026	0.681	-0.202	264	CC	-0.299	0.483	-0.588			
232	CO	-0.551	0.519	-0.136	248	CE1	0.048	0.694	-0.205	264	CG	-0.273	0.477	-0.570			
232	O	-0.574	0.516	-0.132	248	CZ	0.070	0.683	-0.212	264	CB	-0.273	0.454	-0.558			
232	N	-0.531	0.508	-0.122	248	CE1	0.071	0.661	-0.217	264	CA	-0.250	0.448	-0.536			
233	LEU	CD1	-0.552	0.442	-0.095	248	CD1	0.048	0.647	-0.213	264	CO	-0.260	0.448	-0.507		
233	CD2	-0.512	0.430	-0.116	248	CG	0.026	0.658	-0.206	264	O	-0.258	0.465	-0.491			
233	CG	-0.532	0.449	-0.114	248	CB	0.002	0.644	-0.202	265	N	-0.272	0.429	-0.501			
233	CB	-0.517	0.470	-0.103	248	CA	-0.019	0.642	-0.229	265	ARG	NH1	-0.370	0.422	-0.572		
233	CA	-0.534	0.491	-0.100	248	CC	-0.010	0.626	-0.251	265	NH2	-0.367	0.460	-0.564			
233	CO	-0.529	0.503	-0.071	248	C	-0.022	0.607	-0.255	265	CZ	-0.359	0.440	-0.558			
233	O	-0.546	0.504	-0.055	249	N	0.009	0.633	-0.264	265	NE	-0.340	0.437	-0.536			
234	TYR	OH	-0.505	0.511	-0.065	249	GLN	NCE1	0.085	0.648	-0.334	265	CC	-0.331	0.415	-0.525	
234	CD2	-0.503	0.475	-0.008	249	NCE2	0.083	0.611	-0.339	265	CG	-0.330	0.414	-0.493			
234	CE2	-0.510	0.452	-0.009	249	CG	0.073	0.630	-0.334	265	CB	-0.302	0.406	-0.479			
234	CZ	-0.495	0.438	-0.025	249	CB	0.046	0.630	-0.325	265	CA	-0.283	0.426	-0.474			
234	CE1	-0.475	0.446	-0.038	249	CG	0.045	0.630	-0.294	265	CC	-0.261	0.421	-0.449			
234	CD1	-0.468	0.469	-0.037	249	CA	0.020	0.619	-0.286	265	O	-0.260	0.402	-0.437			
234	CG	-0.483	0.484	-0.021	249	O	0.025	0.596	-0.271	266	SER	N	-0.245	0.438	-0.442		
234	CB	-0.475	0.509	-0.020	249	N	0.038	0.595	-0.246	266	CG	-0.193	0.419	-0.447			
234	CA	-0.496	0.523	-0.038	250	ALA	CP	-0.011	0.549	-0.268	266	CB	-0.198	0.439	-0.431		
234	CO	-0.487	0.547	-0.042	250	CA	0.016	0.556	-0.276	266	CO	-0.224	0.437	-0.418			
234	C	-0.482	0.560	-0.021	250	CC	0.022	0.543	-0.301	266	O	-0.221	0.477	-0.408			
235	N	-0.485	0.553	-0.068	250	C	0.005	0.542	-0.324	266	N	-0.227	0.453	-0.372			
235	GLY	CA	-0.476	0.575	-0.067	251	SER	CG	0.045	0.533	-0.298	267	PHE	CD2	-0.280	0.459	-0.390
235	CO	-0.447	0.579	-0.067	251	CG	0.101	0.521	-0.301	267	CE2	-0.297	0.461	-0.416			
235	O	-0.440	0.595	-0.050	251	CB	0.081	0.526	-0.325	267	CZ	-0.306	0.481	-0.427			
236	N	-0.431	0.564	-0.076	251	CA	0.054	0.519	-0.320	267	CE1	-0.299	0.500	-0.413			
236	THR	CG1	-0.397	0.530	-0.047	251	CO	0.056	0.494	-0.312	267	CD1	-0.282	0.459	-0.387		
236	CG2	-0.361	0.543	-0.070	251	C	0.059	0.488	-0.286	267	CG	-0.273	0.475	-0.375			
236	CB	-0.390	0.542	-0.071	252	N	0.052	0.480	-0.334	267	CB	-0.255	0.479	-0.347			
236	CA	-0.403	0.565	-0.069	252	GLY	CA	0.053	0.456	-0.331	267	CO	-0.227	0.470	-0.350		
236	CO	-0.394	0.580	-0.092	252	CC	0.030	0.447	-0.317	267	O	-0.217	0.458	-0.322			
236	O	-0.397	0.575	-0.118	252	C	0.032	0.431	-0.301	267	N	-0.219	0.438	-0.319			
237	N	-0.383	0.599	-0.082	253	N	0.008	0.459	-0.325	268	N	-0.207	0.472	-0.301			
237	SER	OG	-0.352	0.653	-0.104	253	GLY	CA	-0.017	0.453	-0.314	268	VAL	CG1	-0.153	0.471	-0.291
237	CB	-0.363	0.637	-0.085	253	CC	-0.033	0.438	-0.335	268	CG2	-0.153	0.451	-0.243			
237	CA	-0.372	0.616	-0.101	253	C	-0.028	0.437	-0.361	268	CB	-0.167	0.468	-0.265			
237	CO	-0.349	0.607	-0.114	254	N	-0.051	0.426	-0.325	268	CA	-0.196	0.465	-0.273			
237	O	-0.327	0.603	-0.099	254	ILE	CD1	-0.082	0.437	-0.296	268	CC	-0.212	0.475	-0.251		
238	N	-0.354	0.604	-0.142	254	CG1	-0.090	0.412	-0.295	268	O	-0.218	0.495	-0.251			
238	TYR	CH	-0.301	0.495	-0.123	254	CB	-0.087	0.399	-0.323	269	N	-0.220	0.461	-0.232		
238	CD2	-0.321	0.535	-0.177	254	CG2	-0.076	0.375	-0.320	269	PHE	CD2	-0.277	0.445	-0.257		
238	CE2	-0.309	0.519	-0.164	254	CA	-0.068	0.410	-0.342	269	CE2	-0.300	0.446	-0.277			
238	CZ	-0.312	0.514	-0.136	254	CC	-0.085	0.422	-0.365	269	CG	-0.324	0.450	-0.269			
238	CE1	-0.326	0.528	-0.120	254	N	-0.081	0.419	-0.391	269	CE1	-0.326	0.454	-0.241			
238	CG	-0.339	0.548	-0.133	255	N	-0.104	0.435	-0.357	269	CD1	-0.352	0.454	-0.221			
238	CB	-0.335	0.553	-0.162	255	SER	CG	-0.154	0.472	-0.359	269	CG	-0.275	0.448	-0.228		
238	CB	-0.346	0.575	-0.175	255	CB	-0.127	0.470	-0.363	269	CB	-0.255	0.449	-0.206			
238	CA	-0.334	0.595	-0.160	255	CA	-0.122	0.448	-0.377	269	CA	-0.235	0.467	-0.210			
238	CO	-0.328	0.612	-0.182	255	CO	-0.109	0.452	-0.404	269	CO	-0.217	0.471	-0.182			
238	O	-0.345	0.624	-0.196	255	C	-0.120	0.446	-0.428	269	O	-0.199	0.458	-0.174			
239	N	-0.302	0.612	-0.186	256	N	-0.086	0.462	-0.399	270	N	-0.221	0.450	-0.169			
239	LYS	NZ	-0.217	0.692	-0.169	256	ASN	NDD1	-0.079	0.505	-0.393	270	GLU	OE1	-0.147	0.501	-0.157
239	CE	-0.243	0.693	-0.159	256	NDD2	-0.037	0.516	-0.384	270	CE2	-0.142	0.507	-0.111			
239	CD	-0.254	0.670	-0.158	256	CG	-0.052	0.501	-0.393	270	CB	-0.151	0.511	-0.135			
239	CG	-0.271	0.663	-0.186	256	CB	-0.045	0.480	-0.408	270							

TABLE 3 (cont.)

274	THR	N	-0.228	0.573	0.028	285	CA	-0.350	0.413	0.087	296	GLY	CA	-0.411	0.448	-0.252	
274		OG1	-0.249	0.610	0.035	285	CG	-0.346	0.399	0.061	296	CO	CO	-0.403	0.435	-0.277	
274		CG2	-0.253	0.610	-0.016	285	C	-0.348	0.407	0.037	296	O	O	-0.409	0.442	-0.302	
274		CB	-0.235	0.612	0.012	286	N	-0.342	0.378	0.067	297	N	N	-0.391	0.416	-0.270	
274		CA	-0.215	0.592	0.016	286	ILE	CC1	-0.284	0.347	0.067	297	VAL	CG1	-0.341	0.388	-0.261
274		CO	-0.189	0.596	0.035	286	CG1	-0.308	0.335	0.079	297	CG2	CG2	-0.363	0.361	-0.299	
274		C	-0.173	0.610	0.028	286	CB	-0.334	0.337	0.059	297	CB	CB	-0.367	0.380	-0.278	
275		N	-0.185	0.583	0.057	286	CG2	-0.336	0.323	0.032	297	CA	CA	-0.383	0.400	-0.291	
275	GLY	CA	-0.162	0.584	0.078	286	CA	-0.339	0.361	0.046	297	CO	CO	-0.405	0.394	-0.314	
275		CO	-0.168	0.582	0.109	286	CO	-0.362	0.362	0.021	297	O	O	-0.402	0.394	-0.340	
275		O	-0.150	0.583	0.130	286	C	-0.359	0.368	-0.003	298	N	N	-0.428	0.390	-0.305	
276		N	-0.193	0.579	0.111	287	N	-0.385	0.356	0.029	298	ASP	CD1	-0.457	0.360	-0.351	
276	SER	OG	-0.241	0.578	0.164	287	LEU	CD1	-0.470	0.323	0.021	298	CD2	CD2	-0.520	0.370	-0.321
276		CB	-0.232	0.576	0.137	287	CD2	-0.428	0.310	0.046	298	CG	CG	-0.498	0.370	-0.329	
276		CA	-0.203	0.577	0.139	287	CG	-0.441	0.325	0.022	298	CB	CB	-0.477	0.381	-0.309	
276		CO	-0.191	0.556	0.154	287	CB	-0.432	0.349	0.026	298	CA	CA	-0.452	0.384	-0.323	
276		O	-0.181	0.556	0.179	287	CA	-0.409	0.356	0.010	298	CO	CO	-0.457	0.400	-0.349	
277		N	-0.193	0.538	0.137	287	CO	-0.420	0.377	-0.007	298	C	C	-0.460	0.393	-0.374	
277	TYR	OH	-0.260	0.519	0.244	287	O	-0.424	0.377	-0.033	298	N	N	-0.457	0.422	-0.342	
277		CD2	-0.247	0.510	0.170	288	N	-0.424	0.394	0.010	299	THR	CG1	-0.468	0.461	-0.323	
277		CE2	-0.261	0.515	0.194	288	PRO	CG	-0.429	0.415	0.049	299	CG2	CG2	-0.462	0.482	-0.365
277		CZ	-0.247	0.514	0.221	288	CG	-0.413	0.396	0.040	299	CB	CB	-0.454	0.462	-0.347	
277		CE1	-0.221	0.510	0.226	288	CB	-0.432	0.431	0.024	299	CA	CA	-0.441	0.440	-0.363	
277		CD1	-0.207	0.505	0.202	288	CA	-0.433	0.416	-0.002	299	CG	CG	-0.444	0.435	-0.387	
277		CG	-0.221	0.506	0.174	288	CG	-0.415	0.420	-0.024	299	E	E	-0.453	0.435	-0.412	
277		CB	-0.206	0.500	0.149	288	C	-0.424	0.423	-0.050	300	N	N	-0.419	0.431	-0.377	
277		CA	-0.183	0.516	0.147	289	N	-0.390	0.419	-0.014	300	ILE	CD1	-0.348	0.453	-0.348	
277		CO	-0.163	0.507	0.128	289	VAL	CG1	-0.337	0.441	0.005	300	CG1	CG1	-0.367	0.433	-0.350
277		O	-0.148	0.491	0.136	289	CG2	-0.319	0.419	-0.034	300	CB	CB	-0.371	0.422	-0.380	
278		N	-0.165	0.517	0.103	289	CB	-0.341	0.422	-0.016	300	CG2	CG2	-0.354	0.430	-0.402	
278	GLY	CA	-0.148	0.512	0.081	289	CA	-0.369	0.422	-0.032	300	CA	CA	-0.399	0.426	-0.396	
278		CO	-0.149	0.486	0.076	289	CO	-0.372	0.407	-0.059	300	CO	CO	-0.407	0.406	-0.417	
278		O	-0.170	0.477	0.068	289	C	-0.373	0.415	-0.083	300	O	O	-0.409	0.409	-0.443	
278		N	-0.126	0.476	0.081	289	N	-0.375	0.386	-0.054	301	MET	SD	-0.411	0.387	-0.404	
279	PHE	CD2	-0.061	0.476	0.070	290	SER	CG	-0.374	0.328	-0.073	301	N	N	-0.408	0.319	-0.358
279		CE2	-0.049	0.490	0.051	290	CB	-0.385	0.347	-0.060	301	CG	CG	-0.407	0.346	-0.373	
279		CZ	-0.055	0.488	0.023	290	CO	-0.379	0.368	-0.076	301	CB	CB	-0.428	0.349	-0.398	
279		CE1	-0.073	0.474	0.012	290	CA	-0.403	0.374	-0.098	301	CA	CA	-0.419	0.367	-0.419	
279		CD1	-0.086	0.460	0.030	290	C	-0.402	0.372	-0.124	301	CO	CO	-0.441	0.372	-0.443	
279		CG	-0.079	0.461	0.059	291	N	-0.424	0.380	-0.045	301	O	O	-0.440	0.366	-0.468	
279		CB	-0.094	0.446	0.078	291	HIS	CD2	-0.495	0.376	-0.086	302	N	N	-0.461	0.383	-0.435
279		CA	-0.122	0.452	0.078	291	NE2	-0.498	0.355	-0.036	302	GLU	OE1	-0.537	0.422	-0.400	
279		CO	-0.133	0.439	0.101	291	CE1	-0.486	0.341	-0.052	302	OE2	OE2	-0.544	0.386	-0.396	
279		O	-0.141	0.419	0.097	291	ND1	-0.473	0.353	-0.069	302	CD	CD	-0.538	0.402	-0.408	
280		N	-0.134	0.450	0.125	291	CG	-0.479	0.375	-0.065	302	CG	CG	-0.530	0.401	-0.439	
280	ASP	OD1	-0.095	0.439	0.215	291	CB	-0.468	0.394	-0.082	302	CB	CB	-0.501	0.407	-0.441	
280		OD2	-0.107	0.418	0.180	291	CA	-0.450	0.385	-0.103	302	CA	CA	-0.484	0.389	-0.455	
280		CG	-0.109	0.435	0.192	291	CO	-0.447	0.404	-0.124	302	CO	CO	-0.478	0.399	-0.463	
280		CB	-0.130	0.451	0.178	291	O	-0.449	0.401	-0.151	302	O	O	-0.487	0.391	-0.506	
280		CA	-0.144	0.441	0.150	292	N	-0.442	0.424	-0.112	303	N	N	-0.462	0.416	-0.480	
280		CO	-0.174	0.444	0.147	292	GLU	OE1	-0.387	0.481	-0.067	303	HIS	CD2	-0.420	0.469	-0.538
280		C	-0.184	0.458	0.162	292	OE2	-0.414	0.484	-0.034	303	NE2	NE2	-0.427	0.490	-0.549	
281		N	-0.187	0.432	0.127	292	CO	-0.409	0.479	-0.057	303	CE1	CE1	-0.444	0.498	-0.534	
281	HIS	CD2	-0.264	0.459	0.062	292	CG	-0.430	0.470	-0.082	303	ND1	ND1	-0.448	0.485	-0.513	
281		NE2	-0.270	0.481	0.063	292	CB	-0.418	0.457	-0.106	303	CG	CG	-0.433	0.466	-0.515	
281		CE1	-0.253	0.491	0.082	292	CA	-0.438	0.445	-0.128	303	CB	CB	-0.433	0.446	-0.495	
281		ND1	-0.235	0.476	0.093	292	CC	-0.421	0.441	-0.152	303	CA	CA	-0.453	0.428	-0.505	
281		CG	-0.242	0.455	0.081	292	O	-0.432	0.442	-0.178	303	CO	CO	-0.442	0.410	-0.523	
281		CB	-0.226	0.435	0.089	293	N	-0.396	0.437	-0.144	303	C	C	-0.452	0.407	-0.548	
281		CA	-0.216	0.432	0.120	293	LEU	CC1	-0.325	0.424	-0.176	304	N	N	-0.422	0.399	-0.509
281		CO	-0.228	0.415	0.138	293	CD2	-0.340	0.387	-0.165	304	THR	OG1	-0.373	0.386	-0.487	
281		O	-0.224	0.394	0.134	293	CG	-0.334	0.410	-0.153	304	CG2	CG2	-0.376	0.350	-0.509	
282		N	-0.243	0.423	0.156	293	CB	-0.358	0.419	-0.143	304	CB	CB	-0.391	0.369	-0.498	
282	PRO	CG	-0.262	0.449	0.184	293	CA	-0.377	0.433	-0.163	304	CA	CA	-0.409	0.380	-0.522	
282		CD	-0.251	0.446	0.156	293	CO	-0.392	0.418	-0.186	304	CO	CO	-0.429	0.367	-0.543	
282		CB	-0.272	0.425	0.190	293	C	-0.394	0.423	-0.212	304	O	O	-0.431	0.370	-0.569	
282		CA	-0.257	0.409	0.175	294	N	-0.404	0.401	-0.175	305	N	N	-0.443	0.352	-0.530	
282		CO	-0.274	0.393	0.154	294	TRP	CD1	-0.467	0.342	-0.223	305	VAL	CG1	-0.486	0.304	-0.530
282		O	-0.284	0.399	0.130	294	CG1	-0.490	0.353	-0.222	305	CG2	CG2	-0.509	0.342	-0.531	
283		N	-0.276	0.373	0.165	294	CZ1	-0.512	0.345	-0.241	305	CB	CB	-0.483	0.329	-0.527	
283	ALA	CB	-0.294	0.335	0.170	294	CH	-0.510	0.326	-0.258	305	CA	CA	-0.462	0.338	-0.566	
283		CA	-0.291	0.355	0.150	294	CZ2	-0.487	0.316	-0.260	305	CO	CO	-0.477	0.350	-0.572	
283		CO	-0.319	0.362	0.139	294	CE2	-0.466	0.324	-0.242	305	O	O	-0.483	0.340	-0.595	
283		O	-0.329	0.359	0.114	294	NE	-0.442	0.317	-0.239	306	N	N	-0.481	0.371	-0.568	
284		N	-0.331	0.372	0.159	294	CD2	-0.427	0.330	-0.217	306	ASN	NOD1	-0.534	0.368	-0.568	
284	LYS	NZ	-0.332	0.468	0.180	294	CG	-0.442	0.346	-0.208	306	NOD2	NOD2	-0.549	0.400	-0.551	
284		CE	-0.356	0.455	0.180	294	CB	-0.434	0.364	-0.184	306	CG	CG	-0.533	0.391	-0.563	
284		CO	-0.350	0.430	0.177	294	CA	-0.420	0.385	-0.193	306	CB	CB	-0.511	0.403	-0.576	
284		CG	-0.375	0.416	0.175	294	CO	-0.437	0.395	-0.219	306	CA	CA	-0.494	0.386	-0.590	
284		CB	-0.369	0.392	0.179	294	C	-0.434	0.389	-0.244	306	CO	CO	-0.475	0.400	-0.605	
284		CA	-0.358	0.380	0.154	295	N	-0.455	0.409	-0.212	306	G	G	-0.479	0.420	-0.612	
284		CO	-0.365	0.398	0.130	295	ASP	CD1	-0.534	0.440	-0.211	307	N	N	-0.453	0.389	-0.610
284		O	-0.385	0.409	0.129	295	OD2	-0.515	0.409	-0.207	307						

$F_o < 100$ ) and for the highest resolution reflexions ( $\langle |\Delta\phi| \rangle = 57^\circ$  for  $d < 0.28$  nm). Whether one set of phases is more reliable at high resolution than the other may become evident when a new electron density map is made using the structure factor phase angles.

TABLE 5. SUMMARY OF STRUCTURE FACTOR CALCULATION†

resolution range/nm	R factor	$\langle \Delta\phi \rangle$	$\langle \cos  \Delta\phi  \rangle$
$\infty$ -1.00	1.052	26.8	0.796
1.00-0.60	0.866	36.6	0.689
0.60-0.50	0.635	38.6	0.670
0.50-0.42	0.368	39.0	0.669
0.42-0.35	0.329	42.9	0.623
0.35-0.28	0.362	50.9	0.522
0.28-0.20	0.366	57.1	0.444
all data	0.437	50.2	0.529
structure factor range	R factor	$\langle \Delta\phi \rangle$	$\langle \cos  \Delta\phi  \rangle$
$\infty$ -400	0.294	29.8	0.783
400-300	0.358	34.8	0.724
300-200	0.399	44.1	0.610
200-150	0.420	56.3	0.453
150-100	0.538	64.5	0.348
100-0	1.261	73.4	0.209

†  $R = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}$ , where  $F_o$  is a scaled, observed structure factor,  $F_c$  is a calculated structure factor, and the summation is over all reflexions.

The phase angle differences  $\Delta\phi$  are reduced to the range  $0 \leq \Delta\phi \leq 180^\circ$ .  $\Delta\phi = |\phi_{\text{MIR}} - \phi_{\text{SF}}|$ , where  $\phi_{\text{MIR}}$  is an MIR phase, and  $\phi_{\text{SF}}$  is a structure factor phase.



FIGURE 2. Drawing of the polypeptide chain. The Zn atom is the shaded ball near the centre, the disulphide bond is at the right, the C-terminus is at the left at 307, and the N-terminus is at the bottom at 1.

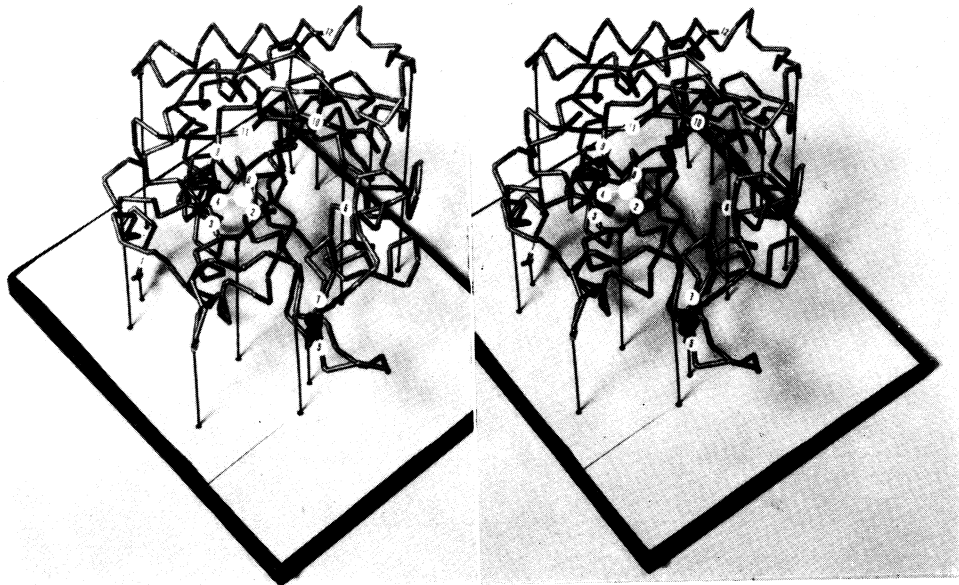


FIGURE 1. For legend see p. 189.

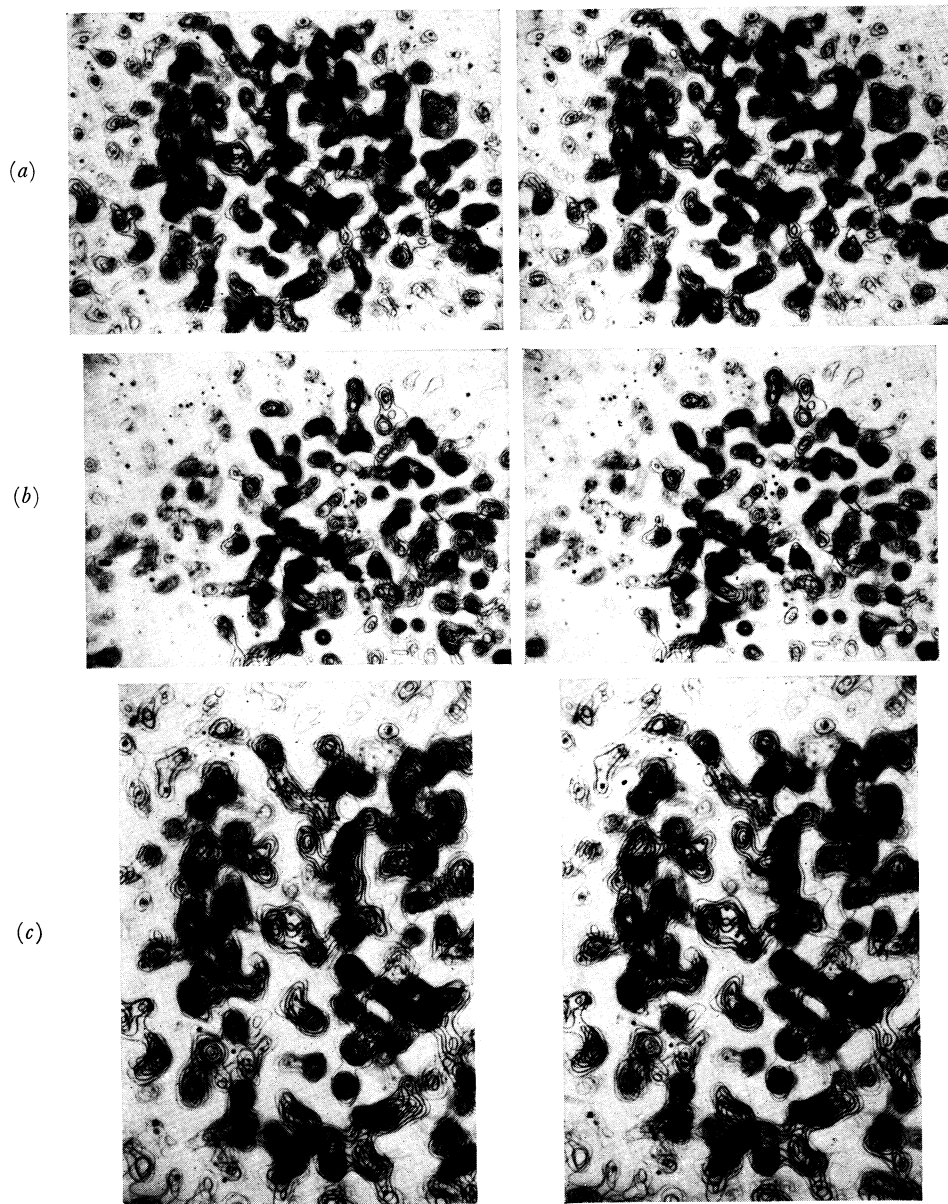


FIGURE 3 *a, b, c*. For legend see p. 189.

(Facing p. 188)

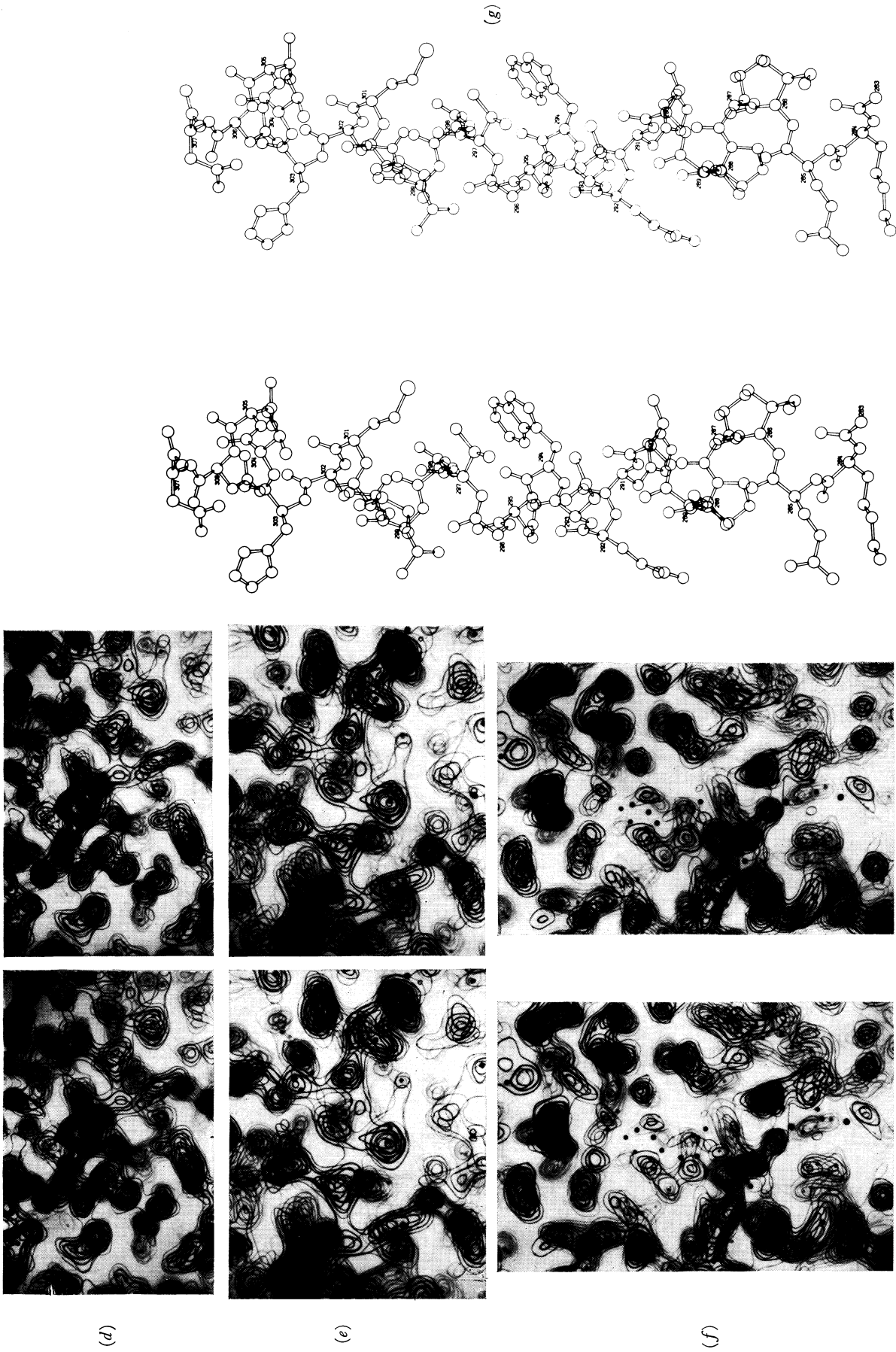
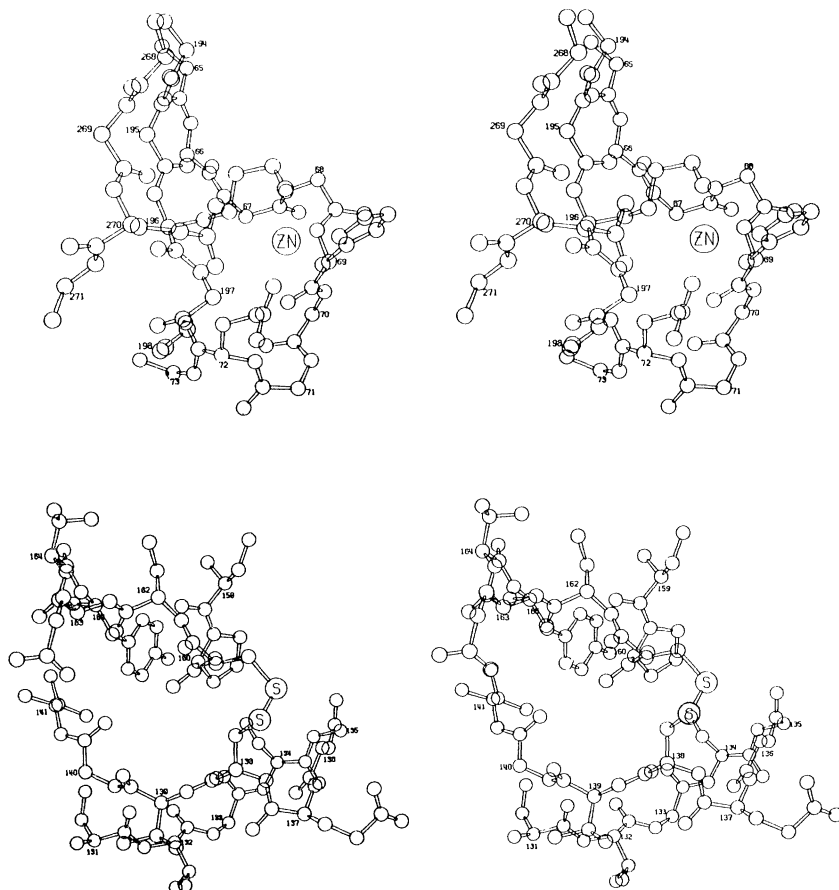


FIGURE 3 *d, e, f, g*. For legend see facing page.

FIGURE 3 *h, i*

## DESCRIPTION OF PLATES 58 AND 59

FIGURE 1. Stereophotograph† of the polypeptide chain of CPA. Each short rod is a peptide unit, and each intersection is one of the 307  $C_{\alpha}$  atoms. The Zn atom is the white sphere and the disulphide bridge is indicated by two black spheres. The numbers are 1, Ala-1; 2, His-69; 3, Arg-71; 4, Glu-72; 5, Cys-138; 6, Arg-145; 7, Cys-161; 8, Glx- or Lys-196; 9, Tyr-198; 10, Tyr-248; 11, Glu-270; 12, Asn-307 (C-terminus).

FIGURE 3 (*a*). Stereocomposite of the electron density sections  $y = 0.38$  to  $y = 0.46$ . From left to right the C-terminal helix, a hydrophobic core, an extended chain, the Zn atom, and the disulphide bond can be seen. (*b*) Stereocomposite of the electron density sections  $y = 0.45$  to  $y = 0.57$ . To the right of the extended chain the pocket region where the substrate binds can be seen. Dots have been added to depict a bound substrate molecule. (*c*) Close-up view of (*a*) which shows the C-terminal helix, the pleated sheet, and the Zn atom. (*d*) The Zn atom and its protein ligands in the electron density map. The water molecule, which completes the near-tetrahedral configuration about Zn, is in sections above those shown. (*e*) The disulphide bond in the electron density map at the right. The bond occurs between S atoms of Cys-138 and Cys-161. The Zn is at the left. (*f*) Close-up view of (*b*) which shows the electron density of the protein in the region where the substrate binds. The substrate molecule is shown as dots which do not match the contours of the protein or solution. (*g*) OR-TEP drawing of (*c*): the C-terminal helix. (*h*) OR-TEP drawing of (*d*): Zn, its protein ligands, and Glu-270. (*i*) OR-TEP drawing of (*e*): the region of the disulphide bond.

† Stereoviewers may be obtained, for example, from Ward's Natural Science Establishment, Inc., Rochester, New York, model 25 W2951. Computer drawn stereo figures were made using the program OR-TEP of Dr Carroll Johnson.

(i) *General features*(b) *Description of the structure*

The gross shape of the CPA molecule is an ellipsoid of approximate dimensions  $5 \times 4.2 \times 3.8$  nm. Views of the entire backbone of CPA are presented in figure 1, plate 58, and figure 2, and smaller portions which contain side chains of particular interest are shown in figure 3, plates 58 and 59. Six of the nine regions of helix are on the left-hand surface of the molecule as viewed in figure 1. A twisted pleated sheet which contains both parallel and antiparallel  $\beta$  structure runs through the centre of the molecule. Between the bank of helices and the pleated sheet there is a core of hydrophobic side chains. A similar hydrophobic trough exists to the right (in figure 1) of the pleated sheet below the level of Zn atom. The Zn is adjacent to the pleated sheet; in fact residue 196 which is one of three Zn ligands is a part of the  $\beta$  structure (figures 3*a*, *d* and *h*). The three ligands from the protein to Zn are His-69, Glu-72 and Lys- or Glx-196. Associated with the Zn and lined on one side by side chains of the pleated sheet are the previously described (Lipscomb *et al.* 1966, 1968) groove and pocket which are essential for substrate binding (figures 3*b* and *f*). With the exception of three short helical regions the half of the molecule to the right of the pleated sheet is random coil. This coil possesses a few hydrogen bonds (approximately 10) and the disulphide bond (Reeke *et al.* 1967) (figures 3*a*, *e* and *i*), but otherwise ought to be quite flexible. Indeed most of the conformational changes observed when a substrate molecule is bound to the enzyme are in this tortuous segment of the molecule.

(ii) *Helical segments*

Table 6 lists the helical segments of the molecule along with their average unit rises, unit rotations and numbers of units per turn. The expected values for an  $\alpha$ -helix are also tabulated (Pauling, Corey & Branson 1951; Pauling 1960). Included in the helix classification are those

TABLE 6. PARAMETERS OF HELICES

residues	unit rise	unit rotation	units per turn
	nm	deg	
14-28	0.143	95	3.8
72-80	0.162	105	3.4
82-88	0.146	107	3.4
94-103	0.153	96	3.8
112-122	0.156	92	3.9
173-187	0.150	94	3.8
215-231	0.159	90	3.9
254-262	0.160	95	3.8
285-306	0.154	97	3.7
$\alpha$ helix	0.149	100	3.6

residues whose conformation is manifestly helical and which participate in at least one hydrogen bond approximately parallel to the helical axis. This criterion necessitates the breaking of the segment 72-88 into two helices because neither hydrogen bond is formed by Trp-81. The region 112-122 is an exception to this definition of a helix since it is so imperfect that only one or two of the potential hydrogen bonds exist. The averages in the table are made over the entire helical segments even though there are imperfect regions near the ends of several segments. Some of the imperfections resemble  $\alpha_{II}$  helix (Nemethy, Phillips, Leach & Scheraga 1967) and others do not fit any of the previously described helical conformations (Bragg, Kendrew & Perutz 1950). The C-terminal helix, residues 285-306 (figures 3*a*, *c* and *g*), is the most nearly perfect helix in the molecule, but even here the helix axis is bent.



(iii)  $\beta$ -structure

The pleated sheet, which is twisted by about  $120^\circ$  from the top to the bottom of the molecule (figure 1), contains in its four parallel and three antiparallel pairs of extended chains about 20% of the backbone atoms. However, only 45 residues, or 17% of the molecule, form at least one of the hydrogen bonds of  $\beta$ -structure (Pauling & Corey 1951). The  $\beta$ -structure is portrayed schematically in figure 4. The dashed lines representing hydrogen bonds have been drawn if the oxygen–nitrogen distance is within 0.06 nm of the expected value of 0.276 nm (Pauling 1960). Under these conditions there are 33 hydrogen bonds among the 45 residues of the pleated sheet.

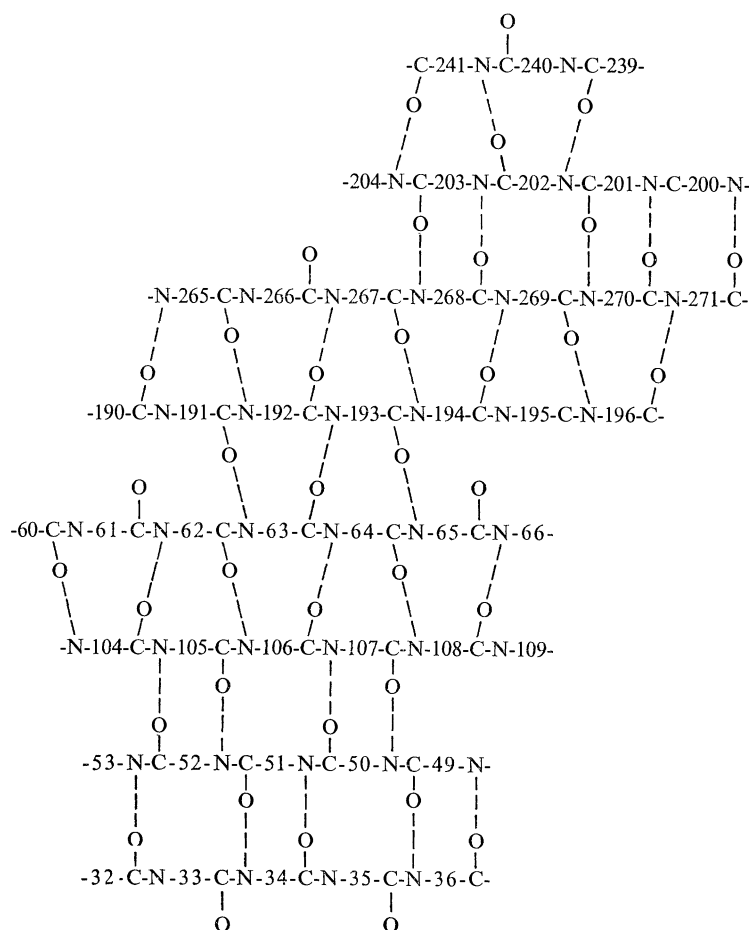


FIGURE 4. Schematic drawing of the pleated sheet ( $\beta$ ) structure of CPA. Hydrogen bonds are shown as dashed lines.

(iv) *Folding of the polypeptide chain*

Carboxypeptidase A is biologically synthesized as the zymogen, proCPA (Anson 1937*a*; Brown, Greenshields, Yamasaki & Neurath 1963). The folding of the CPA chain into its native conformation can be considered independently of the conformation of proCPA, however, because CPA constitutes the C-terminal portion of a proCPA chain, and because proCPA exhibits some substrate binding characteristics of active CPA (Piras & Vallee 1967), suggesting that the CPA portion of proCPA is already nearly in its active conformation in the zymogen.

Examining, then, the folding of CPA from its N-terminus, we find that the various chains of the pleated sheet are not in sequential order (figures 1 and 4). Therefore, the final hydrogen bonds of extended chains 60–66 and 200–204 cannot be formed until chains 104–109 and 265–271 respectively have been folded into place. Furthermore, to insert chain 104–109 into the pleated sheet, residues 1–103 cannot be in their final positions for, if they were, chain 104–109 would have to pass, for example, between helices 14–28 and 72–88, whose axes are 1.2 nm apart, and between Phe-52 and Phe-86, which are 0.35 nm apart in the final structure. We have observed, in addition, three other places in the molecule where the final structure must differ from the conformation during the folding process. First, chain 249–254 must pass between two loops of random coil, which at closest approach in the final structure are only 0.55 nm apart between  $C_{\alpha}$  150 and  $C_{\alpha}$  208. Secondly, residues Leu-233 and Tyr-234 would interfere with placement of the C-terminal helix. Finally, we observe that the disulphide bond must be formed after residues 163–170 pass through the disulphide loop 138–161.

(v) *Ramachandran plot*

A plot (Ramachandran, Ramakrishnan & Sasisekharan 1963) of the dihedral angles of rotation  $\phi$  (about the  $C_{\alpha}$ -N bond) and  $\psi$  (about the  $C_{\text{carbonyl}}-C_{\alpha}$  bond) (Edsall, Flory, Kendrew, Liquori, Nemethy, Ramachandran & Scheraga 1966) for the peptides of CPA is given in figure 5. We have superposed on the usual ( $\phi$ ,  $\psi$ ) plot the boundaries of Ramachandran's 'outer

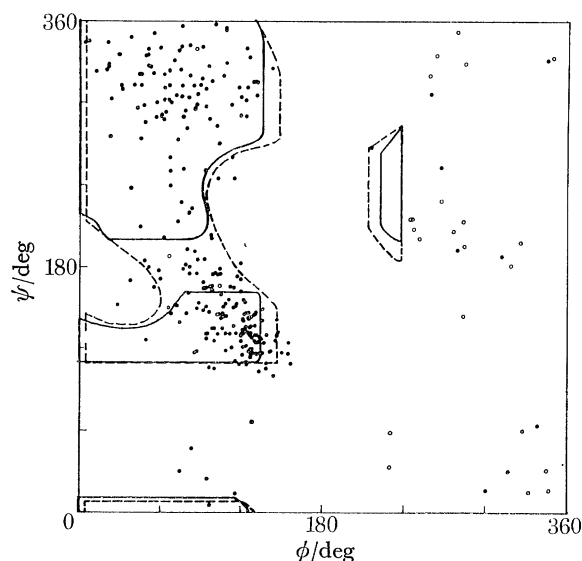


FIGURE 5. Plot of dihedral angles of CPA. Solid lines demarcate allowed regions for  $\tau = 110^\circ$  and dashed lines the regions for  $\tau = 115^\circ$ .

limit' regions allowed for  $\tau = 110^\circ$  (solid lines) and  $\tau = 115^\circ$  (dashed lines) (Ramachandran & Ramakrishnan 1965). Of course, distortions of the peptide unit could change the shapes of the allowed regions. The distribution of points is similar to that found for lysozyme (Blake, Mair, North, Phillips & Sarma 1967*b*), where there are also violations of the  $\tau = 110^\circ$  limits in the regions  $\psi \simeq 180^\circ$ ,  $\phi < 180^\circ$  and  $\psi \simeq 120^\circ$ ,  $\phi \simeq 180^\circ$ .

(vi) *Correlation of sequence and structure*

Table 7, which is similar to the table prepared by Cook (1967) for haemoglobin, myoglobin and lysozyme, is intended to reveal the existence of correlations between structural type and

TABLE 7. CORRELATION OF SEQUENCE AND STRUCTURE

	chemical amino acid analysis	X-ray amino acid analysis	random coil	helix	N helix	C helix	$\beta$
Ala	6.50	5.86	5.67	9.84	3.70	3.70	3.92
Arg	3.58	3.58	6.38	—	—	—	3.92
Asx	9.11	9.45	8.51	9.84	18.52	14.81	3.92
Cys	0.65	0.65	1.42	—	—	—	—
Glx	8.13	9.77	7.80	9.84	14.81	14.81	9.80
Gly	7.32	8.79	14.89	3.28	—	14.81	—
His	2.60	2.93	2.84	4.92	—	7.41	—
Ile	6.50	5.21	3.55	6.56	11.11	—	7.84
Leu	7.48	7.17	2.13	8.20	7.41	3.70	21.57
Lys	4.88	5.86	4.26	6.56	4.71	3.70	9.80
Met	0.98	0.98	—	3.28	—	3.70	—
Phe	5.20	4.89	2.13	6.56	7.41	3.70	9.80
Pro	3.25	3.26	4.26	1.64	7.41	—	1.96
Ser	10.73	10.75	13.48	6.56	7.41	7.41	11.76
Thr	9.11	7.49	7.80	6.56	11.11	11.11	3.92
Trp	2.60	2.61	3.55	1.64	3.70	—	1.96
Tyr	6.18	5.54	9.21	3.28	—	—	3.92
Val	5.20	5.21	2.13	11.48	—	11.11	5.88

The values in columns 1 and 2 are percentages of the total molecule. The values in columns 3 to 7 are percentages of the structural types named. For a given residue, the entries in columns 3 to 7 would equal the entry in column 2 if that residue were distributed randomly among the structural types.

sequence. This tabulation is based on the chemical sequence where available and on the X-ray identifications for the other 93 residues. The chemical amino acid analysis (Bargetzi, Sampath Kumar, Cox, Walsh & Neurath 1963), column 1, and the amino acid totals obtained from the X-ray identifications, column 2, have the largest discrepancies for Glx, Gly, Ile and Thr. In the tabulation of N- and C-ends of helices, columns 5 and 6, three residues are included from the respective ends of each helix. If the distribution of a residue were totally unrelated to secondary structure, a given horizontal row of the table (excepting the first column) would have equal entries. We do not consider the results for Cys, Met, His or Trp to be significant because their percentages of the amino acid composition are small.

The residues which appear to display a preference for one type of secondary structure are: Arg, Gly and Tyr for random coil; Ala and Val for helix; Asx, Glx, Gly and Ile for the ends of helices; and Leu and Phe for  $\beta$  structure. These conclusions are in agreement with those drawn from the haemoglobin-myoglobin-lysozyme tabulation with the exception of the distribution of Asx which Cook found more often in random coil segments. Our result for the  $\beta$

TABLE 8. CONFORMATION OF TYROSINE RESIDUES\*

residue number	$\chi_1$	residue number	$\chi_1$
42	180	166†	85
90	180	169†	49
165	186	208	84
248	200	19	267
259	183	198	98
9	314	204	27
12	295	238	236
48	276		
206	271		
234	315		
277†	285		

\* See also *Note added in proof*, p. 212.

† These residue identifications are by X-ray means only.

structure reflects more the position of the pleated sheet in the core of the molecule than it does the nature of  $\beta$  structure and is not necessarily typical.

It is interesting to note the locations of the 10 proline residues in the CPA structure. Four proline residues, namely 94, 113, 214 and 288, terminate helices at their amino ends. Three other proline residues, 46, 60 and 205, are at the ends of extended chains of the pleated sheet, and the remaining three proline residues, 30, 160 and 282, are situated in random coil.

The conformations of the side chains of the residues are quite varied. For example, table 8 shows the various values of  $\chi_1$  (the angle of the free rotation about the  $C_\alpha-C_\beta$  bond) of tyrosine. The staggered conformations have  $\chi_1$  equal to 60, 180 or 300°.  $\chi_1 = 190$  is found in the tyrosine crystal structure (Smits & Wiebenga 1953). Fourteen of the 18 tyrosine residues identified so far are approximately equally distributed among the three staggered conformations. Four other residues are closer to an eclipsed than a staggered conformation.

A discussion of the locations of charged residues in the CPA molecule must be deferred until the complete chemical sequence is available. For example, the state of amidation of the acidic residues cannot reliably be determined from the crystal structure.

### 3. THE GLYCYL-TYROSINE COMPLEX WITH CPA $_\alpha$

#### (a) Preparation of the difference electron density map

When Gly-Tyr is diffused into crystals of CPA which have been previously crosslinked with glutaraldehyde (Quiocho & Richards 1964) there results an unusually stable enzyme-substrate complex whose crystals are isomorphous with those of the native enzyme (Ludwig *et al.* 1967). Data were collected on this complex to 0.28 nm resolution (Reeke *et al.* 1967) and then to 0.20 nm resolution (Lipscomb *et al.* 1968). The overall agreement factor obtained in the scaling of the various data sets is

$$R = [\sum |F_i^2 - F_j^2|] / [\sum |F_i^2 + F_j^2|] = 0.1037$$

for the 0.20 nm resolution merged Gly-Tyr data set. For further details see table 7 of Lipscomb *et al.* (1968).

Once the structure factor calculation described above was complete, the new phase information could be used to make an improved Gly-Tyr difference map. Two earlier observations had led us to conclude that the Gly-Tyr occupancy in the crystals was low. First, an occupancy of 1/3 was calculated from peak heights in the difference map computed with Fourier coefficients  $(|F_{\text{ES}}| - |F_{\text{N}}|) \exp(i\phi_{\text{MIR}})$  where  $|F_{\text{ES}}|$  and  $|F_{\text{N}}|$  are scaled observed structure factors for the Gly-Tyr enzyme complex and for the native enzyme respectively. Secondly, an attempt to make a map of the enzyme substrate complex using  $|F_{\text{ES}}| \exp(i\phi_{\text{MIR}})$  as coefficients was unsuccessful because the resulting function had the appearance of the native enzyme electron density. Therefore we endeavoured to correct for the low occupancy by calculating a difference map with coefficients

$$(|F_{\text{ES}}| - x|f_c| - (1-x)|F_{\text{N}}|) \exp(i\phi_{\text{SF}}),$$

where  $|F_{\text{ES}}|$  and  $|F_{\text{N}}|$  are as defined above,  $x$  is the fractional occupancy of the substrate, and  $f_c$  is a calculated structure factor for the native enzyme. This function removes the contribution of the unmodified enzyme and enhances substrate features which replace solvent molecules in the complex. The rationale for the function derives from the fact that water is displaced from the active site of CPA when Gly-Tyr is bound. Therefore, for the fraction  $x$  of enzyme molecules

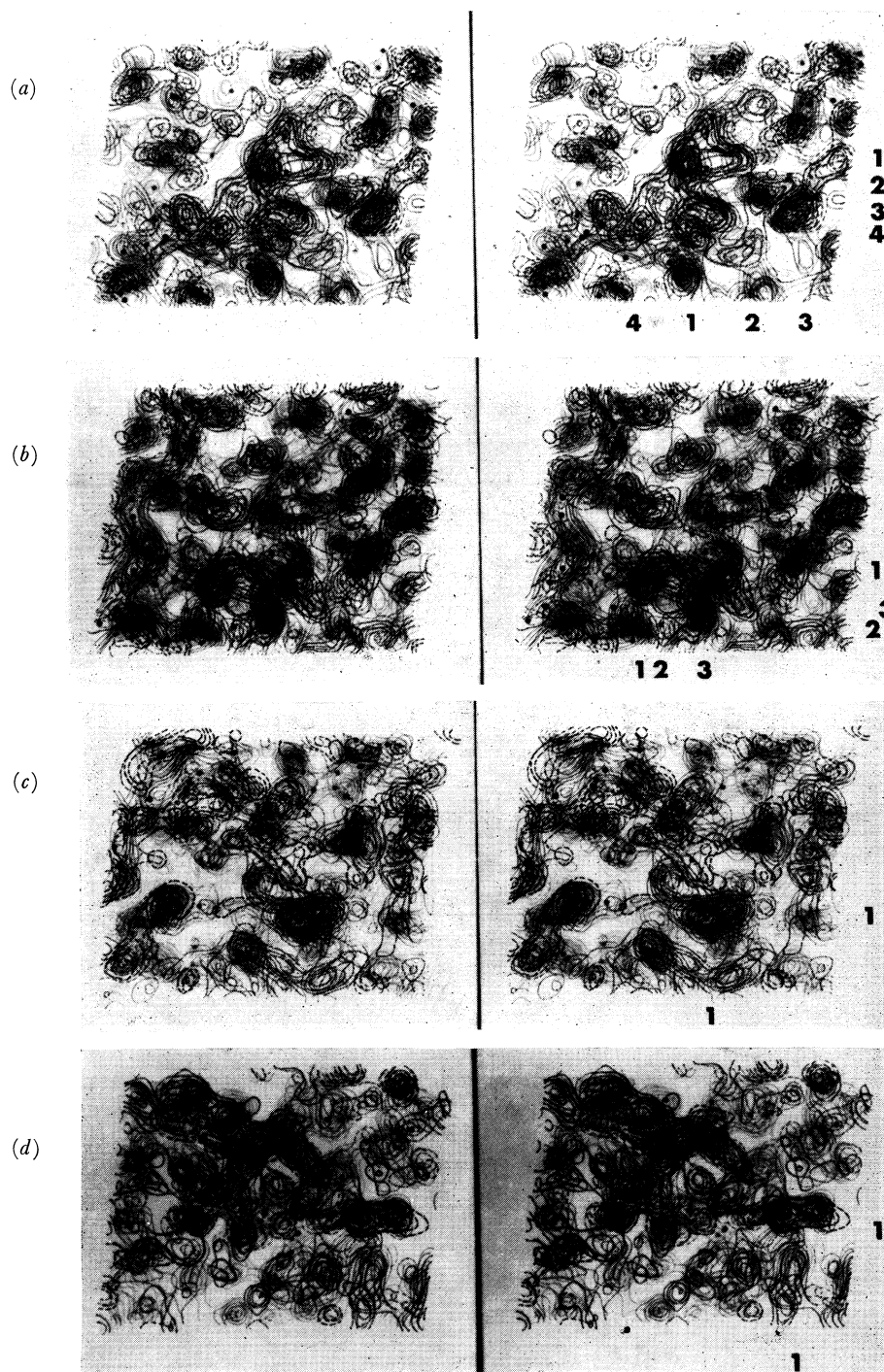


FIGURE 6. The difference electron density function for the complex of Gly-Tyr with CPA is shown as dotted (positive) and dashed (negative) contours. Solid contours show the electron density of CPA at 0.2 nm resolution. Dots are placed at proposed atomic positions. (a) Composite of the difference map sections  $y = 0.47$  to  $y = 0.52$ . Near the top centre the positive contours of the tyrosyl side chain of the substrate are visible (1). To the right of the substrate are the positive (2) and negative (3) contours of the Arg-145 guanidinium group and to the left (4) are the native contours of Glu-270. (b) Composite of the difference map sections  $y = 0.49$  to  $y = 0.56$ . Near the bottom of the picture are the positive contours of the moved Glu-270 (1), the substrate's terminal amino group (3), and the connecting water molecule (square dot, 2). (c) Composite of the difference map sections  $y = 0.56$  to  $y = 0.62$  shows Tyr-248 after its conformational change in the bottom right portion of the picture (1). (d) Composite of the difference map sections  $y = 0.63$  to  $y = 0.68$  shows Tyr-248 before its conformational change (1).

(Facing p. 195)

to which Gly-Tyr is bound,  $|f_e|$ , which does not include any contribution from water, should be subtracted, and for the fraction  $(1-x)$  of enzyme molecules to which Gly-Tyr is not bound,  $|F_N|$ , which does include water, should be subtracted. The parameter  $x$  was decreased in successive maps until at a value of 0.30 water peaks which should be displaced by Van der Waals repulsions with the nearby Gly-Tyr molecule disappeared. The displaced water peaks returned to a small extent when MIR phases were substituted for the calculated structure factor phases in this map. Accordingly, the final difference map of figure 6, plate 60, was made with calculated structure factor phases.

(b) *Description and interpretation of the Gly-Tyr difference electron density map*

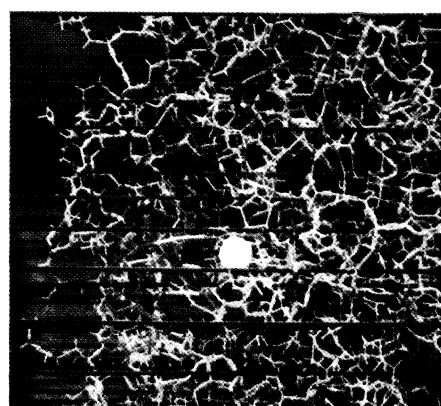
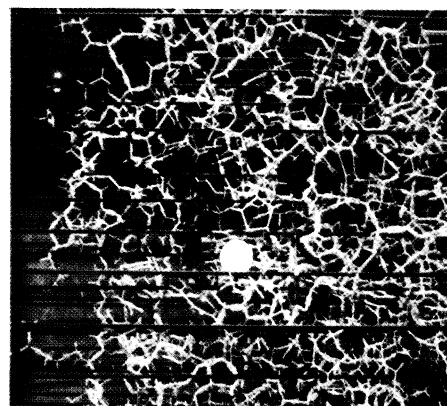
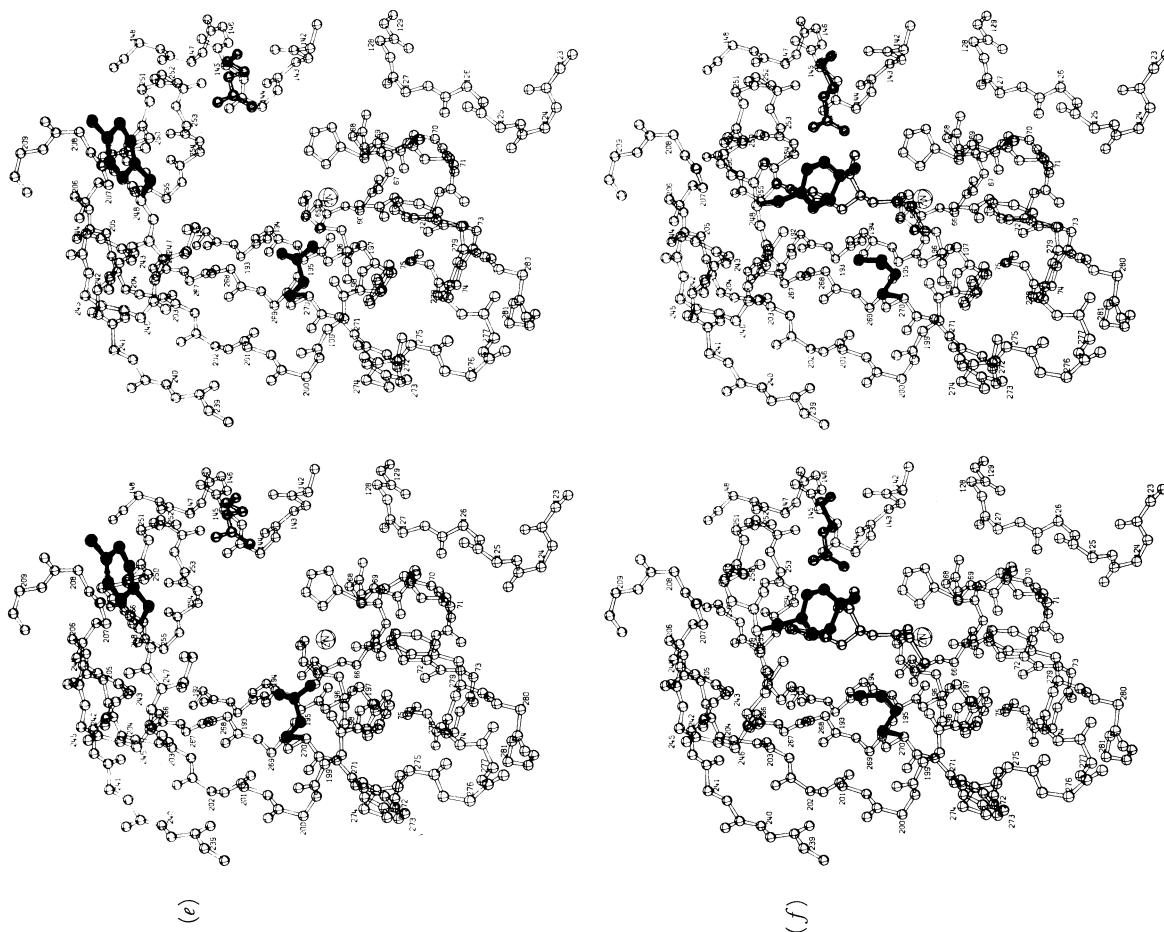
The revised map of the enzyme-substrate complex does not lead to any conclusions different from those previously stated (Lipscomb *et al.* 1968). However, in several instances, for example the conformational change of Tyr-248, features are more distinct than they were previously. The binding of Gly-Tyr to CPA can be summarized by describing four interactions. We believe that the first three of these are characteristic of a productive enzyme-substrate complex. First, the C-terminal side chain of the substrate is inserted into a pocket in the enzyme with concomitant displacement of several water molecules (figure 6*a*). In agreement with the moderate but not high specificity for the C-terminal side chain of the substrate, this pocket contains no specific binding group, probably no charged group, and is large enough to accommodate a tryptophan side chain. Secondly, the terminal carboxylate group of the substrate (which is essential for susceptibility to cleavage) interacts with the positively charged guanidinium group of Arg-145 (figure 6*a*). Thirdly, the carbonyl oxygen of the scissile peptide bond replaces water as a ligand to Zn (figure 6*a*). This feature, although still not certain, is clearer than it was in the previous 0.2 nm difference electron density map. Fourthly, in an interaction possible only with dipeptide substrates, Glu-270<sup>†</sup> binds through water to the  $\alpha$ -amino group of Gly-Tyr<sup>‡</sup> (figure 6*b*). It is probably this interaction which accounts for the unusual stability of the Gly-Tyr-CPA complex. All of these binding interactions are summarized in figure 7.

The native enzyme undergoes several conformational changes when Gly-Tyr is bound. These changes are seen in the difference map as negative density at the native conformation and positive density at the modified conformation. First, the guanidinium group of Arg-145 moves about 0.2 nm by means of a rotation about the  $C_\beta-C_\gamma$  bond of the side chain (figure 6*a*). Secondly, the carboxylate of Glu-270 moves toward the viewer in figure 6*b* by about 0.2 nm. This motion results from rotations about both the  $C_\alpha-C_\beta$  and  $C_\beta-C_\gamma$  bonds of Glu-270. In exception to the above statement, there is no negative density at the native Glu-270 position because this side chain was not included in the structure factor calculation. Thirdly, the phenolic hydroxyl of Tyr-248, identified as being involved in the activity of the enzyme (Roholt & Pressman 1967; Reeke *et al.* 1967), moves about 1.2 nm so that the OH group comes from the surface of the molecule to the vicinity of the peptide bond of the substrate (figures 6*c*, *d*). This

<sup>†</sup> Residue 270 has been identified as glutamic acid from the X-ray map since chemical sequence data are not available for this part of the molecule. Asx and His were also considered, but did not fit the density so well as Glx, nor could they undergo the conformational change observed at 270 when Gly-Tyr binds. The choice of Glu rather than Gln is corroborated by the binding of a Pb ion near this residue (see table 1).

<sup>‡</sup> Yanari & Mitz (1957*a*) deduced from the increased efficiency of Gly-Tyr and Leu-Tyr as competitive inhibitors at higher pH that only the anionic form of a dipeptide, i.e. the species with an uncharged amino group, is bound to CPA. Although we cannot distinguish crystallographically which species of Gly-Tyr is bound, it is reasonable that the  $\text{NH}_3^+$  of the zwitterion would be repelled by the positive charge on Zn.





(a)

(b)

FIGURE 9 *a, b, c, d.* For legend see facing page.





*(b) Binding of active peptide substrates*

The above experimental findings, relating primarily to the substrate structure, provide data with which the X-ray diffraction models of the enzymic binding and catalytic steps must be consistent. We assume that in the productive mode substrates are bound with the C-terminal aromatic residue in the pocket, with the terminal carboxyl group salt linked to Arg-145, with the carbonyl group of the susceptible peptide bond bound to Zn, and with the OH of Tyr-248 near enough to donate a proton to the NH of the susceptible peptide bond (figure 9, plate 61). These points of binding are sufficient to establish the stereospecificity of CPA for C-terminal L amino acids. All of these features are unambiguous in the Gly-Tyr difference map. We further assume that the coordinated motion of Arg-145 and Tyr-248 which occurs when Gly-Tyr is bound also takes place when larger substrates are bound (Steitz *et al.* 1967).

We now extrapolate the results obtained with Gly-Tyr to explain the binding and cleavage of other substrates, based on model building and the chemical properties cited above. If all other groups of the protein are assumed to be in essentially the same position in the active substrate complex as they are in the Gly-Tyr complex, we cannot form a link (through water) between the N-terminus, if it exists, and the carboxyl group of Glu-270 when substrates longer than dipeptides are bound. Nevertheless, Glu-270 will be very close to the carbonyl carbon of the hydrolysable bond. The placement in our structure of a longer chain substrate is determined from the importance of the NH of the second peptide bond, from the general shape of the groove in the enzyme structure itself, and from the detailed results of Abramowitz *et al.* (1967). Maximum interaction of the aromatic R groups in positions 3 and 4 (figure 10) with the aromatic enzyme residues Tyr-198 and His(Phe)-279 results if the second substrate NH is allowed to hydrogen bond to the OH of Tyr-248. In addition, the carbonyl group of the third peptide bond, and possibly also that of the fourth, can be placed near Arg-71 in a stabilized situation.

While we would not claim that all experiments relating to binding are uniquely interpretable on the basis of this model, we know of no results which are incompatible with it. Several examples of readily interpretable observations now follow: (a) The presence of a dead-end pocket, in addition to a groove, provides an explanation of the early observation that CPA is an exopeptidase, not an endopeptidase (Hoffman & Bergmann 1940; Waldschmidt-Leitz 1931). (b) Peptide substrates with a D amino acid at the C-terminus, such as CBZ-Gly-D-Phe, are neither hydrolysed nor bound (Neurath, Elkins & Kaufman 1947). Here we find that the peptide bond, carboxyl group, and R group of a substrate C-terminal D residue cannot easily be accommodated at their correct positions for hydrolysis, and steric interactions which prevent binding may occur. (c) The small  $K_I$  and  $K_m$  (Izumiya & Uchio 1959) of Gly-Tyr as compared to the  $K_I$  of *N*-acetyl-Tyr (Yanari & Mitz 1957*a*) indicates that the former is bound more strongly than the latter. Earlier work comparing *N*-acyl dipeptides with analogous dipeptides also indicated that when a free amino group is present, the dipeptide is bound tightly but is hydrolysed very slowly (Hoffman & Bergmann 1940): we find from models that blocking of this amino group destroys the non-productive binding mode, described above as an interaction of the amino group with the carboxyl group of Glu-270, thus freeing Glu-270 to serve in its probable catalytic role. (d) Model building indicates that steric hindrances occur if the NH group of the sensitive peptide bond is substituted. The most severe example is CBZ-Gly-thiazolidine-4-carboxylic acid, which is neither bound nor cleaved (Smith 1948), probably because of steric interference by Ile-247. In the case of CBZ-Gly-sarcosine and CBZ-Gly-Pro, neither of which is cleaved (Stahmann *et al.* 1946), model building suggests interference with the final stages of

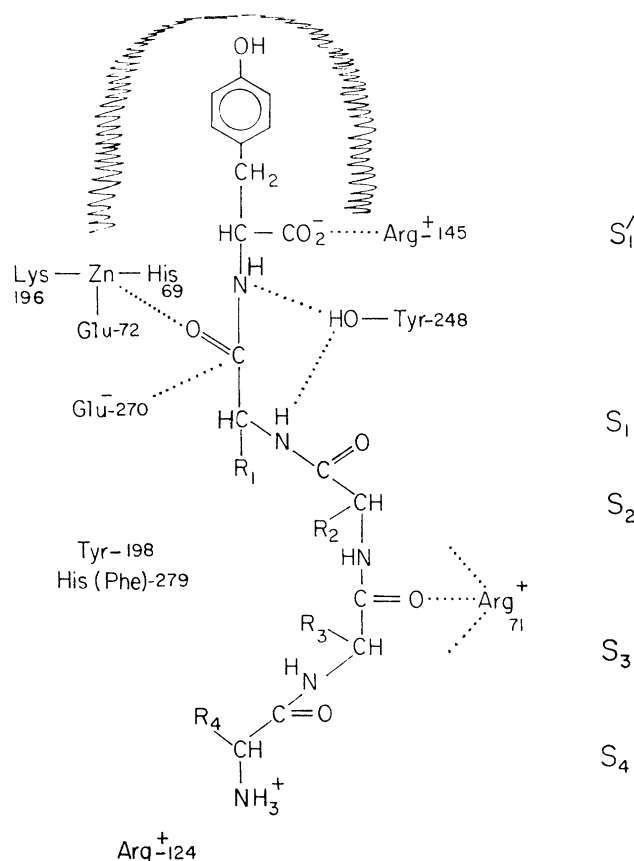


FIGURE 10. Drawing of the binding of a longer substrate to CPA.

the conformation change of Tyr-248. (e) The recent experiments of Abramowitz *et al.* (1967) have mapped carefully several effects, of which those relating primarily to the kinetic constant  $K_m$  are discussed here. Comparison of  $K_m$  for Phe-Ala-Ala with  $K_m$  for Ala-Ala-Ala yields a ratio of 72:1 in favour of this N-terminal aromatic at site S<sub>2</sub> (figure 10), but substitution of an aromatic group is also felt at S<sub>1</sub>, S<sub>3</sub> or S<sub>4</sub>. The effect is strongest when the relatively flexible CBZ group is at S<sub>3</sub>, where it can most favourably be placed near Tyr-198 and His(Phe)-279. Also, a consistently lower  $K_m$  is noted if the terminal NH<sub>3</sub><sup>+</sup> group in a tetrapeptide is blocked by acetylation or by the introduction of a urethane. Our model indicates that this substitution on NH<sub>3</sub><sup>+</sup> provides additional interactions of lone pairs of added oxygen atoms with the guanidinium group of Arg-71 and removes the possible repulsion of the NH<sub>3</sub><sup>+</sup> from Arg-71. (f) Placement of the bulky D-Leu residue as the S<sub>1</sub> peptide in several *N*-acyl-D-Leu-L-Tyr substrates is known to lower the rate of hydrolysis by a factor of about 5000 relative to that of the corresponding *N*-acyl-L-Leu-L-Tyr (Yanari & Mitz 1957*b*): if the amide of the second peptide binds to Tyr-248, the D side chain interferes sterically with the protein, but such repulsion does not occur for the L compounds.

## (i) Zn-carbonyl mechanisms

## (c) Cleavage

The probable mechanisms of cleavage of a peptide substrate proposed below, while not necessarily exhaustive, are consistent with the importance of the amide group of the substrate's penultimate peptide bond, the effect of pH on cleavage, the kinetic experiments on polypeptide

substrates, and the effects of modification of the enzyme on its activity. One important contribution of X-ray diffraction methods is the identification of the various groups of the protein which are close enough to the substrate to function in the catalytic steps. Mechanistic deductions are based on knowledge of these groups, which are Arg-145, Zn, Glu-270, Tyr-248, and H<sub>2</sub>O, and we now discuss the roles of each in turn (figure 11).

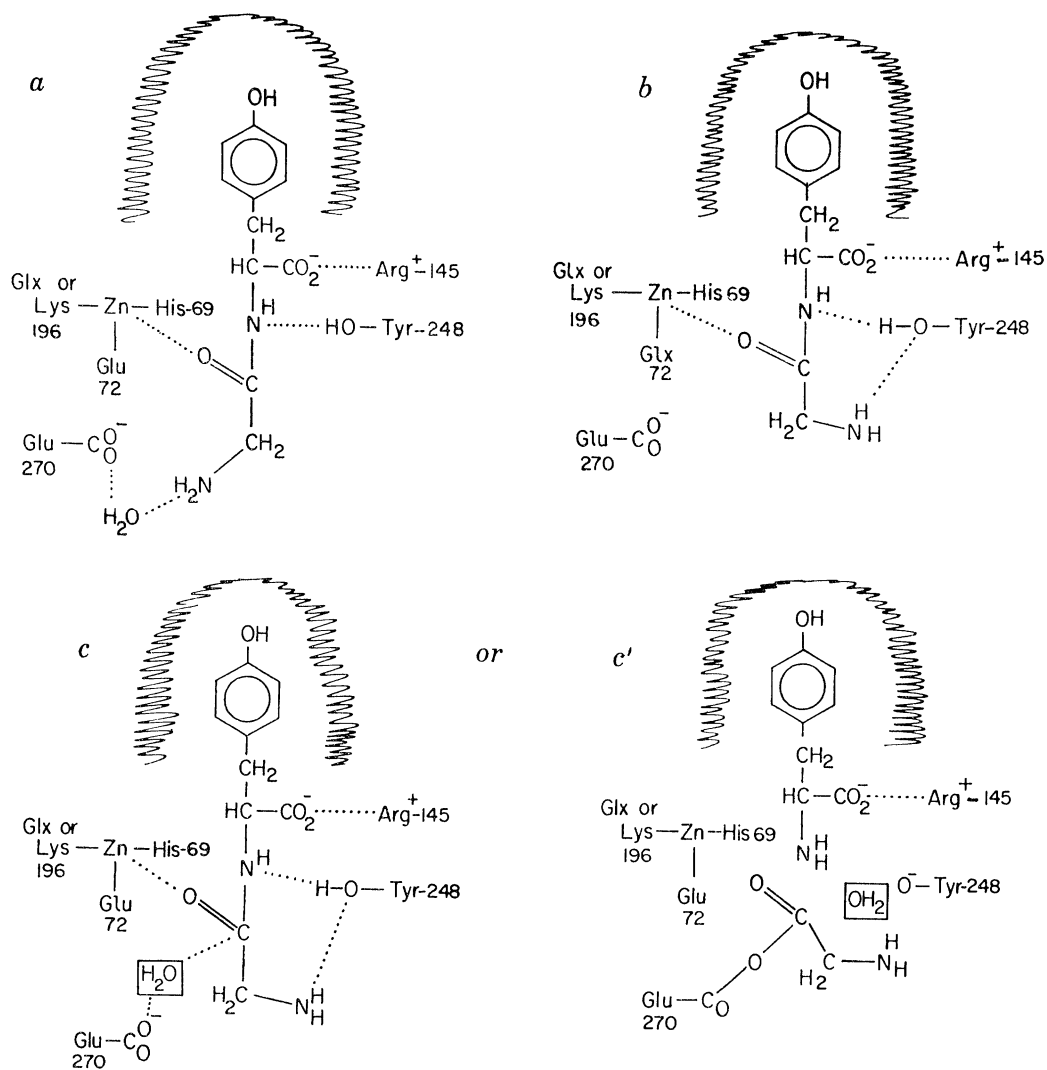


FIGURE 11. Possible stages in the hydrolysis of a peptide by CPA. It is probable that the carbonyl carbon of the substrate becomes tetrahedrally bonded as the reaction proceeds, but it is uncertain at what stage of the reaction the proton is added to the NH group of the susceptible peptide bond. (a) Abortive binding of the dipeptide Gly-Tyr. (b) Productive binding mode. (c) General base attack by water upon the carbonyl carbon of the substrate. (c') Nucleophilic attack by Glu-270 upon the carbonyl carbon of the substrate.

The function of Arg-145 has been shown to be the binding of the terminal carboxyl group of Gly-Tyr. On this basis and from the results at 0.6 nm resolution of binding of longer substrates to acetyl-CPA (Steitz *et al.* 1967), we assign this same function to Arg-145 in the proposed productive mode of binding of longer substrates.

Smith & Hanson (1948) showed that a metal is essential for the activity of CPA, and this metal was later shown by Vallee & Neurath (1954) to be Zn. The Zn atom serves to polarize

the carbonyl group of the substrate,  $Zn^{+} \dots O^{\delta-} - C^{\delta+}$ , in order to render the carbon atom of this carbonyl group more susceptible to nucleophilic attack. Thus the Zn atom functions as a general acid, which attacks the basic oxygen atom and serves to reduce the double bond character of this carbonyl group of the substrate. In addition, bonding about this carbonyl carbon can thereby more easily become non-planar, rendering this atom more susceptible to nucleophilic attack.

Glu-270 probably functions either in nucleophilic attack on the carbon of the susceptible carbonyl group ('Nucleophilic pathway' figure 11*c'*) or in promoting general base catalysis of the oxygen atom of a water molecule at this carbon atom of the susceptible peptide bond ('General base path' figure 11*c*). Chemical evidence relating to these possibilities is discussed below. Another possible role for Glu-270 may be to enhance the polarization of the susceptible carbonyl group in the sense  $Zn^{+} \dots O^{\delta-} - C^{\delta+} \dots Glu^{-}-270$ .

After its conformational change Tyr-248 is in a position to donate a hydrogen bond to the NH of the susceptible peptide bond of the substrate, and, within the limits of error of the Gly-Tyr difference map, to receive a hydrogen bond from the NH group of the penultimate peptide bond. The hydrogen bond to the N of the susceptible peptide bond tends to make the bonding around this N non-planar. At some point in the reaction the H of this hydrogen bond may be incorporated in the  $NH_2$  group of the product, the OH of tyrosine being regenerated perhaps simultaneously by water. A second possibility is that the second proton of  $NH_2$  comes directly from water, but the geometry of the Gly-Tyr complex does not favour this mechanism for peptides. The system of two hydrogen bonds between Tyr-248 and the two NH groups of the substrate would further induce strain in the substrate. Finally, in the case of the 'nucleophilic pathway' mentioned above, if the proton transfer to the susceptible NH occurs before the anhydride is cleaved, then either the Tyr  $O^{-}$  could promote cleavage by water, or incipient  $OH^{-}$  formed in the regeneration of Tyr-248 could attack the anhydride directly in a nearly concerted mechanism. Of course, this incipient  $OH^{-}$  would not be in equilibrium with the aqueous medium.

One of the most striking effects of the binding of the substrate to CPA is the conversion of the enzyme cavity from a water-filled to a hydrophobic region. At least four water molecules must be expelled when a substrate C-terminal side chain such as Tyr is inserted into the pocket, and one water molecule is displaced from the Zn atom when the carbonyl group of the substrate is bound. Also, the charge of Arg-154 is somewhat neutralized by its interaction with the terminal carboxylate ion of the substrate. Finally, Tyr-248, in making its conformational change, closes off the enzyme cavity so that it is not in equilibrium with the solvent. The Zn atom probably has one positive charge after the substrate is bound so that the conversion of the surroundings of Zn to a hydrophobic region enhances the ability of Zn to polarize the susceptible carbonyl group. Thus the positive charge is brought more to the surface of the enzyme-substrate complex. It is hard to escape the conclusion that the displacement of water upon binding the substrate and the resultant conversion of the active centre of the enzyme to a hydrophobic area provide a driving force for the reaction.

In summary, we regard the attack on the susceptible carbonyl group as either nucleophilic catalysis or general base catalysis, and we remain unsure of the mechanism or timing of proton transfer to the susceptible NH group. However, the roles of Zn in polarizing the susceptible carbonyl group and of Arg-145 in binding the terminal carboxyl group of the substrate are clear. Nevertheless, Glu-270 and Tyr-248 definitely appear to be involved in the catalysis even though their exact functions remain to be clarified. We now turn to some of the chemistry

relating to this proposed mechanism and later consider briefly some of the less probable alternative mechanisms.

The pH-rate profile for peptidase activity is usually bell shaped, with inflexions at pH 6.7 and pH 8.5 (Utsunomiya 1942; Neurath & Schwert 1950; Lumry, Smith & Glantz 1951; Riordan & Vallee 1963). Assuming that the pH-activity relation arises from two groups on the enzyme, Carson & Kaiser (1966) derived p*K* values of 6.9 and 7.9 for these two residues from a plot of  $k_{cat}/K_m$  against pH for the hydrolysis of *O*-acetyl-mandelate. If these values of p*K* are assigned to Glu-270 and Tyr-248 respectively, both groups titrate very abnormally. However, factors such as a hydrophobic environment (Blake, Johnson, Mair, North, Phillips & Sarma 1967*a*; Rupley 1967), changes in conformation with pH (Oppenheimer, Labouesse & Hess 1960), the proximity of Tyr-248 to Arg-145, the nature of the rate determining step (Bruice & Schmir 1959) and the nature of the substrate (Izumiya & Uchio 1959) may contribute to the effect of pH on activity.

Unfortunately, we cannot at present choose between the nucleophilic and the general base mechanism of attack on the carbonyl carbon. Steric aspects seem to favour the anhydride pathway. The relative orientation and separation (about 0.25 nm) of the carboxyl group of Glu-270 in its position in native CPA and the hydrolysable bond of the substrate are ideal for nucleophilic attack. On the other hand, the placement of water for general base reaction necessitates a re-orientation of the side chain of Glu-270 from its native conformation (the backbone being fixed by the  $\beta$ -structure) in order to facilitate attack by water and in order not to violate closest acceptable distances of approach between various atoms in the enzyme, the water, and the substrate. This re-orientation and positioning of water is not only possible, but does occur when Gly-Tyr is bound to CPA. The failure to observe transpeptidation (Ginodman, Mal'tsev & Orekhovich 1966; but see also Wood & Roberts 1954) or transesterification (Hall & Kaiser 1967) is not a strong argument against an acyl intermediate, because the anhydride intermediate would be readily susceptible to hydrolysis by water. Moreover, the approach of another C-terminal residue for the transfer reaction would be hindered by Tyr-248. Furthermore, it is known that L-phenylalanine and L- $\beta$ -phenyllactate, products of the respective peptide and ester substrates, are competitive inhibitors (Whitaker, Menger & Bender 1966; Hall & Kaiser 1967). Thus, the binding of the original C-terminal residue would also hinder its ready replacement by a new C-terminal residue.

Although some implications can be drawn, neither deuterium isotope effects nor  $^{18}\text{O}$  exchange in a virtual substrate distinguishes unequivocally the nucleophilic attack by Glu-270 from the attack by  $\text{H}_2\text{O}$ . Deuterium substitution effects seen in the hydrolysis by CPA of CBZ-Gly-L-Phe ( $k_{\text{H}_2\text{O}}/k_{\text{D}_2\text{O}} = 1.22$ ) (Lumry & Smith 1955; Lumry *et al.* 1951) or of hippuryl-L-Phe (1.1) (F. A. Quioco, unpublished results 1968) are small when compared to the effects expected in general base catalysis, which often demonstrates two- to fivefold higher rates in  $\text{H}_2\text{O}$  than in  $\text{D}_2\text{O}$ , as seen in the deacylation step with chymotrypsin (Bender, Clement, Kézdy & Heck 1964) and in similar non-enzymic general base catalysed reactions (Bender, Pollack & Neveu 1962). However, the results observed for CPA, which are fully compatible with nucleophilic attack, may be equally well explained in the case of any general mechanism<sup>†</sup> where the rate determining step does not involve a proton transfer. Furthermore, although catalysis by CPA of the

<sup>†</sup> Contrary to common usage (Bruice & Benkovic 1966, pp. 27–29), the phrase general base mechanism is meant here to include all pathways where water attacks the susceptible carbonyl carbon even though proton transfer is not the rate limiting step, and the mechanism is actually concerted.

incorporation of  $^{18}\text{O}$  from  $\text{H}_2^{18}\text{O}$  into the carboxyl group of a virtual substrate such as acetyl-L-phenylalanine displays a bell-shaped pH-rate profile which is not consistent with a simple ortho-acid mechanism (general base) for this exchange (Ginodman *et al.* 1966), this result is not sufficient to distinguish the peptidase mechanisms in view of the presence of several potential catalytic groups for virtual reaction. However, in cases where a similar virtual reaction occurs in other proteolytic enzymes, the kinetics for these reactions have been shown to be in accord with a double displacement reaction mechanism (Vaslow 1958; Inagami & Sturtevant 1964; Sharon, Grissaro & Neumann 1962; Neumann, Levin, Berger & Katchalski 1959; Grissaro & Sharon 1964; Sun & Tsou 1963). Finally, in studies of model compounds involving the participation of neighbouring carboxyl groups in ester and amide hydrolysis, several examples of nucleophilic catalysis have been observed (Bender, Choupek & Neveu 1958; Bruice & Benkovic 1966, pp. 173–186; Fersht & Kirby 1968 *a-c*). However, even after a proton is added to the NH group, the resulting amide is a poor leaving group, and a carboxyl group, such as that of Glu-270, is a poor nucleophile. Ordinarily, such a situation would not be a good case for nucleophilic attack (St Pierre & Jencks 1968; Jencks 1969), but we cannot be sure that these arguments are valid in the case of an enzyme where the geometrical disposition of the catalytic groups and the polarization in the enzyme substrate complex are so important.†

The proposed mechanism is also consistent with the kinetic data on longer peptide substrates and suggests why certain of these substrates show maximal hydrolytic rates. Binding of a model polypeptide substrate to the enzyme is most reasonably accomplished by placing it so that its C-terminal side chain is in the pocket, so that it extends along the groove near aromatics Tyr-198 and His(Phe)-279 and so that the carbonyl of the third peptide bond is directed toward Arg-71. With the substrate in this position, a hydrogen bond can be formed between the NH of the penultimate peptide bond and the OH of Tyr-248 (figure 10). There is no other residue of the protein which can form a hydrogen bond to this second NH group. When the amide of the second peptide is substituted or displaced, e.g. when  $\text{R}_1$  is sarcosine or  $\beta$ -alanine respectively, the rates of cleavage are 800 to 2400 times slower than the rates for the analogous glycine compounds (Snoke & Neurath 1949; Hanson & Smith 1948). On the basis of these observations, it was proposed that the penultimate NH is either a hydrogen bond acceptor or donor (Snoke & Neurath 1949). Similarly, the rates of cleavage of acetylated amino acids and acetylated dipeptides differ by a factor of about 1000 (Bergmann & Fruton 1942). This effect is not due primarily to variations of  $K_m$ , because nearly analogous acetylated amino acids and dipeptides have comparable  $K_m$  values within a factor of about 5 (Snoke & Neurath 1949). These data suggest that in the usual substrate an H bond from Tyr-248 to the penultimate amide might produce distortion at the carbonyl carbon of the susceptible peptide. Three types of distortion are possible: (*a*) the N atom of the peptide bond can tend toward a tetrahedral configuration; (*b*) the peptide bond can rotate to become non-planar; and (*c*) the carbon of the peptide bond can become somewhat tetrahedral as the intermediate is formed. When  $\beta$ -alanine is the penultimate amino acid, our model building experiments indicate that its NH group can still be made to form a hydrogen bond to Tyr-248, but the strain in the  $\beta$ -alanyl residue can be taken up by

† An experiment to detect introduction of an  $^{18}\text{O}$  terminal atom into Glu-270 of the protein or into the newly formed carboxyl group of the product under single turnover conditions upon hydrolysis of substrates by CPA in  $\text{H}_2^{18}\text{O}$  might distinguish a nucleophilic reaction from a general base mechanism, unless cleavage of the anhydride were to occur preferentially at the Zn polarized carbonyl carbon. The  $^{18}\text{O}$ -labelling experiments by Kowalsky & Boyer (1960) and by Ginodman *et al.* (1966) do not bear upon this distinction, because at the small enzyme to substrate ratios employed,  $^{18}\text{O}$  would necessarily be incorporated in most product molecules.

the methylene carbon. The importance of adjacent peptides has also been demonstrated for chymotrypsin and papain (Kaufman & Neurath 1949; Bender & Turnquest 1955; Schechter & Berger 1967), and the role of strain has been discussed often (Jencks 1966; Lumry 1959).

Another structural alteration in the substrate which produces a large decrease in  $k_{\text{cat}}$  with little change in  $K_m$  is the substitution of a D residue in the third position of a peptide having at least four residues. For example Ala-L-Ala-Ala-Ala and Ala-D-Ala-Ala-Ala have a  $K_m$  ratio of 1.1:1 but a  $k_{\text{cat}}$  ratio of 50:1 (Abramowitz *et al.* 1967), and we find from our model that if the carbonyl group of the third peptide is directed toward Arg-71, then the methyl group of D-Ala ( $R_2$  in figure 10) interferes sterically with Tyr-248 after the conformational change has taken place. The same type of interference can also occur with the substrate CBZ-sarcosyl-L-Phe, which is hydrolysed very slowly (Snoke & Neurath 1949).

(ii) *Other mechanisms*

A 'Zn-hydroxyl' mechanism (Lipscomb *et al.* 1968) now seems substantially less probable than it did earlier, in view of the absence of density in the region that would be occupied by a substrate carbonyl oriented away from Zn, and the greater density in the region of the Zn-(carbonyl) oxygen bond in the new difference electron density map of the complex of Gly-Tyr with CPA. In the 'Zn-hydroxyl' mechanism, the initial binding step has the susceptible CO group oriented with its O atom away from the Zn, which has  $\text{OH}^-$  or  $\text{H}_2\text{O}$ , depending on the pH, as a fourth ligand. Possible further steps of the reaction, and the steric problems associated with this mechanism have been discussed (Lipscomb *et al.* 1968), but the characteristics of this proposal are as follows: (a) The mechanism is consistent with the lack of a strong effect of Zn removal on the binding of peptides (Coleman & Vallee 1962). In the case of ester substrates, metal substitutions have less effect on catalysis than in the case of peptides (Vallee, Riordan & Coleman 1963); this 'Zn-hydroxyl' mechanism may therefore apply to the hydrolysis of certain esters (Lipscomb *et al.* 1968). (b) The p*K* values of 6.9 and 7.2, characteristic of the free enzyme and of the enzyme-substrate complex, and obtained respectively from the ascending limbs of the  $K_{\text{cat}}/K_m$  against pH and  $k_{\text{cat}}$  against pH curves for the hydrolysis of acetyl mandelate (Carson & Kaiser 1966), could be due (as one possibility) to the ionization  $\text{Zn}(\text{OH}_2)^+ = \text{ZnOH} + \text{H}^+$  (Verpoorte, Mehta & Edsall 1967). (c) The low activity or lack of activity of CPA toward substrates in which the H atom of the susceptible peptide NH is replaced by a larger group (Stahmann *et al.* 1946; Smith 1948) is explained in the case of this mechanism by a resulting steric repulsion of this group by the Zn or  $\text{OH}^-$  on the Zn. However, CBZ-Trp-Pro is cleaved by CPA (Smith 1948) in a reaction which could not proceed by this mechanism. (d) The Zn would be neutral, being bonded to Glu-72 and an  $\text{OH}^-$ , as well as to His and Lys (or Glx). (e) This mechanism is similar to a class of mechanisms now under active consideration for carbonic anhydrase (Riepe & Wang 1968; Smith 1949*a*; Davis 1958; Coleman 1967; Whitney, Nyman & Malström 1967). A determination of whether or not water or hydroxide is displaced from the metal upon substrate binding would be helpful in testing this proposed mechanism. Already nuclear magnetic resonance techniques have demonstrated displacement of water or hydroxide from the metal when the inhibitor  $\beta$ -phenylpropionate is bound to MnCPA (Shulman, Navon, Wyluda, Douglass & Yamane 1966).

Proposals have been made earlier for the mechanism of action of CPA. After the necessity of the presence of a metal for CPA activity was demonstrated (Hanson & Smith 1948), Smith (1949*b*) and Lumry & Smith (1955) suggested initial binding of both the free carboxyl group



and the carbonyl group of the scissile bond to the metal. However, we find that the binding site of the substrate's carboxyl group is Arg-145 in the Gly-Tyr complex, and hence we do not believe that the suggested binding of carboxyl to Zn is present in peptide hydrolysis.

Use of one of the Zn ligands, then thought to be Cys, as a nucleophile (Williams 1964) requires enormous motion of the protein backbone in the case of His-69 or Glu-72, or gross distortion of the peptide or anhydride intermediate in the case of Lys- or Glx-196. While we cannot rigorously rule out the involvement of Lys- or Glx-196 in later steps, there is no evidence from our map of the Gly-Tyr complex that such motion occurs in the binding stage. Other possible mechanisms which might involve groups some distance away from the Gly-Tyr location have been considered, but are unlikely (Lipscomb *et al.* 1968).

The extremely general proposal (Vallee, Riordan & Coleman 1963; Vallee 1964) that a proton donor and a nucleophile are required for peptide hydrolysis are features which are to be expected in enzymatic hydrolysis of peptides. In this proposal, the Zn had four bonds, two to the protein (Cys and N-terminus) and two to the substrate (the N atom of the NH and the O atom of the CO, both of the susceptible peptide bond). The suggestion that an NH can be bound to the enzyme seems first to have been made by Balls & Köhler (1930-31). Since modification of CPA by acetylation, iodination, or photo-oxidation decreased the peptidase activity, the suggestion was made that in these reactions the nucleophile was modified (Vallee *et al.* 1963; Vallee 1964). The nature of these modification reactions appeared to implicate either tyrosine or histidine. However, there is only one tyrosine near the susceptible peptide bond of bound Gly-Tyr, and we identify that critical tyrosine as residue 248. We have discussed above various roles for Tyr-248, but in no case do we believe it to be the nucleophile. Concerning a possible role for histidine in the catalysis, Sokolovsky & Vallee (1967) have suggested that His plays a role in either catalysis or binding of peptide substrates, or both. However, the X-ray sequence of the active site region shows that apart from the zinc ligand, His-69, the only other nearby histidine residue is His(Phe)-279, which is far removed (about 0.7 nm) from the glycyl-tyrosine binding site (figure 10). We consider it unlikely that either of these residues is directly involved in the activity, but of course modification of the Zn ligand, His-69, is likely to produce alterations of the enzymatic activity.

### (iii) Active group modifications

Chemical modifications of the two tyrosine residues which produce changes in the activity of CPA have formed an elegant series of studies from Vallee's group. Study of acetylated CPA indicates the modification of two tyrosine residues which are protected when acetylation is carried out in the presence of the inhibitor  $\beta$ -phenylpropionate (Simpson, Riordan & Vallee 1963). One of these tyrosines can be selectively nitrated by tetranitromethane (Riordan, Sokolovsky & Vallee 1967) and the other can be modified by a small excess of 5-diazonium-1*H*-tetrazole (Sokolovsky & Vallee 1967). In agreement with these data, we find only two tyrosine residues, 248 and 198, in the general area of the active site. Crystallographic studies at 0.6 nm resolution have shown that  $\beta$ -(*p*-iodo)-phenylpropionate binds to CPA at three major sites (Steitz *et al.* 1967). Now that the structures of the native enzyme and of the Gly-Tyr complex are known it seems likely that two phenylpropionate molecules are required to protect both tyrosines.† Binding of a phenylpropionate molecule in the pocket induces the large

† We therefore now believe, contrary to our previous suggestion (Steitz *et al.* 1967), that  $\beta$ -iodo-phenylpropionate can simultaneously occupy both sites.

conformational change of Tyr-248 and accounts for its protection. The second phenylpropionate binding site is near enough to Tyr-198 to account directly for its protection. Nitration destroys peptidase activity while mild diazotization does not, and we therefore suggest in accord with the roles assigned to Tyr-248 in our mechanism that it is Tyr-248 which is nitrated and Tyr-198 which is diazotized. Retention of peptidase activity and an increase in esterase activity occur when CPA is acetylated in the presence of Gly-L-Phe (F. A. Quioco 1968, unpublished results). Since there is only one dipeptide binding site, only Tyr-248 can be protected. These results confirm our conclusion that of the two tyrosines only Tyr-248 is functional in the catalytic step for peptides. In these experiments, however, substrate inhibition of esterase activity is alleviated, suggesting that Tyr-198, although too far from the bound substrate to be directly involved in the reaction, is involved in the mechanism of substrate inhibition. These results are also consistent with the effects of mild diazotization with 5-diazonium-1*H*-tetrazole.

Subsequent to our suggestion that Arg-145 and Glu-270 are involved in the activity of CPA, Vallee & Riordan (1968) succeeded in producing modification of enzymic activity by reacting arginine groups in CPA with diacetyl and carboxyl groups with cyclohexylmorpholinoethyl carbodiimide.

(iv) *Inhibition and activation*

Carboxypeptidase A is known to exhibit a variety of kinetic anomalies, such as inhibition and activation by certain substrates. Substrate inhibition has been demonstrated both for ester substrates, e.g. hippuryl-L- $\beta$ -phenyllactate (HPLA) (Snoke & Neurath 1949; McClure, Neurath & Walsh 1964) and hippuryl-L-mandelate (Awazu, Carson, Hall & Kaiser 1967) and for peptide substrates, e.g. CBZ-Gly-Phe (Lumry *et al.* 1951). However, neither hippurylphenylalanine (Whitaker *et al.* 1966) nor acetylated aromatic  $\alpha$ -amino or  $\alpha$ -hydroxy acids (Snoke & Neurath 1949) exhibit substrate inhibition. Substrate activation is exhibited by CBZ-Gly-Phe (Whitaker *et al.* 1966) and by hippuryl-glycolate (Kaiser, Awazu & Carson 1964). In addition, amino acid products inhibit (Utsunomiya 1942) and acylamino acid products, e.g. CBZ-Gly (Whitaker *et al.* 1966), activate the enzyme. It is noteworthy that these kinetic anomalies have been associated almost exclusively with dipeptide or analogous ester substrates or products which have aromatic *N*-acyl groups.

Model building experiments have so far elicited two abortive modes of binding for substrates which may result in substrate inhibition. In both of these cases the productive binding site which we have deduced from the Gly-L-Tyr crystallographic experiment is occupied and substrate-protein interaction is maximized. Furthermore, in generating these models we have assumed that Tyr-198 is associated with substrate inhibition (see above), and we have been aware that only *N*-acyl dipeptides or their ester analogues with both the *N*-acyl and C-terminal R groups aromatic have so far been shown to exhibit substrate inhibition. The first abortive binding mode has the substrate (CBZ-Gly-L-Phe in figure 12*a*) shifted with respect to the productive mode, so that the C-terminal carboxyl group is bound to the Zn atom, the C-terminal side chain partly extends into the hydrophobic pocket, the aromatic acyl group has a favourable  $\pi$ - $\pi$  interaction with Tyr-198, and the second peptide carbonyl is bound to Arg-71. The second mode of binding (figure 12*b*) has the substrate reversed so that the aromatic acyl group is in the pocket which accommodates the C-terminal side chain in productive binding. The terminal carboxyl group of the substrate can then bind to Arg-71, and the C-terminal side chain is near Tyr-198. So far we have been unsuccessful in devising a crystallographic experiment to

demonstrate either of these abortive binding modes. The shifted substrate position does not allow as much protein-substrate interaction as does the reversed one, and should be less specific since it admits binding of inhibitors of any length. In both abortive binding modes it is important that the *N*-acyl group or the C-terminal residue respectively be aromatic or large aliphatic in

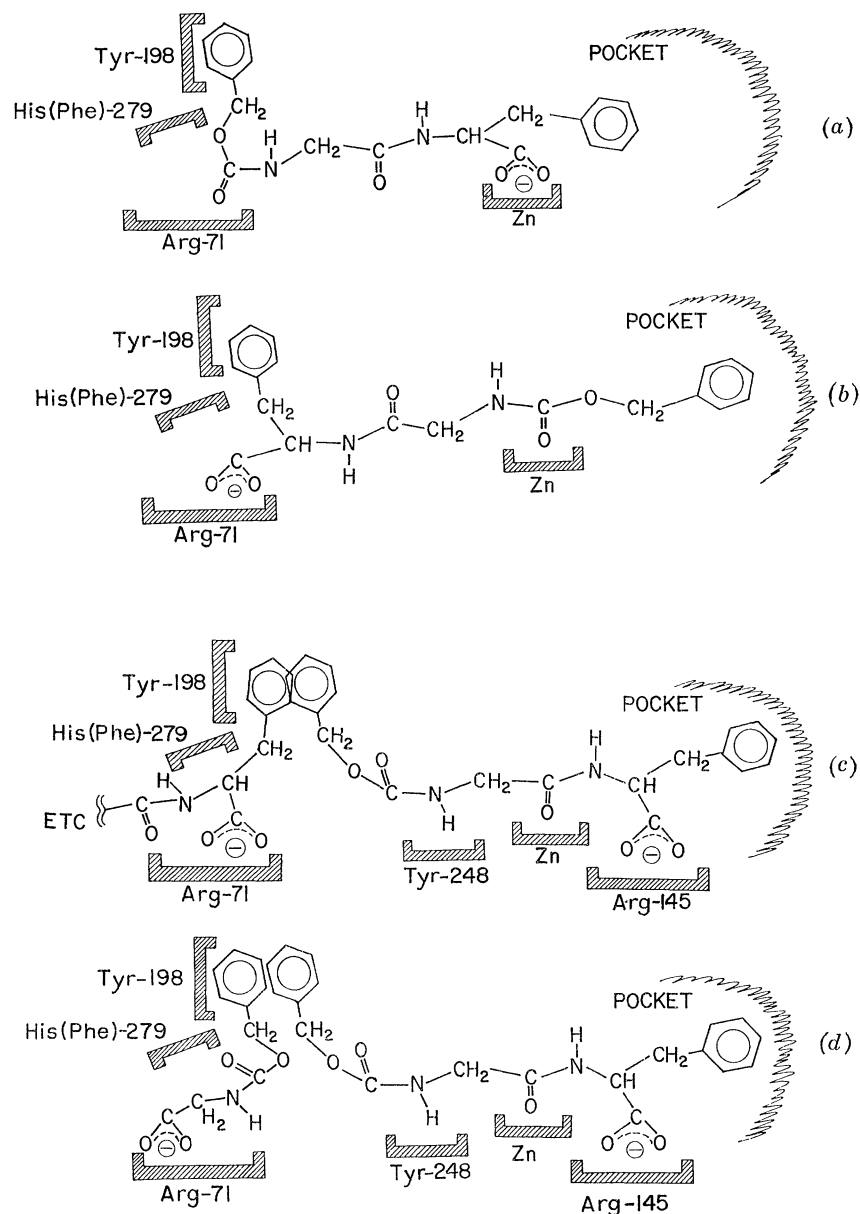


FIGURE 12. Proposed modes of binding of substrate molecules showing: (a) displaced binding, (b) reversed binding for CBZ-glycylphenylalanine, (c) binding of two substrate molecules giving rise to inhibition which would not be competitive, and (d) binding of a substrate and of a product molecule in an activation process.

order to have a favourable hydrophobic interaction with Tyr-198. In the case of short substrates, such as acyl amino acids and similar esters, which do not show substrate inhibition, there would be less favourable binding in either of these two abortive modes. For example, in shifted binding the *N*-acyl group would not be near Tyr-198, and in reversed binding the *N*-acyl group would no longer extend into the pocket. Of course the two binding modes discussed here do not

preclude other possibilities, and indeed yet another mode must be involved in order to satisfy the kinetic data (Snoko & Neurath 1949; Lumry *et al.* 1951; Whitaker *et al.* 1966; McClure *et al.* 1964; Quioco & Richards 1966), which imply binding of more than one substrate molecule in order to inhibit one active centre (Thoma & Koshland 1960). One way to bind two substrate molecules is suggested in figure 12*c*, which shows a second molecule bound farther down the groove in a way that could interfere with precise placement of the first molecule in the active site for hydrolysis. In disagreement with kinetic analyses which suggest that as many as 3 to 5 molecules of CBZ-Gly-Phe (Lumry *et al.* 1951) or HPLA (Dennard & Williams 1960) are bound, we find that the binding region is probably limited to two substrate molecules of this size.

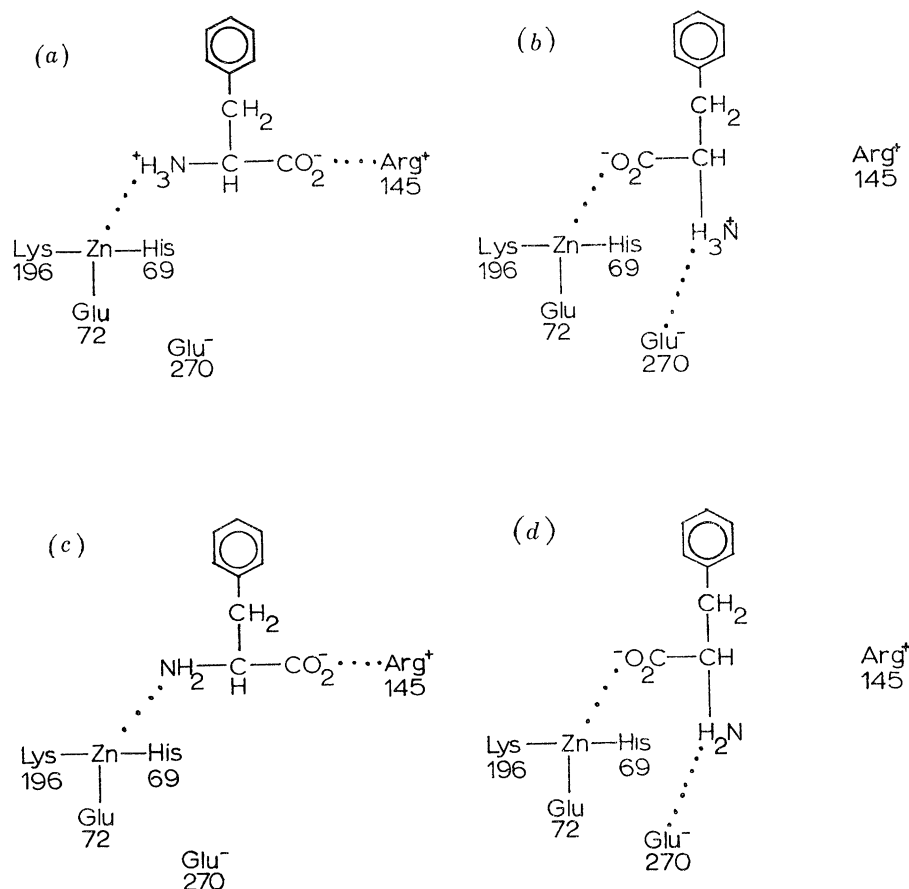


FIGURE 13. Probable binding of (a) L-Phe at pH 7.5, (b) D-Phe at pH 7.5, (c) L-Phe at pH 9.0 and (d) D-Phe at pH 9.0. Inhibition (a) is one-eighth that of (b), and dependent upon phosphate buffer concentration, but at pH 9 L-Phe is about as effective as D-Phe. The binding at (a) with the  $\text{CO}_2^-$  on Arg-145 and the  $\text{H}_3\text{N}^+$ , toward  $(\text{ZnL}_3)^+$ , where L is a Zn ligand, is suggested by the occurrence of the Tyr-248 conformational change in an X-ray study at 0.6 nm resolution of the binding of L-Phe at pH 7.8.

The cluster of side chains composed of Arg-71, Tyr-198, and His(Phe)-279 which is located farther down the cleft may also be used to explain substrate and product activation. If CBZ-Gly, which does activate the enzyme (Whitaker *et al.* 1966) is placed so that the free carboxyl group is bound to Arg-71, then the phenyl ring can interact favourably with Tyr-198 and with the phenyl ring of a productively bound substrate molecule (see figure 12*d*). Similar interactions occur when a second molecule of hippurylglycolate or CBZ-Gly-Phe is bound. Substrates long enough to cover the Arg-71–Tyr-198 region are then not expected to exhibit any of the above

anomalous binding modes. In fact, the data which are presently available indicate that substrates longer than acyl dipeptides are not subject to substantial substrate inhibition or to product activation (Abramowitz *et al.* 1967; Auld 1968).

Competitive inhibition by products such as L-phenylalanine has been found in the CPA-catalysed hydrolysis of CBZ-Gly-L-Phe at pH 9.0 (Neurath & DeMaria 1950). This inhibition is primarily due to the anionic form of L-phenylalanine. Recently, however, it has been shown that L-Phe is a competitive inhibitor of CPA even at pH 7.5 (Whitaker *et al.* 1966). Moreover, L-Phe has been shown crystallographically to bind only in the pocket, and this binding is accompanied by a structural change presumably involving Tyr-248 (Steitz *et al.* 1967). The detailed structural interpretation from model building of the binding of L-Phe places the  $\alpha$ -COO<sup>-</sup> near Arg-145. D-phenylalanine, on the other hand, inhibits competitively and more effectively at pH 7.5 than at pH 9.0 (Neurath & DeMaria 1950; Elkins-Kaufman & Neurath 1949). This finding has been interpreted by assuming that the charged  $\alpha$ -amino group of D-Phe, unlike that of L-Phe, is oriented near a negative group of the protein. Accordingly, our structural interpretation of the binding of D-Phe, based upon model building only, places the  $\alpha$ -COO<sup>-</sup> near Zn (Dennard & Williams 1960), and the charged  $\alpha$ -amino group near Glu-270 (see figure 13). The loss of a proton from the  $\alpha$ -NH<sub>3</sub><sup>+</sup> at pH 9.0 would then result in less effective binding.

The strictly competitive inhibition by  $\beta$ -phenylpropionate in the region of equimolecular inhibitor:CPA binding would be analogous to that depicted for L-Phe. However, mixed inhibition, competitive and non-competitive (or uncompetitive), by  $\beta$ -phenylpropionate has also been found (Lumry & Smith 1955) and is probably related to the ability of this inhibitor to bind in two meaningful sites, as demonstrated by X-ray crystallography.

#### (v) *Esterase activity and metals*

It has long been known that CPA possesses esterase as well as peptidase activity (Snoke & Neurath 1949). The study of the hydrolysis of esters, in particular with the substrate HPLA, has shown differences from the hydrolysis of corresponding peptide substrate analogues. These differences are:† (1) the values of  $k_{\text{cat}}$  for the esters and peptides are within a factor of five of one another, but the  $K_m$  of the esters are 20- to 110-fold less than those of the peptides (Snoke & Neurath 1949; Whitaker *et al.* 1966); (2) tyrosine modifications (Simpson & Vallee 1966; Riordan *et al.* 1967; Vallee 1967) which result in inactivation of peptide hydrolysis do not abolish esterase activity but rather diminish substrate inhibition (for example Whitaker *et al.* 1966); and (3) the pH-rate profile which has apparent points of inflection at pH 6.7 and 8.5 in peptidase activity is replaced in esterase activity by a pH-activity profile which shows an initial rise between pH 5.5 and 7.0, a plateau between 7.0 and 9.0, and a rapid second rise reaching a maximum at pH 10.5 (Riordan & Vallee 1963). The last two observations have led Vallee and co-workers to propose that peptides and esters are hydrolysed by different mechanisms (Vallee *et al.* 1963; Vallee 1964). Specifically, it was proposed that the nucleophilic group B which is required for peptide hydrolysis is not necessary for ester hydrolysis and that a hydroxide ion is the sole attacking group. In order to account for the differences given above for ester and peptide hydrolysis, it has been further proposed that esters have a locus of binding different from that of peptides (Vallee 1967). On the other hand, Bruice & Benkovic (1966, p. 5) have objected to the proposed role of OH<sup>-</sup> as nucleophile, on the grounds that the observed increase in rate of

† Although further studies are necessary, it also appears that the isotope effect  $k_{\text{H}_2\text{O}}/k_{\text{D}_2\text{O}}$  is in the range of 2.0 to 2.5 for HPLA (F. A. Quioco 1968, unpublished results; B. L. Kaiser & E. T. Kaiser 1968, private communication), substantially greater than the effect for certain peptide substrates.

esterase activity with a thousandfold increase in hydroxide ion concentration is only minimal or non-existent in the pH range of interest.

Unfortunately we have not as yet succeeded in preparing crystals of CPA complexes with ester substrates. However, we present the following observations, which are derived mostly from model building experiments. The only obvious difference between analogous peptides and esters is the steric and electronic configuration about the ester oxygen. The fact that peptide nitrogen can be replaced by oxygen in corresponding active substrates suggests that the essential point of substrate-enzyme interaction at the sensitive bond is the carbonyl group of the peptide or ester. The very nature of the active site especially with respect to its shape, the location of the zinc, and the probable binding and catalytic groups, combined with the characteristics of ester substrates which are most efficiently hydrolysed, severely limits the possible productive modes of binding. Since most of the requisites of substrate and enzyme, i.e. L configuration, C-terminal aromatic group, penultimate peptide (Snoke & Neurath 1949), and presence of a metal (Coleman & Vallee 1962) for peptide substrates are also found in the case of analogous ester substrates, we expect that the most favourable productive binding mode for ester substrates is in its essential interactions that depicted for peptides. However, the tyrosine modification experiments demonstrate that Tyr-248 need not necessarily participate catalytically in hydrolysis of some ester substrates, e.g. HPLA.

There is a class of short ester substrates, including *O*-acetyl-L-mandelate, whose pH-activity characteristics differ from those of HPLA. The pH dependences of  $K_{cat}$ ,  $k_{cat}/K_m$ , and the activity of CPA toward the hydrolysis of *O*-acetyl-L-mandelate are all qualitatively similar to the pH-activity curves for CBZ-Gly-L-Phe and suggest a common mechanism for both this ester and peptides (Carson & Kaiser 1966). This analogy is further substantiated by the finding that acetylated CPA is inactive toward the hydrolysis of *O*-acetyl-L-mandelate (F. A. Quioco 1968, unpublished results). It would seem essential for *O*-acetyl-L-mandelate and similar short ester substrates, such as *O*-acetyl-DL- $\beta$ -phenyllactate (Ogilvie, Riordan & Vallee 1963) which behave like peptides to bind like peptides.

Our comments on metal substitution are given previously (Lipscomb *et al.* 1968), and we therefore here refer only to the experimental result that when Hg replaces Zn the Hg atom is at a position shifted about 0.1 nm along the *a* axis (away from the pleated sheet) relative to the original Zn position. Detailed studies of the bonding geometries in the various metal substituted CPA (Vallee *et al.* 1963; Coleman & Vallee 1960) are required in order to provide additional physical data relating to the interesting chemical properties of these metal substituted forms of the enzyme.

## 5. SUMMARY

The positions of the atoms in the CPA crystal structure have been deduced, based on a tentative amino acid sequence drawn up by us from the known N-terminal sequence, our location on the structure of several chemically sequenced fragments, and X-ray identifications of the remaining 93 amino acids. The resulting atomic coordinates have been used in a structure factor calculation. Even though 5.3% of the protein atoms and all of the solvent molecules were omitted from the structure factor calculation, the standard crystallographic R factor is 0.44. The degree of perfection of the helices (about 35% of the structure) and of the  $\beta$  structure (about 20% of the structure) has been examined. The relation between primary and secondary structure has been considered.

A new function, with Fourier coefficients  $(|F_{ES}| - x|f_c| - (1-x)|F_N|) \exp(i\phi_{SF})$ , has been used to prepare a difference electron density map of the complex of Gly-Tyr with CPA. This map shows that (1) the C-terminal side chain of the substrate is inserted into a pocket of the enzyme; (2) the substrate's C-terminal carboxylate ion interacts with Arg-145; (3) the carbonyl group of the susceptible peptide bond is bound to the Zn atom; and (4) in an interaction restricted to dipeptides the terminal amino group of the substrate interacts with the carboxylate of Glu-270 via water. As a result of substrate binding Arg-145 moves 0.2 nm in order to bind the substrate's carboxylate ion. This motion of Arg-145 is coordinated with the 1.2 nm motion of the phenolic oxygen of Tyr-248 which brings Tyr-248 from the exterior of the molecule to a position where it can participate in catalysis. In addition, Glu-270 moves 0.2 nm away from the scissile bond toward the terminal amino group of the substrate. Arg-145 and Glu-270 have been identified only by inspection of the X-ray electron density map but Tyr-248 is contained in a tetrapeptide which was isolated by Roholt & Pressmann (1967) and located in the sequence by us.

Based on the information gained from the Gly-Tyr complex, the placement of substrates in the three-dimensional model of CPA and the wealth of chemical data in the literature we have made several deductions concerning the mechanism of action of CPA. In the active complex we believe that Arg-145 binds the terminal carboxylate ion of the substrate while the C-terminal side chain is inserted into the enzyme's pocket. The carbonyl group of the scissile peptide bond binds to Zn and is consequently polarized thereby rendering its carbon more susceptible to nucleophilic attack. Longer substrates have the carbonyl groups of the third and fourth peptide bonds directed toward Arg-71 and side chains of  $R_2$ ,  $R_3$  and  $R_4$ , especially if they are aromatic, located in the hydrophobic region of Tyr-198 and His(Phe)-279.

The displacement of most of the water molecules from the pocket of the enzyme and from the region near the zinc atom as well as the movement of Tyr-248 into the region of the substrate make the environment of the substrate hydrophobic in nature. Tyr-248, after its conformational change, is in an appropriate position to form two hydrogen bonds with the substrate. The first is donation to the N of the susceptible peptide bond, and the second is reception of H from the second peptide bond of the substrate. At some point in the reaction Tyr-248 may actually donate H to the substrate and be regenerated by  $H_2O$ . Glu-270 either acts as a nucleophile to attack the carbonyl carbon of the susceptible bond directly or induces the attack by water in a general base mechanism. Structurally, it is necessary that the conformational change of Tyr-248 be reversed before the product amino acid can leave the enzyme's pocket and another substrate molecule can be bound.

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*Note added in proof.* (18 August 1969): The sequence of CPA has now been determined chemically and reported in part (H. Neurath, R. A. Bradshaw & R. Arnon, International Union of Biology, Symposium on Structure–function relationships of proteolytic enzymes. Abstracts, Copenhagen, 1969). The X-ray identifications of the critical residues Arg-145, Tyr-248, and Glu-270 have been confirmed, and no changes are required in the mechanistic conclusions of this paper. The most important change in our identifications occurs at Zn ligand residue 196, which is found to be histidine. Also, in table 8 residue 166 should be deleted, and 240 and 265 should be added.

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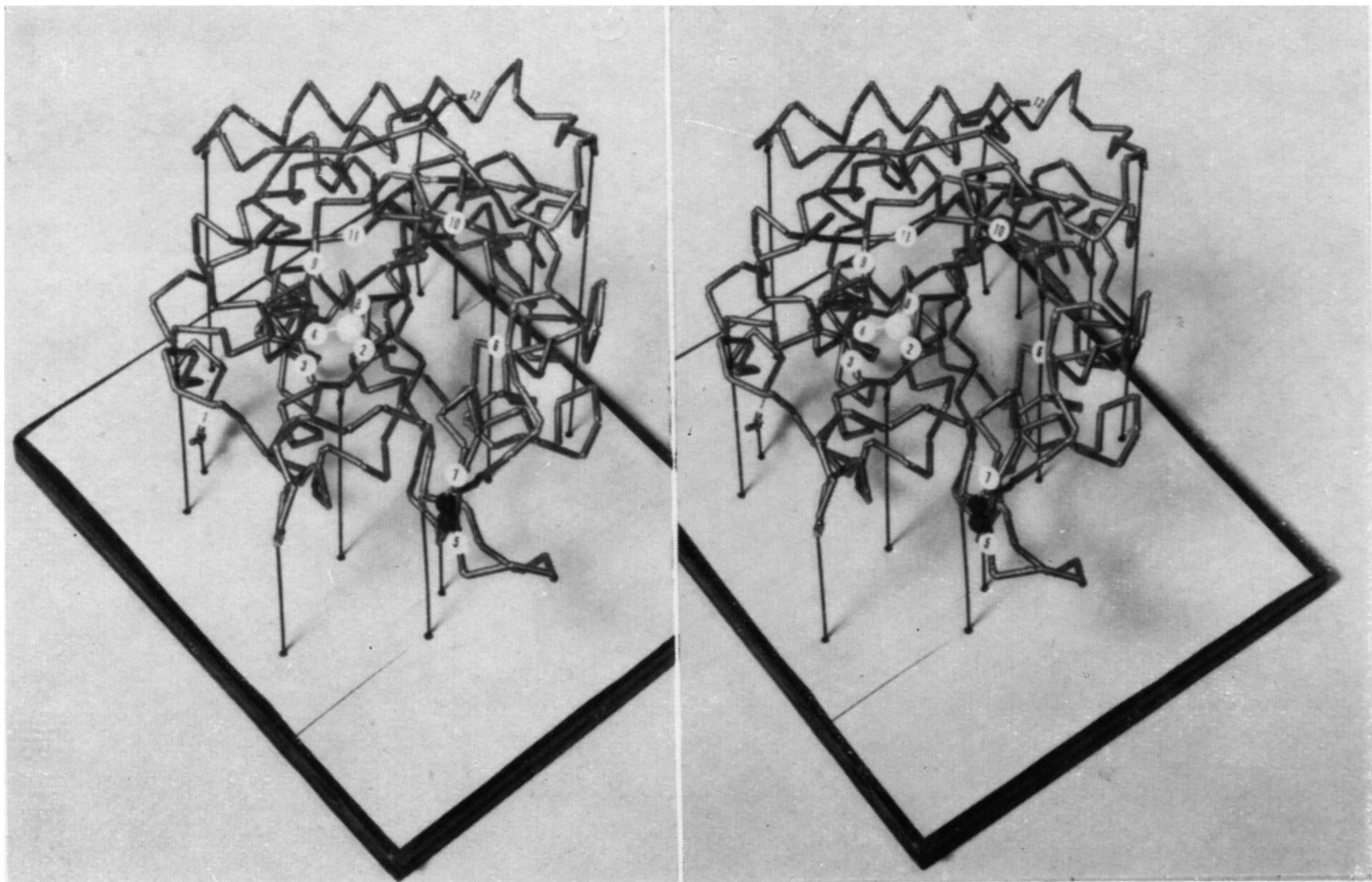
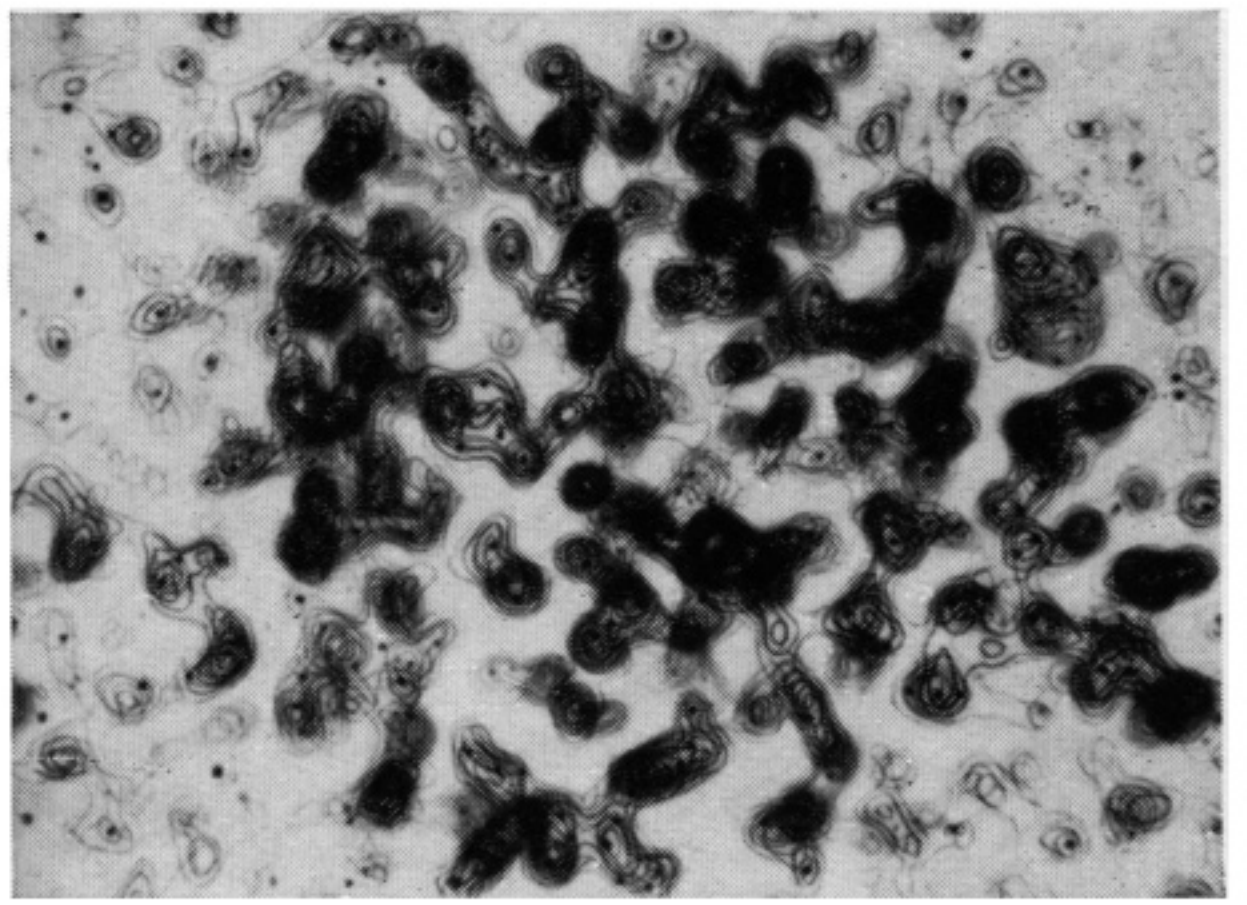
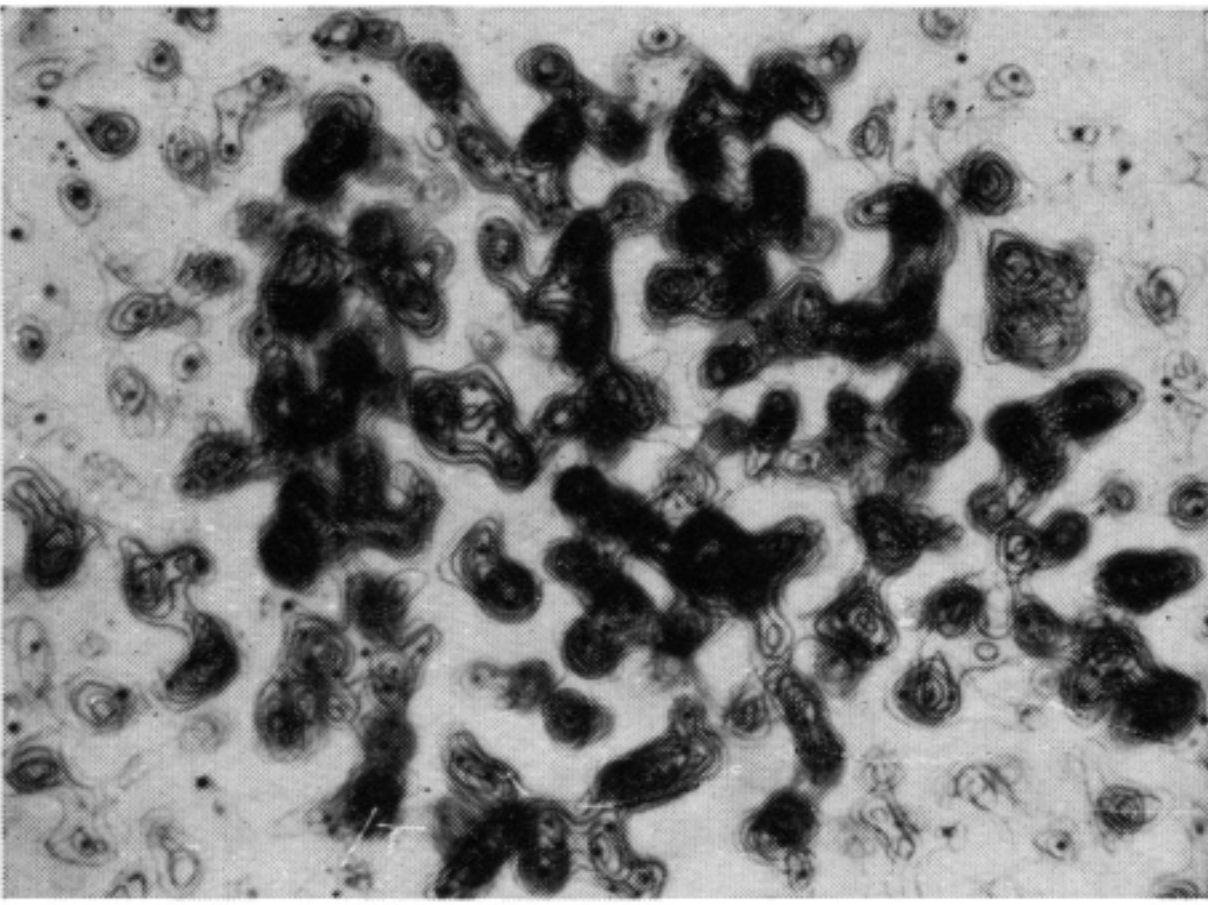
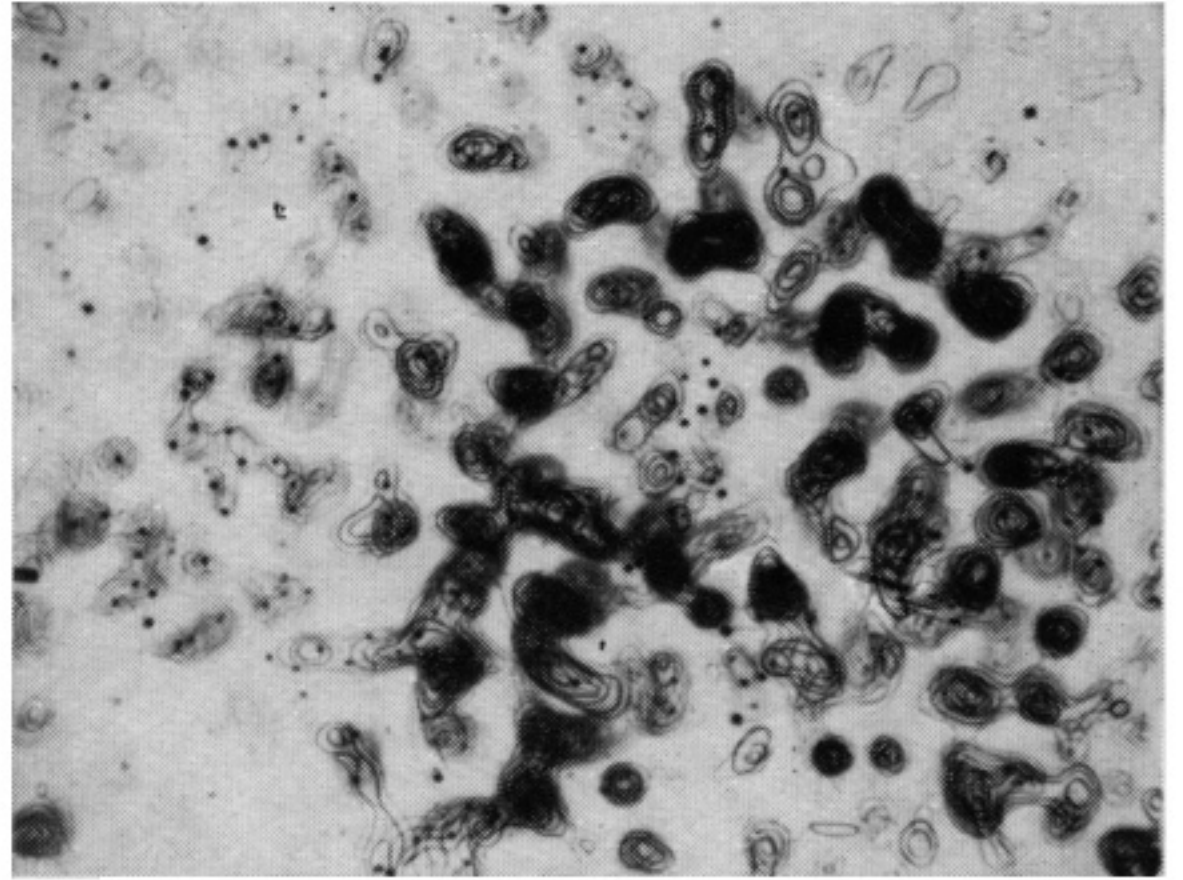
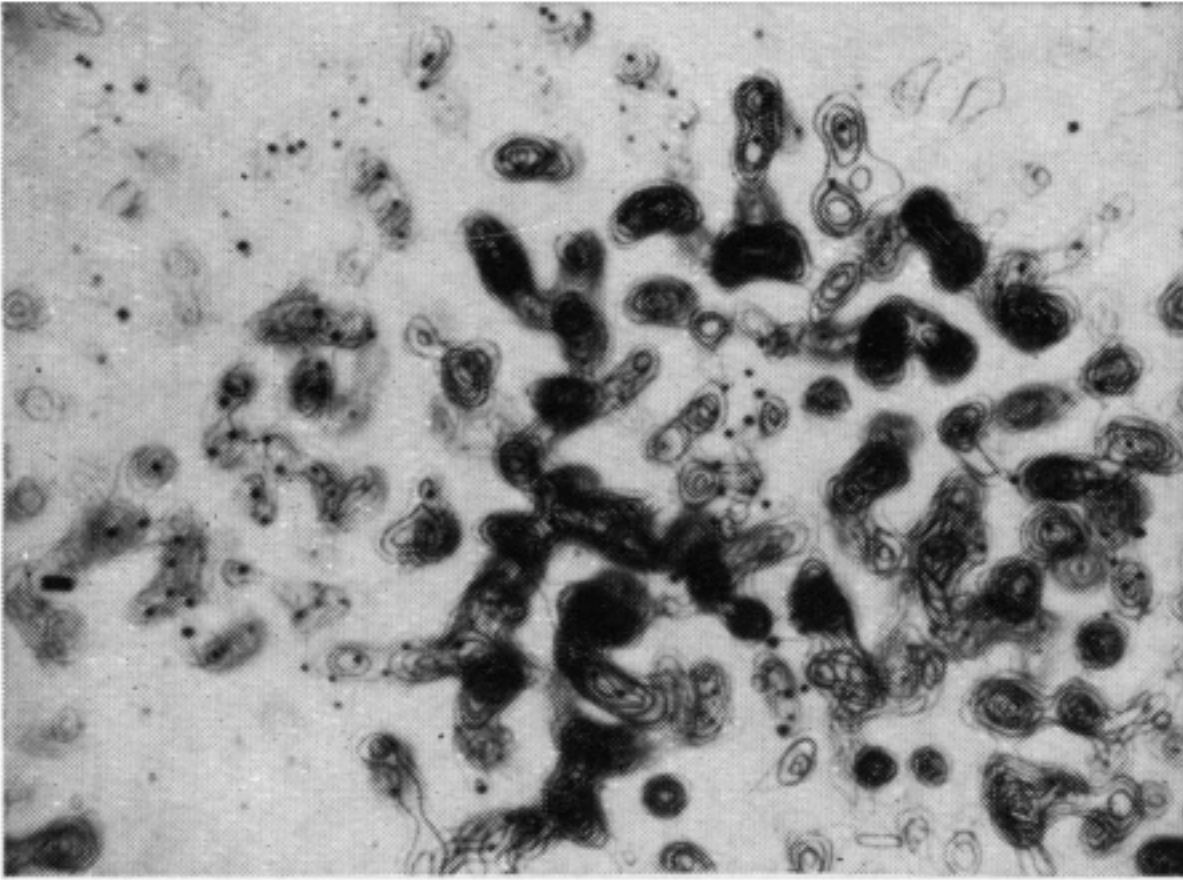


FIGURE 1. For legend see p. 189.

(a)



(b)



(c)

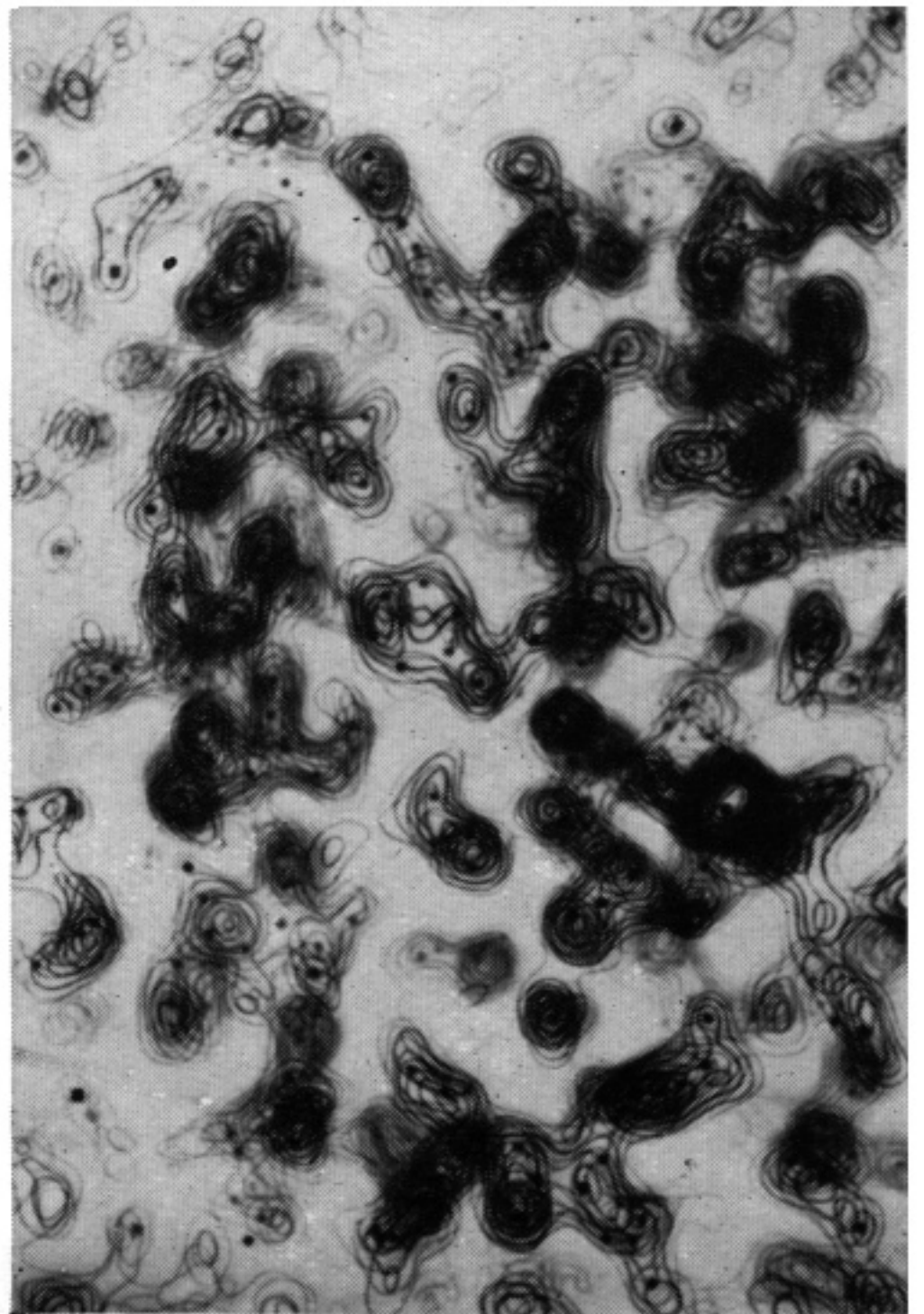
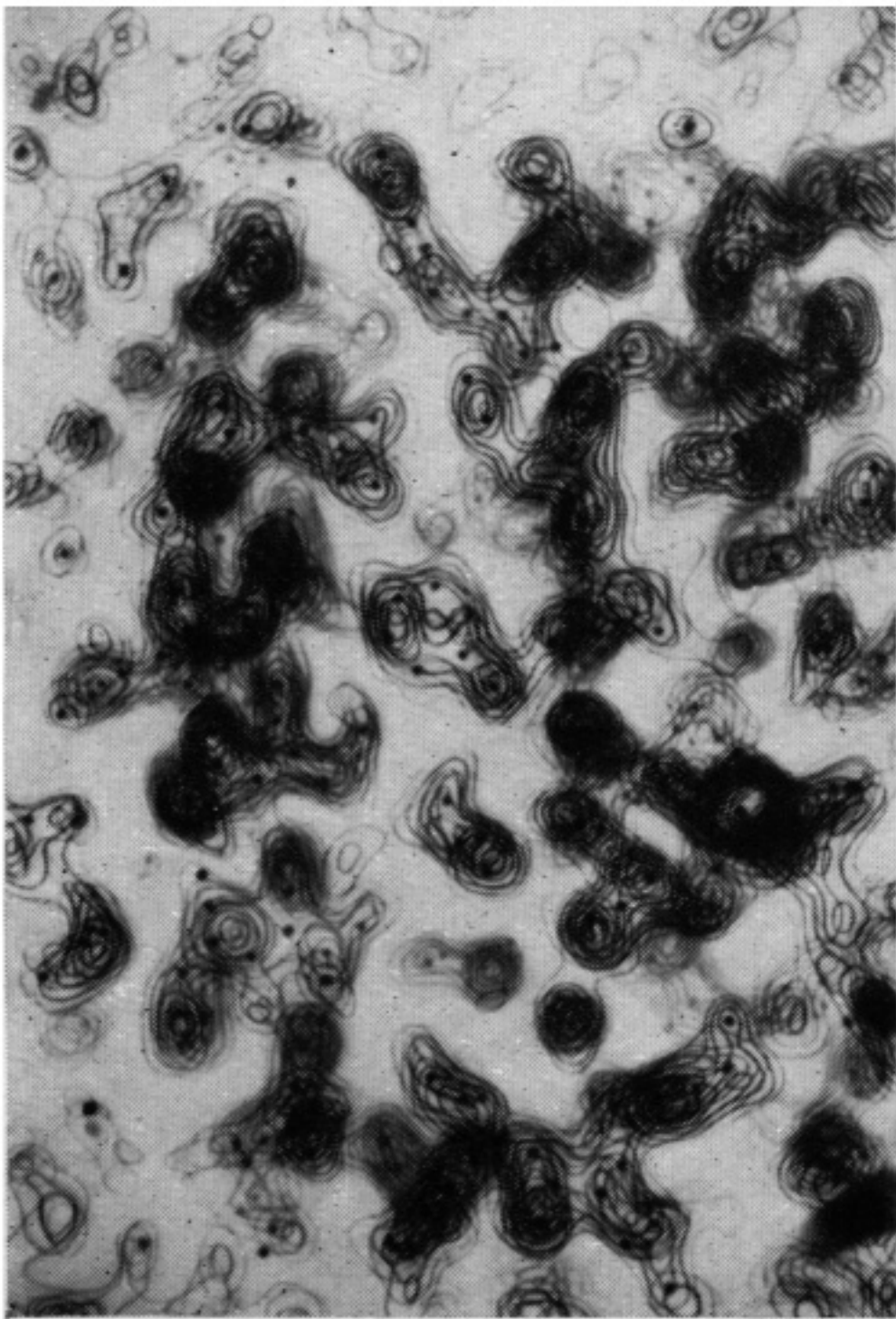
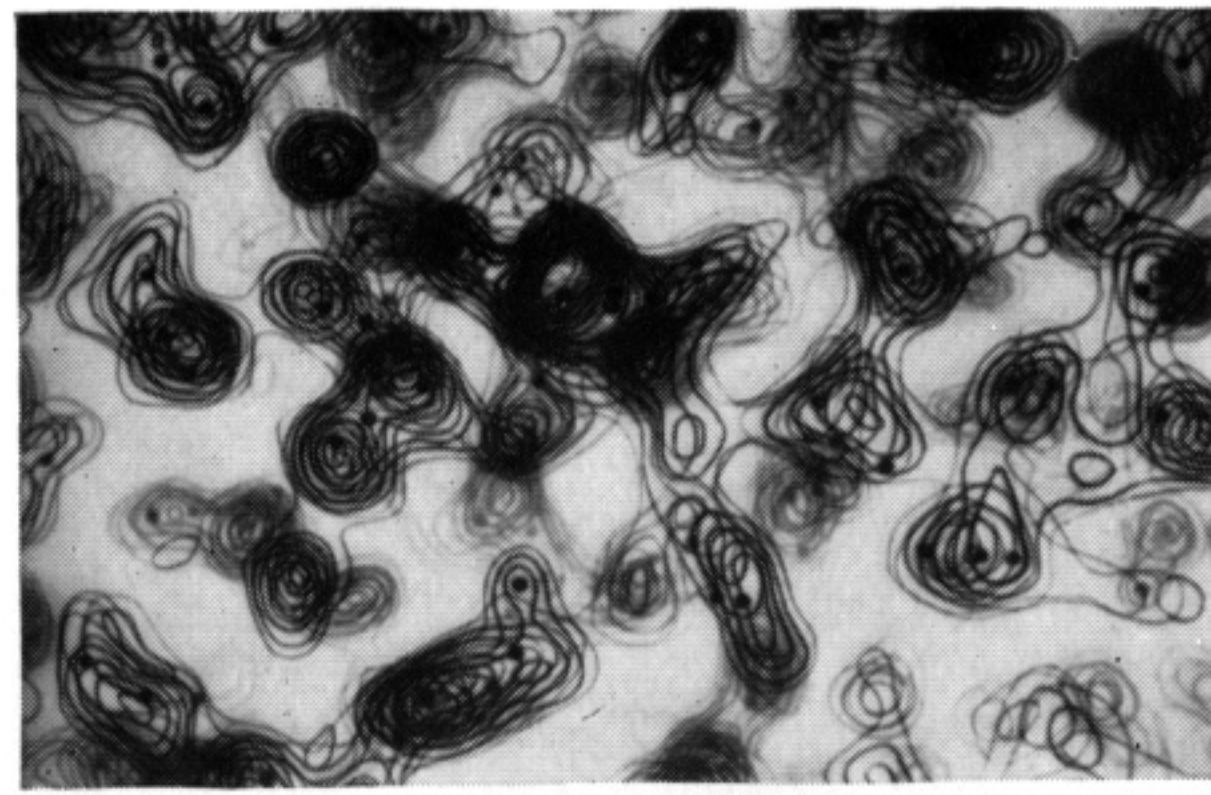
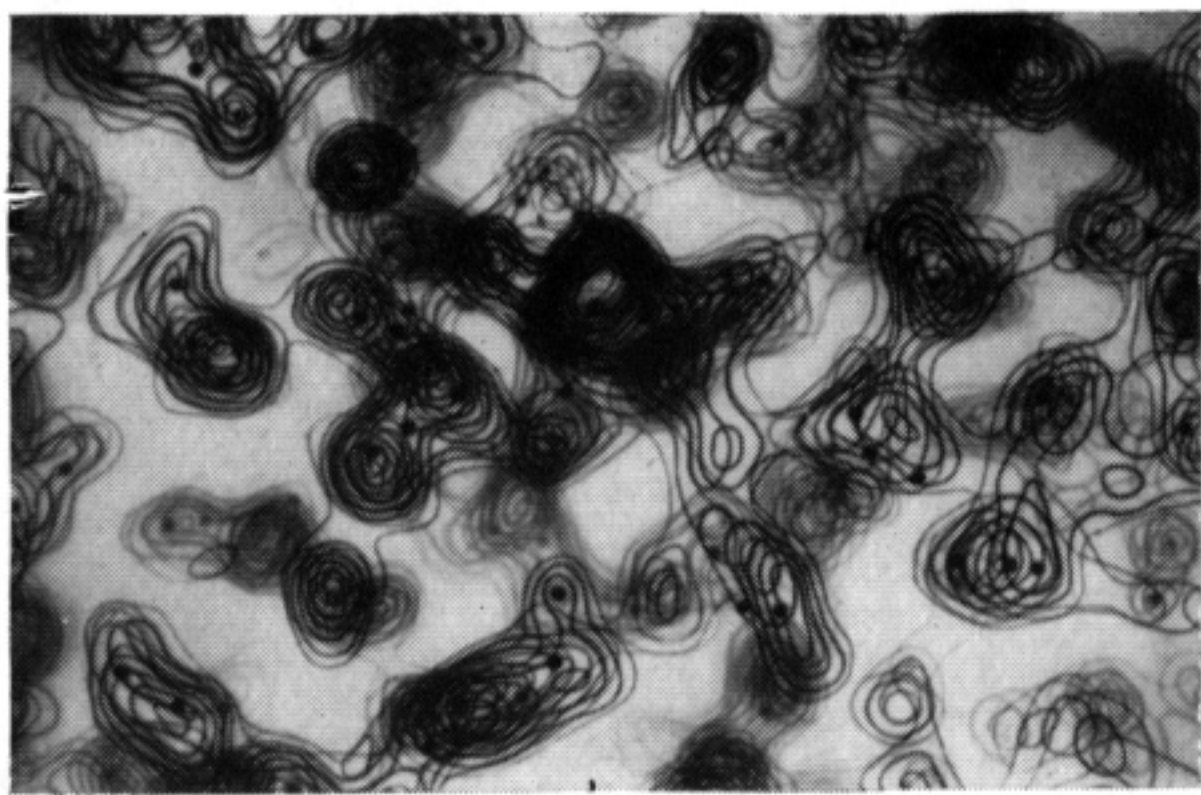
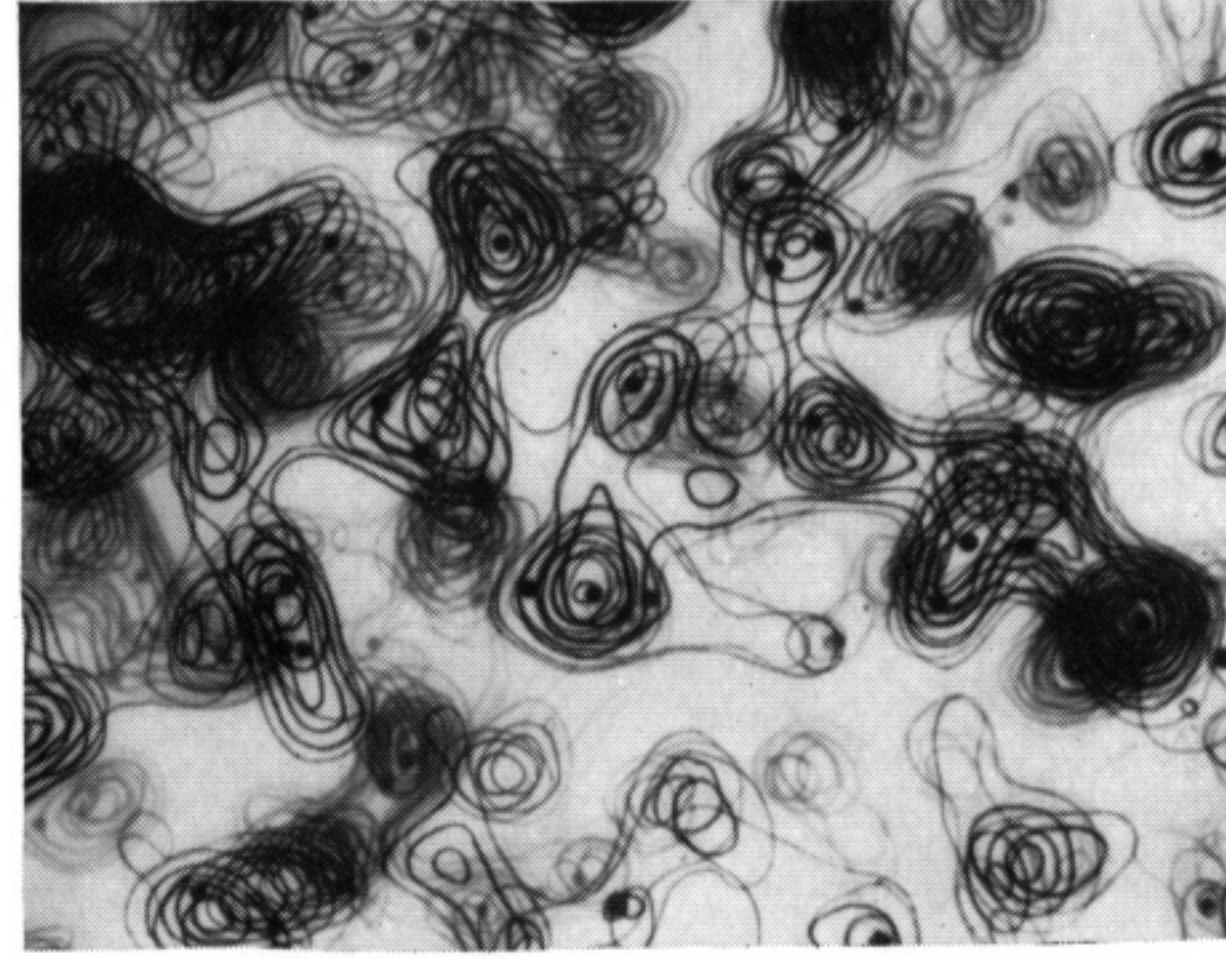
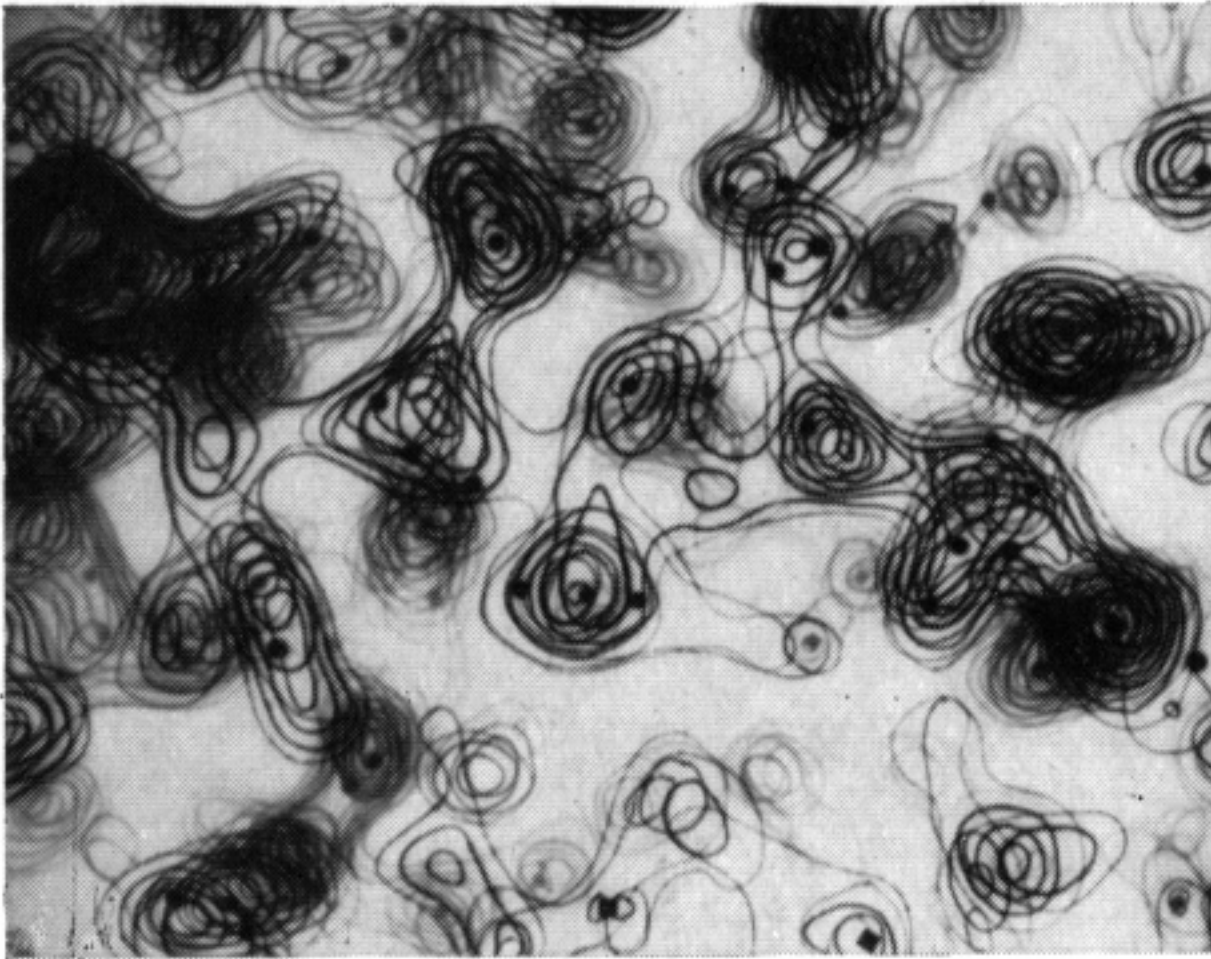


FIGURE 3 *a, b, c*. For legend see p. 189.

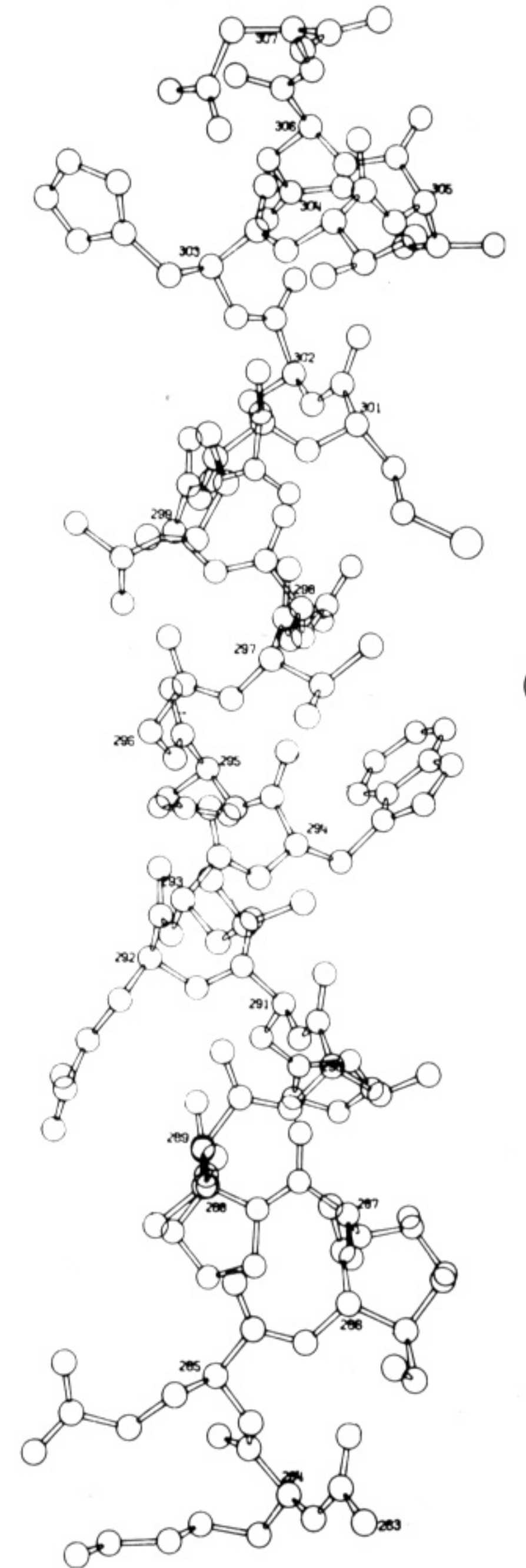
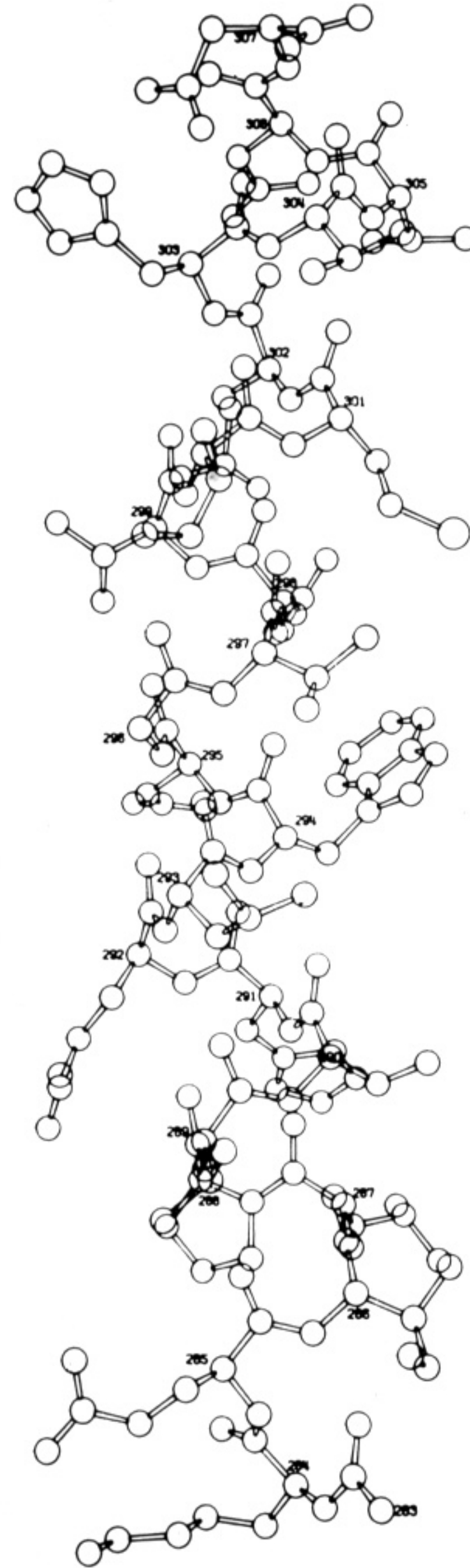
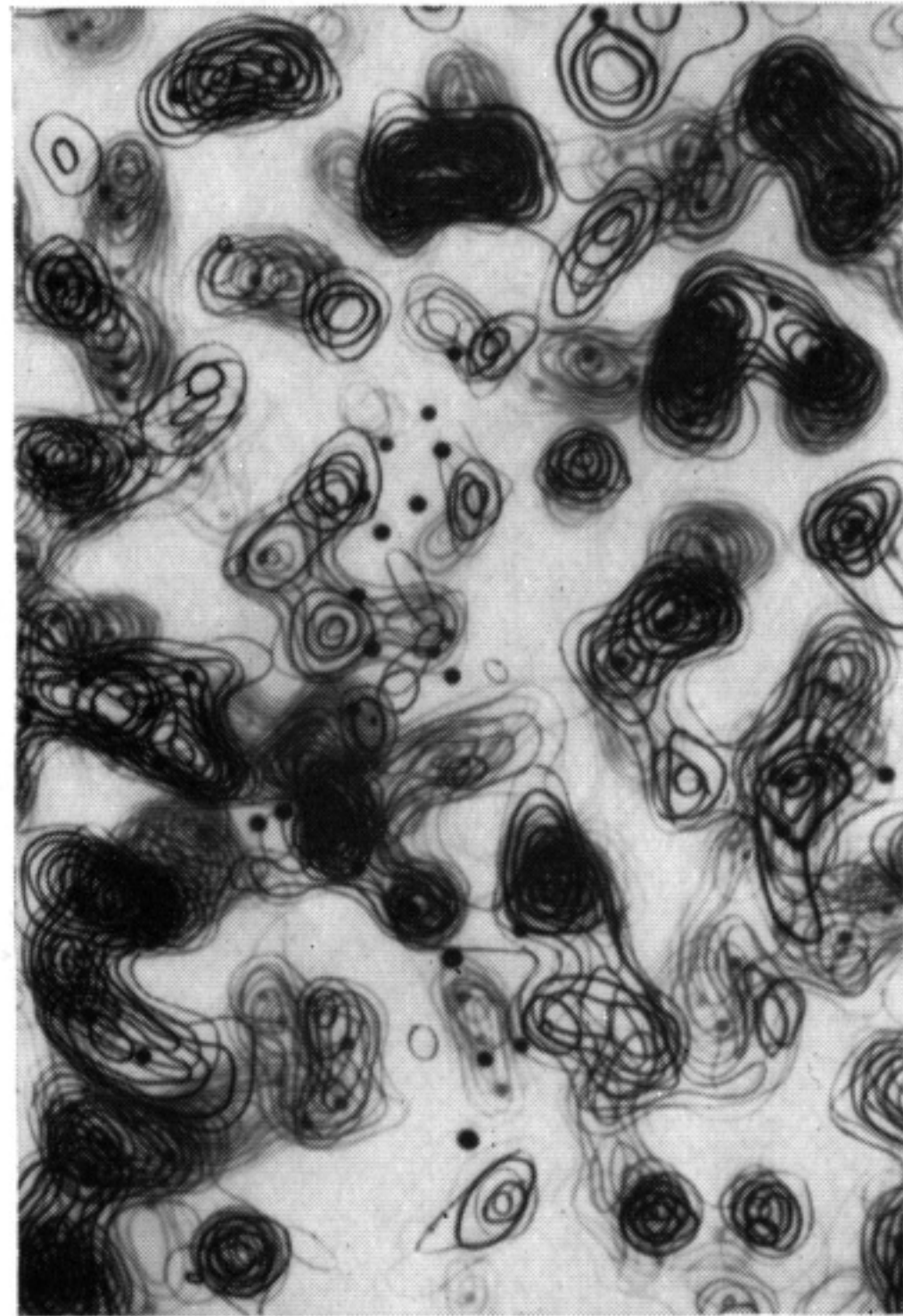
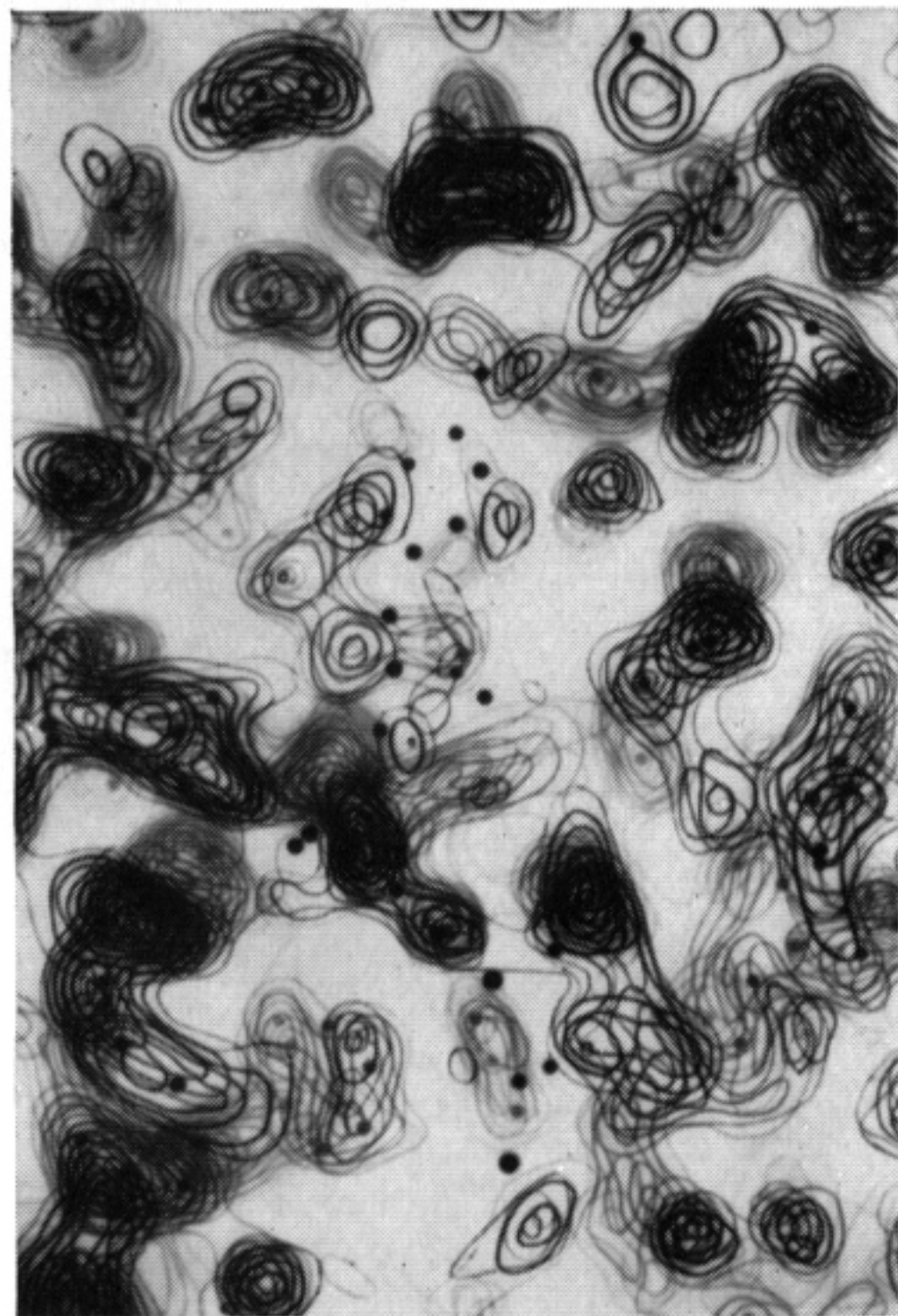
(d)



(e)



(f)



(g)

FIGURE 3 *d, e, f, g*. For legend see facing page.

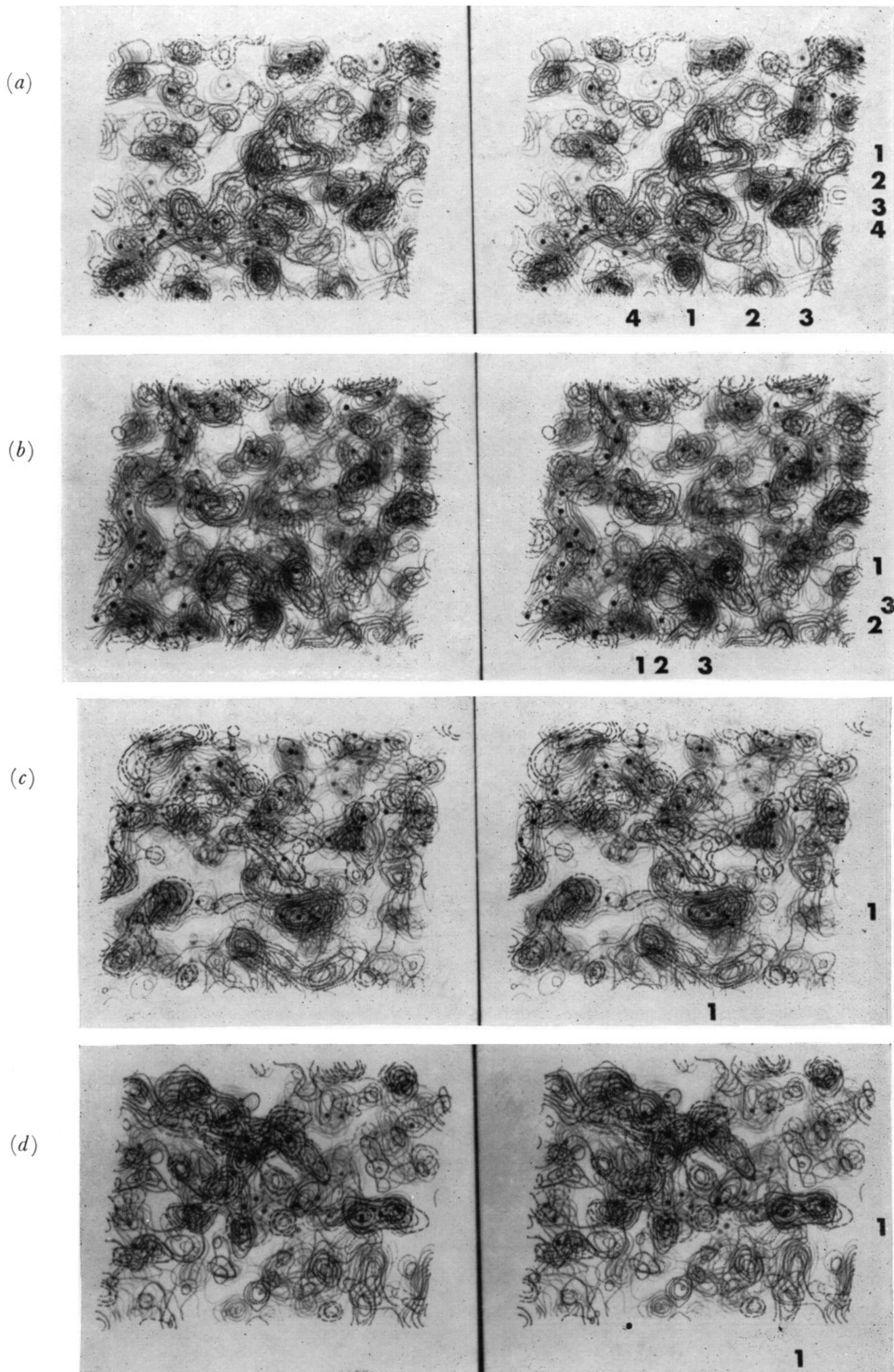
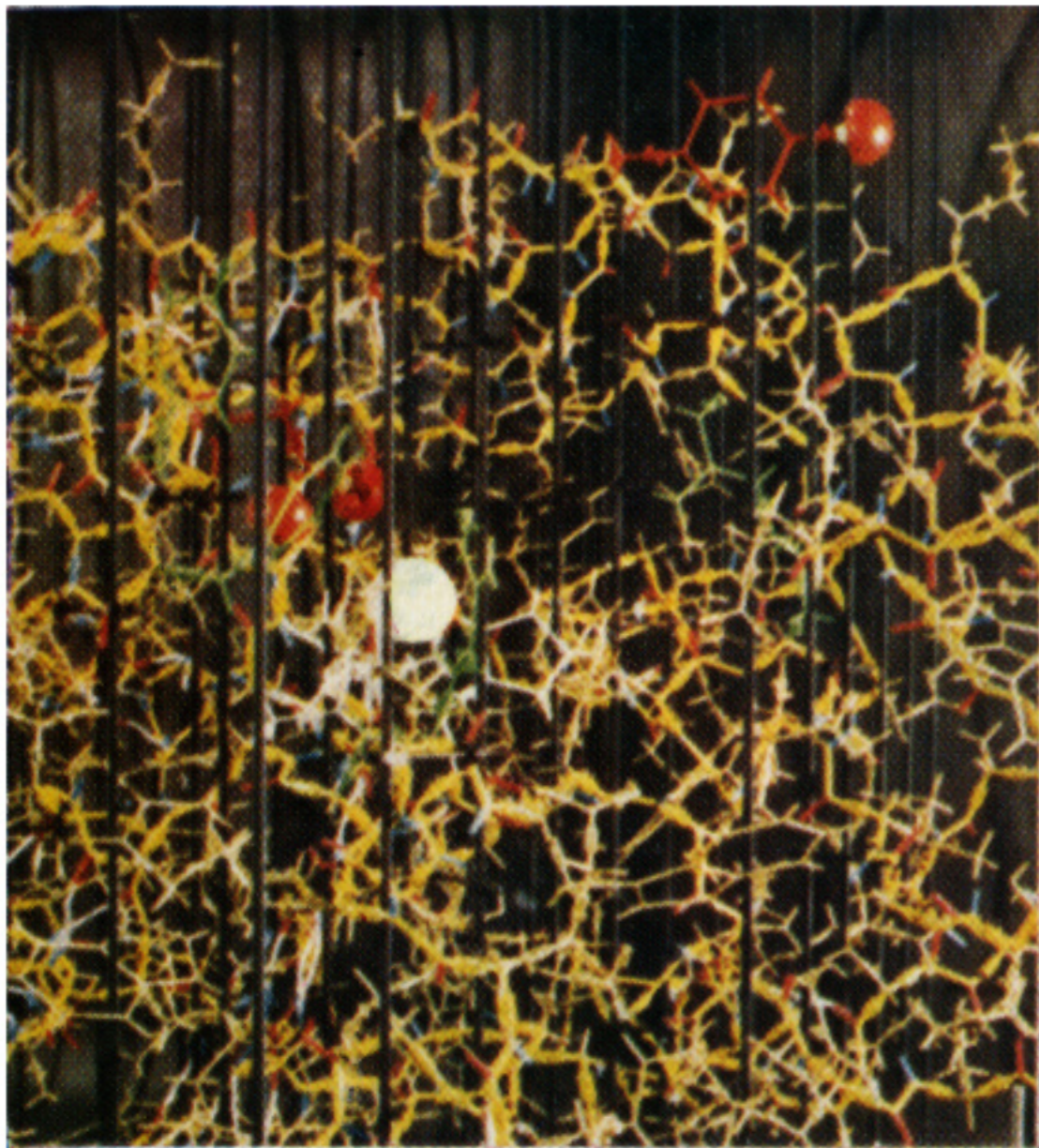
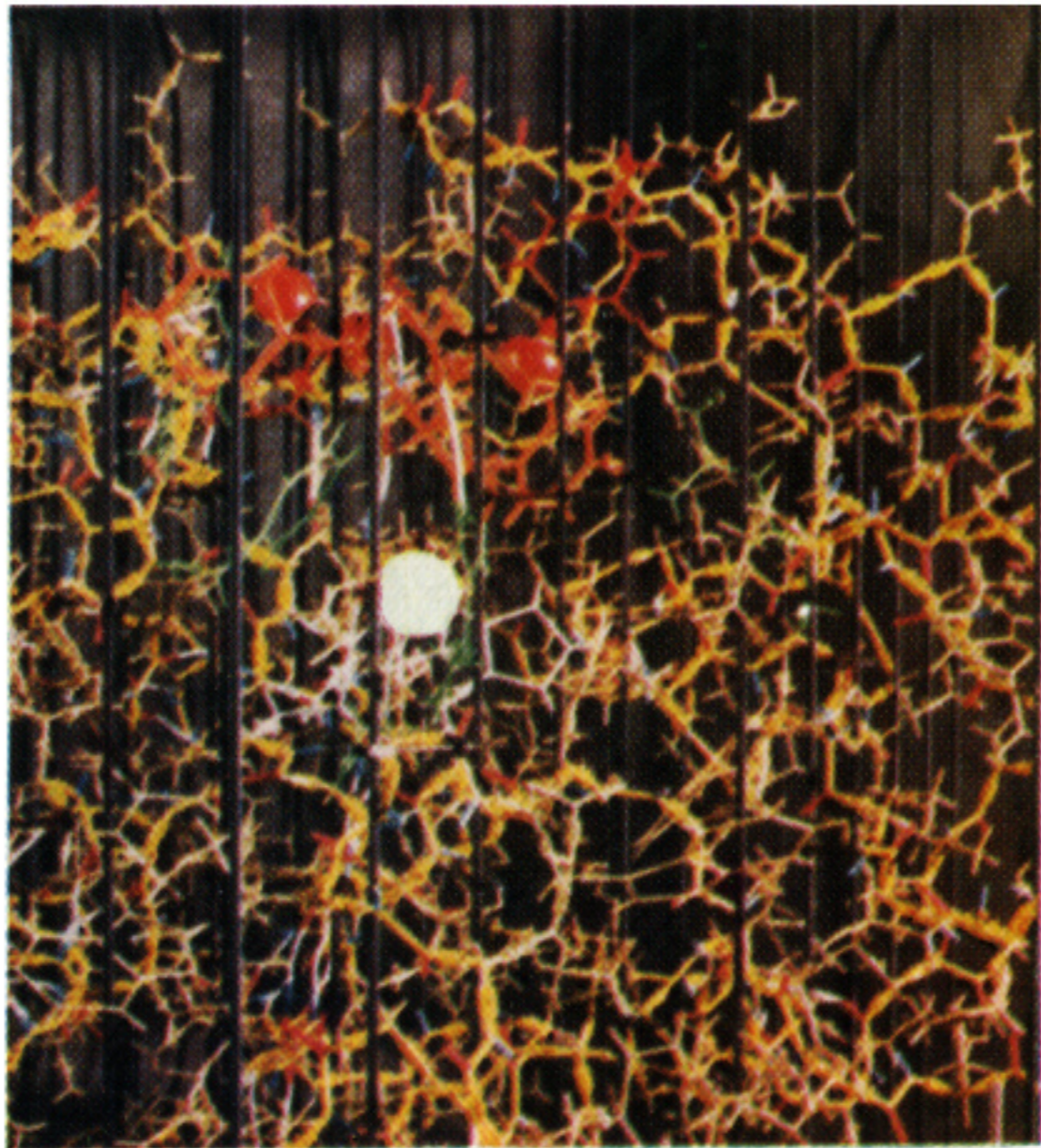


FIGURE 6. The difference electron density function for the complex of Gly-Tyr with CPA is shown as dotted (positive) and dashed (negative) contours. Solid contours show the electron density of CPA at 0.2 nm resolution. Dots are placed at proposed atomic positions. (a) Composite of the difference map sections  $y = 0.47$  to  $y = 0.52$ . Near the top centre the positive contours of the tyrosyl side chain of the substrate are visible (1). To the right of the substrate are the positive (2) and negative (3) contours of the Arg-145 guanidinium group and to the left (4) are the native contours of Glu-270. (b) Composite of the difference map sections  $y = 0.49$  to  $y = 0.56$ . Near the bottom of the picture are the positive contours of the moved Glu-270 (1), the substrate's terminal amino group (3), and the connecting water molecule (square dot, 2). (c) Composite of the difference map sections  $y = 0.56$  to  $y = 0.62$  shows Tyr-248 after its conformational change in the bottom right portion of the picture (1). (d) Composite of the difference map sections  $y = 0.63$  to  $y = 0.68$  shows Tyr-248 before its conformational change (1).

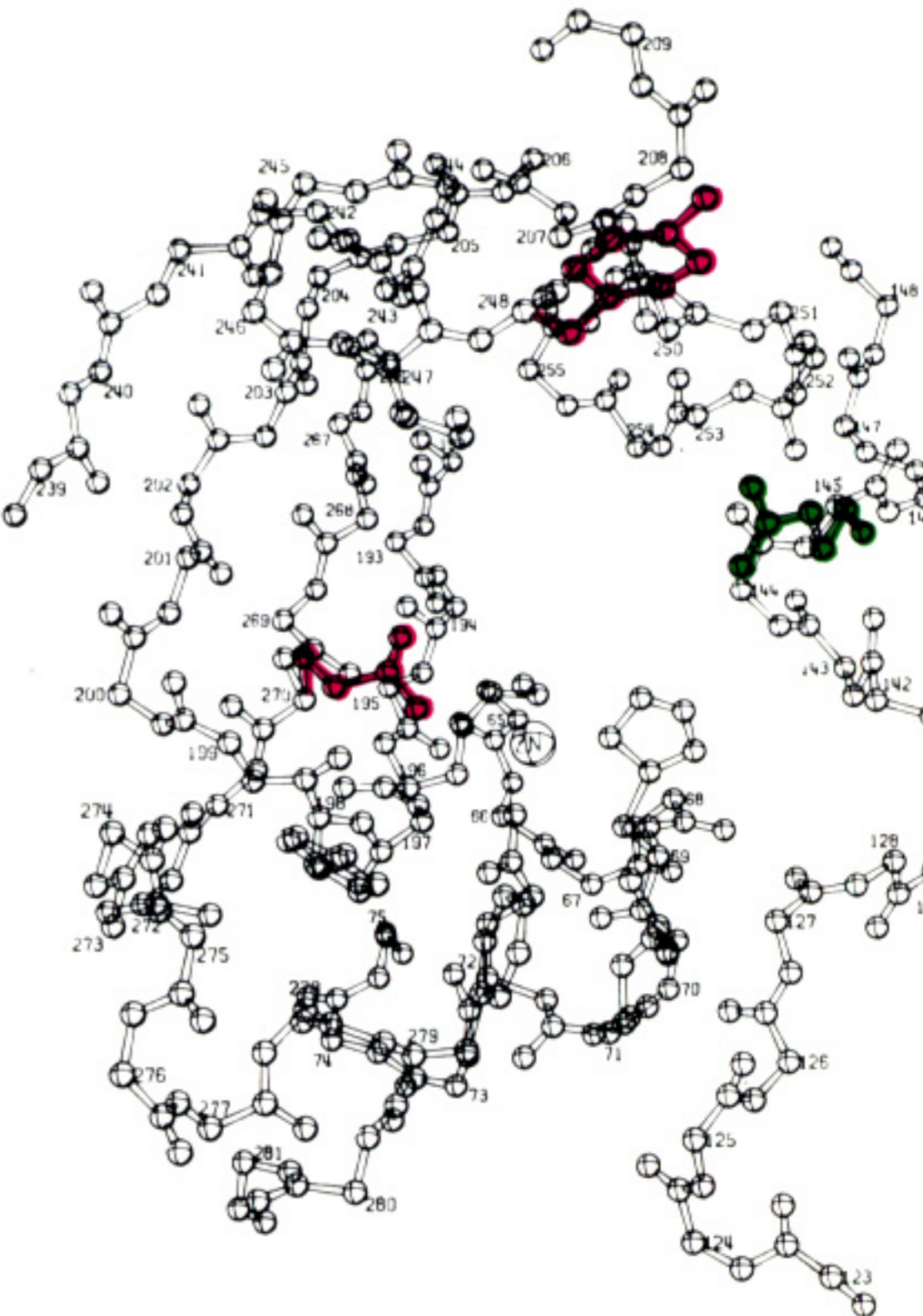
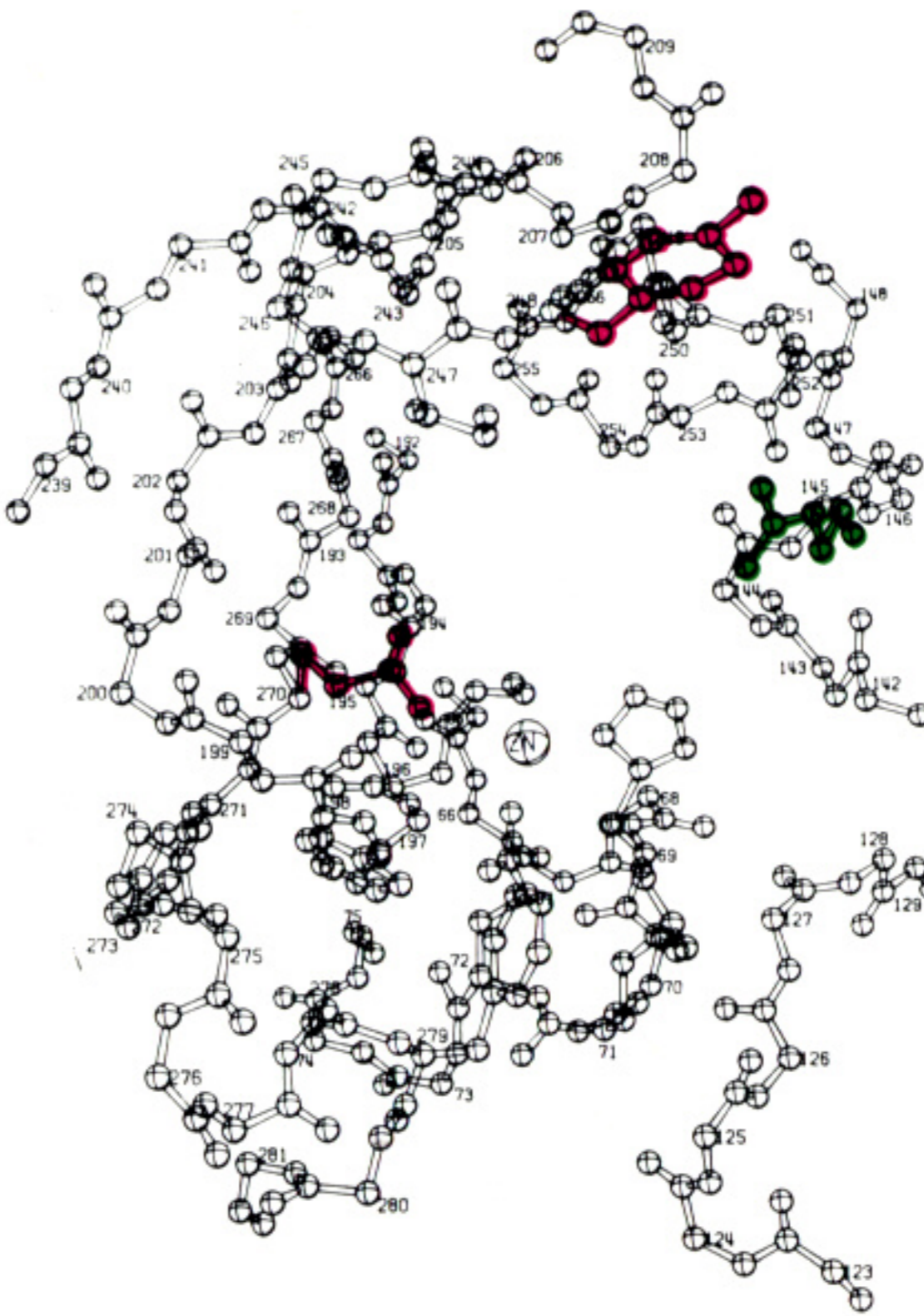
(a)



(b)



(e)



(f)

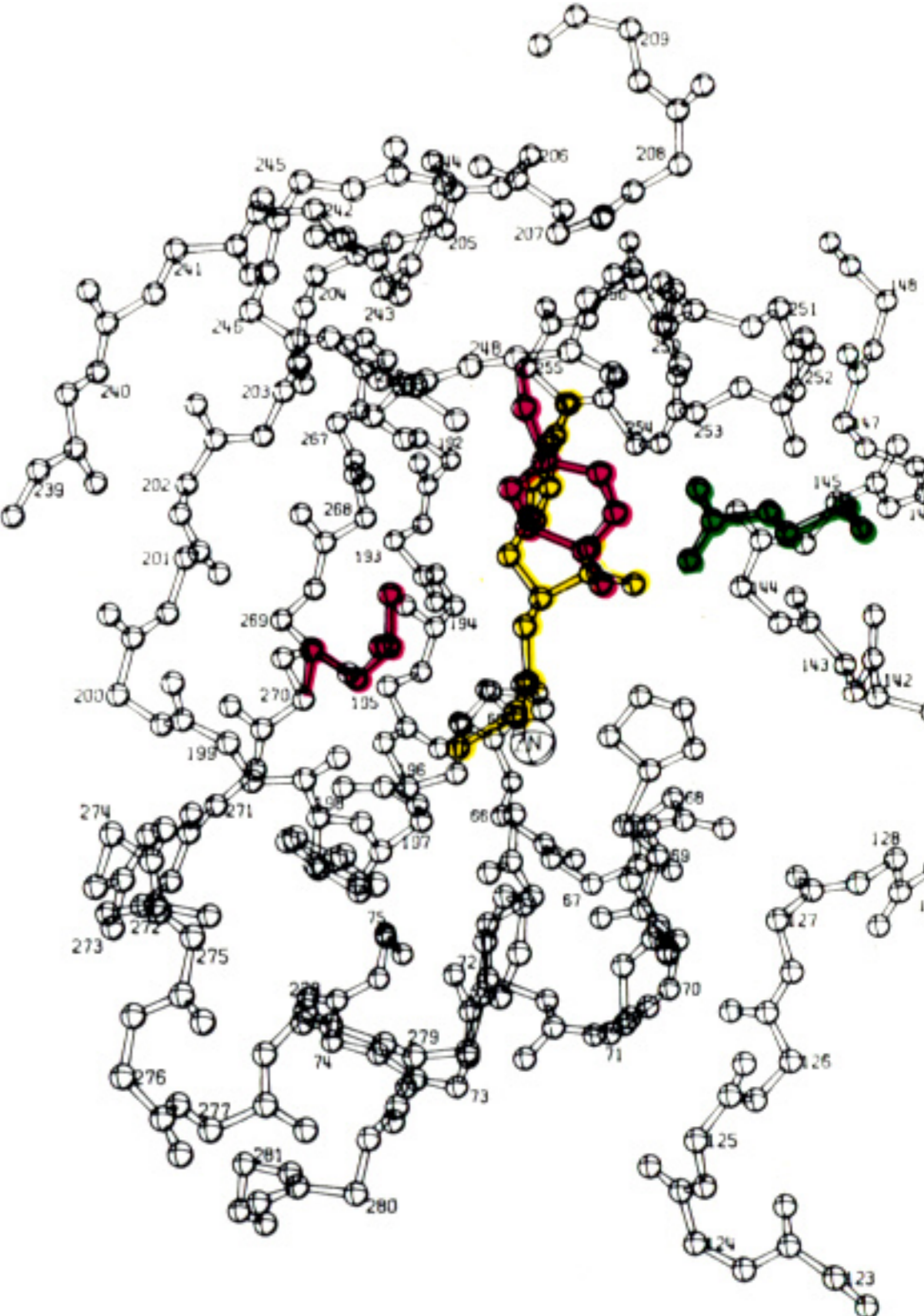
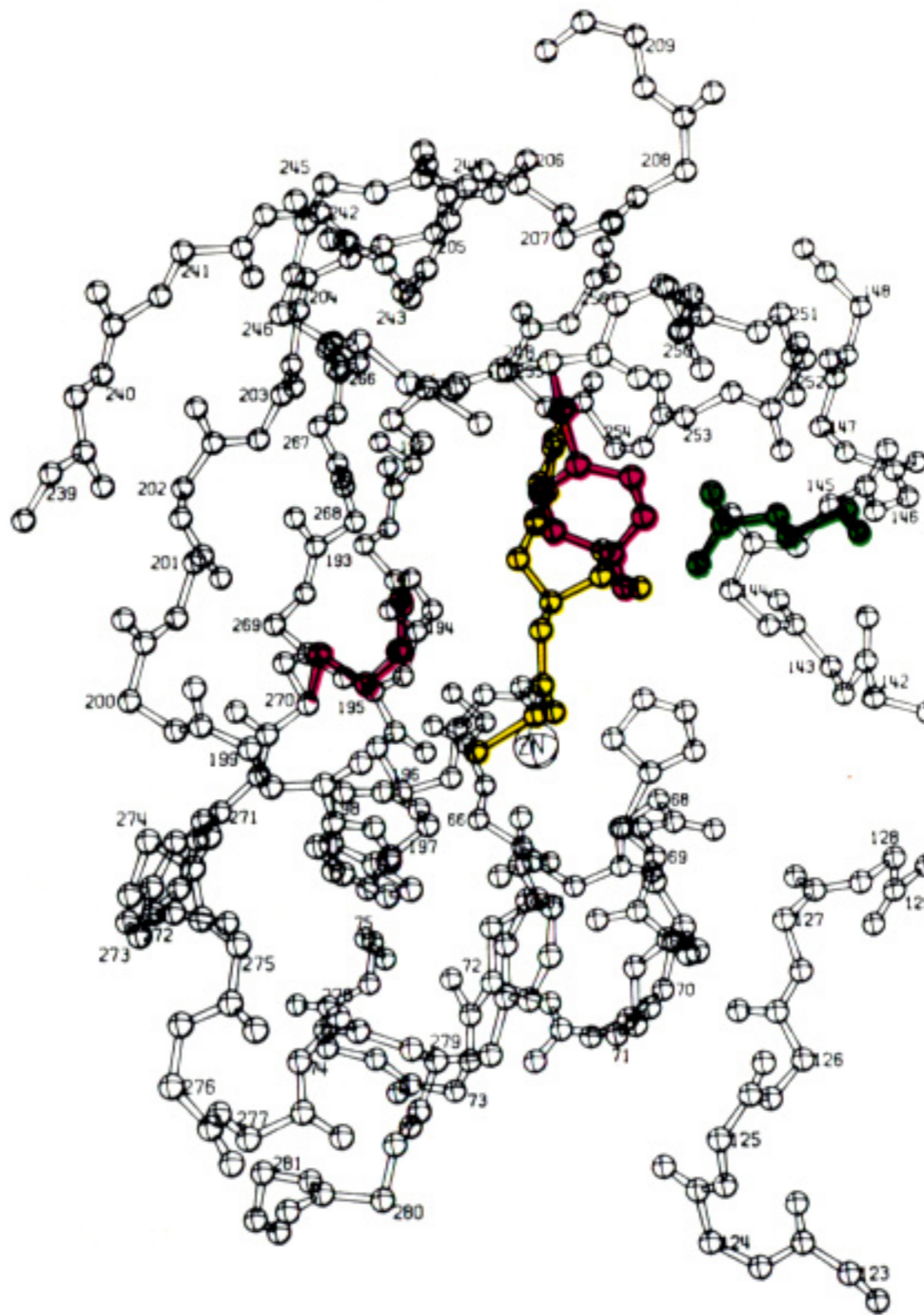


FIGURE 9 *a, b, e, f*. For legend see facing page.